

Bioaccumulation of Cadmium in Freshwater Fish: An Environmental Perspective

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ABSTRACT

Background: Cadmium (Cd) is considered as one of the most toxic heavy metal. Intake of Cd by fish has serious implications. Metal pollution from multifarious sources has adverse effects on aquatic ecosystems. In aquatic systems, Cd is most readily absorbed by organisms directly from the water in its free ionic form. In many contaminated situations with heavy metals, Cd has become an important element of concern because of its bioaccumulative nature in food webs. Therefore, fish living in polluted waters tend to accumulate heavy metals in their tissues. **Objectives:** To assess the influence of Cd on freshwater fish, this review briefly addresses the Cd emission sources, uptake and impacts of Cd on freshwater fish and bioaccumulation nature of Cd by emphasizing the Cd accumulation affinity in freshwater fish tissues. **Results and Discussion:** Many studies have been carried out on Cd thresholds in diet and tissues of freshwater fish species. Affinity of freshwater fish to Cd is diverse. Generally, Cd accumulation depends on concentration, time of exposure, way of uptake, environmental conditions and intrinsic factors. Metal accumulation primarily depends on waterborne and dietary pathways. It shows a relationship of Cd level in fish tissues with the age and size of fish. Some species of fish show the highest Cd accumulation in the liver while others in kidneys and gills. Accumulation of Cd by the body muscles is always reported as comparatively low. Cd in freshwater environments results biological and environmental implications by altering reproductive and physiological behaviors of freshwater fish and abilities which ultimately affect environmental permanence and biodiversity of the ecosystem.

Key words: Environmental permanence, physiological implications, fish tissues, dietary pathways, metal pollution

Insight Ecology 4 (1): 1-12, 2015

INTRODUCTION

The term bioaccumulation is defined as a process by which the chemicals are taken up by an organism either directly from exposure to a contaminated medium or by consumption of food containing the chemical¹. Where metals are concerned, it can be defined as the net accumulation of a metal in a tissue of interest or a whole organism that results from exposure². Metal bioaccumulation is influenced by multiple routes of exposure (diet and solution) and geochemical effects on bioavailability³. As metals are not metabolized, bioaccumulation of metals and metalloids is of particular value as an exposure indicator. Similarly, bioaccumulation is often a good integrative indicator of the chemical exposures of organisms in polluted

ecosystems⁴. All trace metals are toxic at some bioavailability⁵. Thus, aquatic organisms exposed to atypically high local bioavailable toxic metal may come under selection for changes in one or more physiological processes, including the rate of metal uptake from an available source of the metal, the rate of efflux and the rate of detoxification of accumulated metal into a relatively metabolically inert form⁶.

The metal contamination in aquatic ecosystem is considered to be unsafe not only for aquatic organisms but also for terrestrial organisms including the human. The long-term consumption of fish from the polluted waters may result in bioaccumulation of persistent pollutants in ultimate recipient (perhaps human) of the food web. In fish, heavy metals are taken up through different organs because of the affinity of such organs for the accumulation of heavy metals⁷. In this process, many heavy metals are concentrated at different levels in

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different organs of the body⁸⁻¹⁰. As an example, cadmium (Cd) is considered as one of the most toxic heavy metals¹¹ and an environmental pollutant toxic to a number of tissues¹². The persistence and ubiquitous nature of Cd is coupled with their tendency to accumulate in organisms ultimately produce toxic reactions in aquatic biota, especially in fish. Thus, the deleterious effects of metals on aquatic ecosystems necessitate the continuous monitoring of their accumulation in key species, since it affords indication of temporal and spatial extent of the process and impact on organism's health¹³. As the bioaccumulation of Cd in aquatic biota is a serious current issue, it should be closely monitored. The intent of the present review is to briefly address the Cd emission sources, uptake and impacts of Cd on freshwater fish and bioaccumulation. Mainly, try to emphasis the Cd accumulation affinity in freshwater fish tissues; e.g., gills, kidney, liver and the muscle tissue. This study also gives an overview of differences in the magnitude of Cd residues accumulated in different freshwater fin-fish tissues in several natural environments of the world. Nevertheless, metal bioaccumulation in terms of biodynamic modelling and physiological handling of metals like Cd in freshwater fish, through the studies of genomics and proteomics studies will not be discussed.

CADMIUM IN THE ENVIRONMENT

Cadmium (Atomic Number: 48 and Relative Atomic Weight: 112.41U) is a relatively rare, silvery grey metallic, soft solid (standard state). It never occurs in nature in its elemental form and is always found in a compound with another element; i.e., cadmium oxide, cadmium chloride, cadmium sulphide, cadmium cyanide, cadmium carbonate and cadmium nitrate. Although rare in surface waters, Cd is highly toxic to some aquatic life. In comparative acute toxicity testing of all 63 atomically stable heavy metals in the periodic table, Cd clearly was the most toxic metal¹⁴. The important releases of Cd to the biosphere can be discussed as natural and anthropogenic activities. Natural emissions are mainly due to mobilization of naturally occurring Cd from the earth's crust and mantle; e.g., volcanic activity and weathering of rocks. Anthropogenic releases are mainly from the mobilization of Cd impurities in raw materials (e.g., phosphate minerals, fossil fuels) and releases by manufacturing, use, disposal, recycling, reclamation or incineration of products intentionally¹.

Chemical elements in soil are referred to as trace elements (heavy metals) because of their occurrence at concentrations less than 100 mg kg⁻¹¹⁵. The term

"Heavy metals" is the most popularly used and widely recognized term for a large groups of elements with density greater than 6 g cm⁻³. Agro-fertilizers are indispensable for ensuring sustainability of agricultural production¹⁶. Nitrogen, sulphur and potassium fertilizers are relatively free of impurities but phosphorus fertilizers contain several contaminants, of which Fluorine (F) and Cd are considered to be of most concerned. Application of phosphate fertilizers, lime and agrochemicals (pesticides) use in agriculture can be significantly contribute to enhance potentially hazardous trace elements in soils¹⁵. Therefore, crop fields that are heavily contaminated with Cd may have resulted from long-term overuse of phosphate fertilizers¹⁷. Once soils with elevated Cd concentrations are exposed, Cd will leach from the solids and dissolve in water. The soil is the primary source of trace elements for plants, animals and humans¹⁵. Pavlik¹⁸ has shown that, up to 90% Cd taken up by plants originates from soil and only 10% from the atmosphere. Trace elements such as Cd are retained in soils indefinitely because they are not degradable. Therefore, an increased level of Cd in water and soil increases its uptake by live organisms.

Cadmium has no known biological use in animals¹⁹⁻²⁰ or only little evidence to suggest that it plays a nutritive role in higher plants and animals¹⁵. Cadmium is poorly regulated by organisms, thereby increasing the likelihood that whole-body residues will increase with increasing exposure concentration²¹.

IMPACTS ON THE AQUATIC ECOSYSTEM

When Cd is introduced to freshwaters, the great bulk of the metal precipitates and resides in the bottom of sediments. Thus, sediment may be a significant source for Cd emitted to the aquatic environment²². The effects of Cd on aquatic organisms can be directly or indirectly lethal and can impact populations and ecosystems as well as individuals. As a persistent environmental pollutant, Cd can alter trophic levels for centuries and freshwater organisms such as fish are particularly vulnerable to Cd exposure²³. Partitioning of Cd between the adsorbed-in-sediment state and dissolved-in-water state is therefore, an important factor in whether Cd emitted to waters is available or not to enter the food chains. Subsequently, Cd in sediments will be taken up by bottom feeding fauna and sediment-rooted flora. It then proceeds up the food chain to fish, with the precise pathways dependent upon the species present²⁴. Metal concentrations in benthic macro-invertebrate tissues and fish tissues were strongly correlated, suggesting a transfer of metals through a dietary pathway. Therefore, fish living in polluted waters

tend to accumulate heavy metals in their tissues²⁵. Even if, most of the data on the effects of Cd to aquatic species were generated from tests where Cd was tested as the single contaminant of concern, it probably seldom occurs as a single contaminant of concern in ambient waters (e.g., commonly occurs with zinc and copper)²⁶.

UPTAKE OF CADMIUM BY FISH

In general, considerations of metal bioavailability and bioaccumulation in aquatic media can be direct and indirect exposure. Fish takes in Cd directly from exposure to a contaminated medium or by consumption of food containing the chemical²⁷. Direct uptake of Cd by fish from the water is mainly in its free ionic form (Cd^{2+})¹¹ and the indirect exposure is possible as dietary means when consumer organisms subsequently ingest metals bioaccumulated in organisms at a lower trophic level with the potential for effects or bioaccumulation². Despite the fact that direct and indirect exposure of bioavailability and bioaccumulation are considered separately, both of these processes act integrated with each other in the natural systems. Metal uptake in fish differs fundamentally from that of terrestrial animals because fish have gills that are constantly submerged in a solution of metal ions. Therefore, in freshwater fish, gills are the main point of entry for dissolved metals. The gastrointestinal tract also acts as an important route for metal absorption in fish. Li and the colleagues²⁸ reported that a small fraction of Cd present in the dissolved form in water might be ingested directly by fish through the skin as well. Metal ions are usually absorbed through passive diffusion or carrier mediated transport over the gills while metals associated with organic materials are ingested and absorbed by endocytosis through intestine. It has been suggested that Cd ions enter the chloride cells in the gills through calcium channels²⁹. Cadmium that absorbs across the gills or the intestinal walls is distributed via the circulation, bond to transport proteins and distributed, to different tissues of the body²⁹. Once Cd enters into the cells, the metal is made available for the interaction with cytoplasmic components such as enzymes causing toxic effects and Metallothioneins (MTs).

Metallothioneins (MTs) are a low-molecular-weight metal binding proteins and are known to play several important roles, especially in the metabolism and protection against heavy metal toxicity³⁰. In addition to the detoxification of toxic metals such as Cd, MTs are involved in the maintenance of homeostasis of essential trace elements such as zinc and copper³¹. Metallothionein synthesis is considered as one of the

best-known biochemical detoxification mechanisms for metal and it is widely demonstrated that its induction may be influenced by metal contamination. This low- molecular weight protein binds to Cd, limits its availability to cell and tissues³². The MTs play a role in transport, detoxification and storage of Cd³³. After the absorption, Cd is bound to albumin and transported to liver³⁴. Following release from the liver, MT-bound Cd enters the plasma and appears in the glomerular filtrate, from where it is re-released intracellularly by renal tubule cells³⁵. At this stage, Cd is cleaved from the MT-Cadmium complex by lysosomal action and Cd^{2+} ions are re-excreted into the tubular fluid and finally eliminated in the urine.

FACTORS AFFECTING THE ABSORPTION OF CADMIUM IN FISH

Although metals are non-degradable (can neither be created nor destroyed)³⁶, they might be altered into more toxic forms or complex to more stable and less toxic compounds. Natural waters are usually contaminated with mixtures of metals and other toxic compounds. Accumulation of some metals in fish may be affected due to the occurrence of the others metals³⁷ and interactions among various metals may be related to their different affinities to various organs. In an aquatic environment, the impact of metals (perhaps the metal toxicity) can also be influenced by various abiotic environmental factors; e.g., water hardness, temperature and pH^{13,25,38,39}. Hardness is one of the most important factors that affect fish physiology and metal toxicity⁴⁰. Elevated dietary Ca^{2+} protected fish against both, dietary and waterborne Cd uptake⁴¹. The mobility and bioavailability of Cd in aquatic environments is enhanced under conditions of low pH, low hardness, low suspended matter levels, high redox potential and low salinity²⁴. Water acidification directly affects metal accumulation rates by the fish^{25,42}. Water acidification affects bioaccumulation of metals by the fish in an indirect way, by changing solubility of metal compounds or directly, due to damage of epithelia which become more permeable to metals²⁵. Therefore, toxicity of Cd generally increases with reducing water hardness and reducing concentrations of dissolved organic matter^{1,43}. Water temperature may cause the differences in metal deposition in various organs^{25,44} of freshwater fish. Humus content in water also contributes to determine the concentration of Cd in fish⁴⁵. Cadmium is more toxic in freshwater than in saltwater because it combines with chlorides in saltwater to form a molecule that is less available from solution⁴⁶⁻⁴⁷.

Mebane⁴⁸ shows that fish incubated in higher hardness water were about two times more resistant to Cd toxicity than the fish incubated in extremely soft water. However, the impacts of water hardness on Cd toxicity need further assessment and analysis because alkalinity varied with hardness levels in many of the toxicity experiments⁴⁹.

BIOACCUMULATION OF CADMIUM IN FISH

Bioaccumulation is the net result of the interaction of uptake, storage and elimination of a chemical. Nevertheless, it is a normal and essential process for the growth and nurturing of organisms. Fish daily bioaccumulate many vital nutrients, such as vitamins, trace minerals essential fats and amino acids. Many toxic organic chemicals are concentrated in biota several orders of magnitude greater than their aqueous concentrations and therefore, bioaccumulation can cause a serious threat to both the biota of surface waters and the humans that feed on these surface-water species. Cadmium that enters into the aquatic ecosystem may not directly impart toxicity to the organisms at low concentrations. Nevertheless, it can be accumulated in aquatic organisms through bio-concentration, via the food chain process and eventually threaten human health as they consume fish. Various species of fish living in the same aquatic environment may accumulate different amounts of metals in their tissues. Interspecies differences in metal accumulation may be related to living and feeding habits²⁵. Nakayama *et al.*⁵⁰ show that carnivorous fish (*Serranochromis thumbergi*) shows low accumulated levels of Cd in its tissues and heavy metal concentration is generally inversely correlated with the trophic level. However, Croteau *et al.*⁵¹ and Croteau and Luoma⁵² show that toxic effects of Cd are likely to occur with increasing trophic positions. However, there have been unequivocal evidences that the resistance differences are genetically based⁵³ and little is known about the effects of Cd on genetic and biochemical adaptive responses of aquatic species under chronic and long-term exposure⁵⁴. Toxicological studies at cellular level have shown that Cd inhibits the mitochondrial electron transfer chain and induces relatively oxygen species production⁵⁵.

TOXICITY OF CADMIUM IN FRESHWATER FISH

Cadmium as a toxic element might act as stress inducing agent for fish⁵³. Cadmium exposure may lead to the results of some pathophysiological damages including growth rate reduction in fish⁵⁶. Moreover, Cd produces both hepatic and renal injuries in mammals

and fish²⁸ with the potential to induce oxidative stress⁵⁷. It was also found to interfere with many protein and carbohydrate metabolism by inhibiting the enzymes involved in the processes⁵⁸ and known to perturb ion balance in teleost fishes⁵⁹. Furthermore, it also interacts with the calcium metabolism and causes abnormally low calcium levels (hypocalcaemia), probably by inhibiting calcium uptake from the water as discussed under the section “factors affecting the absorption of cadmium in fish”. However, high calcium concentrations in water protect fish from Cd uptake by competing at uptake sites. It would be chronically toxic when the animal is exposed over a long period of time at a lower concentration. Effects of long-term exposure can include larval mortality and temporary reduction in growth⁶⁰. Cadmium would be acutely toxic and animal may die from exposure to a high concentration over a short period of time. Cadmium may also accumulate in aquatic biota chronically exposed to sub-lethal concentrations⁴⁵. Chronic exposure can lead to mortality; sub-lethal effects such as reduced growth and reproductive failure are more common⁶⁰. As discussed above, the impact of Cd on aquatic organisms depends on a variety of possible chemical forms of Cd which can have different toxicities and bio-concentration factors. In most well oxygenated freshwaters that are low in total organic carbon, free divalent Cd will be the predominant form. The most bio-available form of Cd is also the divalent ion (Cd²⁺). Exposure to this form induces the synthesis of metallothionein which binds with Cd and decreases its toxicity^{46,47,61} and this normally takes place in the liver of fish. However, when the Cd concentration is high, the metallothionein detoxification system can become overwhelmed and the excess Cd will be available to produce toxic effects. Reproduction processes and early life stages of fish are the most sensitive for this elevated Cd levels. Skeletal deformities in fish can cause impaired ability of the fish to find food and avoid predators; hence, this sub-lethal effect becomes a lethal effect⁴⁶⁻⁴⁷. Interestingly, after exposing three freshwater species (*Cyprinus carpio*, *Carassius gibelio* and *Corydoras paleatus*) to different doses of Cd, Cavas and co-workers⁶² clearly shows that the frequencies of micronucleated and binucleated erythrocytes were elevated in peripheral blood, gill epithelial cells and liver cells giving further evidence that Cd has cytotoxic and genotoxic effects.

TREND OF ACCUMULATION OF CADMIUM IN FISH TISSUES

A growing number of evidences have shown that several factors influence Cd accumulation in fish tissues. Some studies have been carried out exposing fish to

different doses of Cd in laboratories under control conditions. Several studies have examined the relationship between metal exposure, accumulation and toxicity⁶³⁻⁶⁴. However, accumulated levels among organs varied following different treatment doses of Cd²⁺ and exposure times⁶⁵. In the natural environment, explaining the reasons that affect the Cd bioaccumulation may be difficult because of complex and unidentified time related changes.

Several studies show that tissue-specific Cd accumulation in fish with the chronic exposure⁶⁶⁻⁶⁷ but different tissues show different capacity for accumulating heavy metals⁶⁸. Metal distribution in various organs is also time-related and the effect of time on metal

distribution within the organism is a complex issue due to different affinity of various metals to the tissues of various fish species²⁵. In addition, they show that the accumulation of metals in different organs of fish is a function of uptake and elimination rates and metal concentrations in various organs may change during and after exposure. Most of these investigations of tissue level, Cd have been carried out to quantify this particular metal in several common fish tissues such as liver, kidneys, gills, muscles and alimentary canal. Cadmium shows different affinity to these fish tissues. According to many studies, Cd is accumulated primarily in the kidney and liver but it may reach high concentrations in the gill, alimentary canal and muscles as well (Table 1).

Table 1: Accumulating tendency of Cadmium in different tissues of some freshwater fish species in their natural environments

References	Fish species	Common name	Tissues considered	Highest Cd level	Lowest Cd level	Research location
Ahmed <i>et al.</i> (2012) ⁷²	<i>Sperata aor</i>	Long-whiskered catfish	M,K,G,Ac,L	Ac	L	Dhaleshwari river, Bangladesh
Akan (2009) ⁸²	<i>Clarias anguillaris</i>	Mudfish	L,G,K,Ac,B	L	Ac	Lake Chad, Nigeria
Alhashemi <i>et al.</i> (2012) ⁸³	<i>Barbus grypus</i>	Shirbot barb	M,G,Go,L	L	M	Southwest Iran
Ambedkar and Muniyan (2011) ⁷³	<i>Tilapia mossambica</i>	Mozambique tilapia	L,K,G,M,Ac	L	K	Kollidam river, Tamilnadu
	<i>Heteropneustes fossilis</i>	Stinging catfish	Ac,L,K,G,M	Ac	M	
Anim-Gyampo <i>et al.</i> (2013) ⁸⁴	<i>Sarotherodon galilaeus</i>	Mango tilapia	M,G	M	G	Tono reservoir, Ghana
Brázová <i>et al.</i> (2012) ⁸⁵	<i>Perca fluviatilis</i>	European perch	K,L,M,B	K	M	Ru_in reservoir, Slovakia
Chi <i>et al.</i> (2007) ⁸⁶	<i>Aristichthys nobilis</i>	Bighead carp	L,M	L	M	Lake Taihu, China
Coulibaly <i>et al.</i> (2012) ⁸⁷	<i>Sarotherodon melanotheron</i>	Black-chinned tilapia	L,G,K,M	L	M	Ebrie lagoon, Ivory Coast
Cyrille <i>et al.</i> (2012) ⁸⁸	<i>Sarotherodon melanotheron</i>	Black-chinned tilapia	M,G,L	M	G,L	Aby lagoon system, Cofe d'Ivoire
Demirak <i>et al.</i> (2006) ⁸⁹	<i>Leuciscus cephalus</i>	Chub	G,M	G	M	Southwest Turkey
Dimari and Hati (2009) ⁹⁰	<i>Heterotis niloticus</i>	African bonytongue	L,Ac,G	L	G	Lake Alau, Nigeria
Duman and Kar (2012) ⁹¹	<i>Squalius cephalus</i>	European chup	L,G,M	L	M	Yamula dam lake, Turkey
Ebrahimpour <i>et al.</i> (2011) ⁹²	<i>Carassius gibelio</i>	Prussian carp	L,K,G,Ac,M	L	M	Anzali, Iran
	<i>Esox lucius</i>	Northern pike	L,K,G,Ac,M	L	M	
Eneji <i>et al.</i> (2011) ⁷⁴	<i>Tilapia zilli</i>	Redbelly tilapia	G,Ac,M	G	M	Benue river, Nigeria
Eneji <i>et al.</i> (2011) ⁷⁴	<i>Clarias gariepinus</i>	African catfish	G,Ac,M	Ac	M	
Fatma (2008) ⁹³	<i>Lates niloticus</i>	Nile perch	L,G,Ac,M	L	M	Lake Nasser, Egypt
Gwaski <i>et al.</i> (2013) ⁹⁴	<i>Polypterus ansorgii</i>	Guinean bichir	G,K,L,Ig,M	L	K,M	Lake Chad, Nigeria
	<i>Clarias anguillaris</i>	Mudfish	G,K,L,Ig,M	Lg	K	
	<i>Oreochromis niloticus</i>	Nile tilapia	G,K,L,Ig,M	L	K	
Gwaski <i>et al.</i> (2013) ⁹⁴	<i>Synodontis nigrita</i>	catfish	G,K,L,Ig,M	G	L,M	Lake Chad, Nigeria
	<i>Bagrus docmac</i>	Semutundu catfish	G,K,L,Ig,M	L	M	
	<i>Lates niloticus</i>	Nile perch	G,K,L,Ig,M	M	L	
Has-Schön <i>et al.</i> (2006) ⁹⁵	<i>Cyprinus carpio</i>	Common carp	K,G,L,M,Go	K	M	Neretva river, Croatia
Liu <i>et al.</i> (2012) ⁹⁶	<i>Ctenopharyngodon idellus</i>	Grass carp	L,G,K,M	L	M	Southeast China
Mahesh <i>et al.</i> (2010) ⁹⁷	<i>Etioplos suratensis</i>	Pearlspot		K		Kerala, India
Malik <i>et al.</i> (2010) ⁹⁸	<i>Labeo rohita</i>	Rohu	L,K,G,M	L	M	Lake Bhopal, India
	<i>Ctenopharyngodon idella</i>	Grass carp	G,L,K,M	G	M	
	<i>Labeo rohita</i>	Rohu	L,K,G,M	L		
Nwani <i>et al.</i> (2010) ⁹⁹	<i>Tilapia zillii</i>	Redbelly tilapia	G,M	G	M	Afikpo, Nigeria
Obasohan (2007) ¹⁰⁰	<i>Parachanna obscura</i>	African snakehead	G,L,M	G	M	Ogba River, Nigeria
Rauf <i>et al.</i> (2009) ¹⁰¹	<i>Catla catla</i>	Catla	L,Sk,K,Sc,M,G	L	G	Ravi river, Pakistan
	<i>Cirrhina mrigala</i>	Mrigal	L,Sk,K,Sc,M,G	L	G	
	<i>Labeo rohita</i>	Rohu	L,Sk,K,Sc,M,G	L	G	
Sönmez <i>et al.</i> (2012) ¹⁰²	<i>Capoeta capoeta umbra</i>	-	M,L,G		M	Karasu river, Turkey
	<i>Chalcalburnus mosullensis</i>	-	M,L,G		M	
Squadrone <i>et al.</i> (2013) ¹⁰³	<i>Silurus glanis</i>	European catfish	M,L,K,G	K		Italian rivers
Tawee <i>et al.</i> (2011) ¹⁰⁴	<i>Oreochromis niloticus</i>	Nile tilapia	L,M,G	L	M	Langat river, Malaysia
Tekin-Özan (2008) ¹⁰⁵	<i>Cyprinus carpio</i>	Common carp	M,L,G	L	M	Bey'ehir kale, Turkey
Ural <i>et al.</i> (2012) ¹⁰⁶	<i>Capoeta umbra</i>	-	G,L,K,H,M	K		Uzunçayir dam lake, Turkey
Uwem <i>et al.</i> (2013) ¹⁰⁷	<i>Tilapia zillii</i>	Redbelly tilapia	B,L,G	L	G	Cross river state, Nigeria
	<i>Oreochromis niloticus</i>	Nile tilapia	B,L,G	L	B	
	<i>Schilbe mystus</i>	African butter catfish	B,L,G	L,B	G	
Yousafzai <i>et al.</i> (2010) ¹⁰⁸	<i>Wallago attu</i>	Wallago	G,M,L,Sk,Ac	G	Ac	Kabul river, Pakistan

Ac: Alimentary canal, G: Gills, K: Kidneys, L: Liver, M: Mussels, Sc: Scales, Sk: Skin, H: Heart, B: Bone, Br: Brain, Lg: Lung

Contamination of freshwater ecosystems by Cd and the accumulation and toxicity in aquatic animals through both waterborne and dietary routes are more concerned by the experts. During dietary administration of metals, their concentrations in the digestive tract increase and remain high until the end of exposure and rapidly decrease during depuration²⁵. Therefore, in diet-exposed fish, Cd concentrations were generally higher in the gastrointestinal tract than in the gills whereas in case of waterborne exposure, Cd concentrations are generally higher in gills and kidneys⁶⁹. Another study on *Oncorhynchus mykiss* (rainbow trout) has demonstrated that waterborne Cd causes toxicity in freshwater fish by inducing hypocalcemia because Cd²⁺ ions compete with waterborne Ca²⁺ ions for the active bronchial uptake pathway which normally ensures internal homeostasis of calcium levels⁷⁰. Therefore, increases in waterborne calcium concentrations protect against waterborne Cd uptake and toxicity in both acute and chronic exposures. Natural waters are concerned; water-borne calcium may affect the detected Cd levels in gills and other organs. Therefore, detected Cd levels in fish may be limited through the bronchial pathway. Wood and co-workers⁷⁰ also state that dietary Cd can protect against diet-borne Cd exposure, although the physiological mechanisms appear to differ from those at the gills. Surprisingly, the principal site of this inhibitory action of dietary calcium on gastrointestinal Cd uptake appears to be the stomach which is also the major site of gastrointestinal calcium uptake. Thus, this can also affect the Cd levels in the fish alimentary canal. However, the toxicity of Cd may be different according to waterborne or dietary pathways. As an example, Sajid and Muhammad⁷¹ stated that after investigating the sensitivity of fish towards toxicity of water-borne and dietary Cd, the dietary Cd were significantly less toxic than that of water-borne Cd.

Cadmium levels in various freshwater fish tissues according to the studies carried out by several researchers in different locations are given in Table 1. As an example, Ahmed and co-workers⁷², Ambedkar and Muniyan⁷³ and Eneji and co-workers⁷⁴ detected highest Cd levels in the alimentary canal of fish (Table 1). In contrast, several other studies show that Cd accumulation in the gills is higher compared to other tissues (Table 1). These observations reveal that even if the liver and kidney are considered as the prime sites of metal accumulation, gills can be more vulnerable than any other organs in some situations. Although the fish liver is considered as a good monitor of water pollution with metals since their concentrations accumulated in this organ are often proportional to those present in the environment⁷⁵. Twardowska *et al.*²⁵ Showed in case of

waterborne and dietary exposure, Cd concentrations in the gills and the digestive tract increase rapidly. However, the liver accumulates high concentrations of metals, irrespectively of the uptake route. Although the metal accumulation is primarily depends on waterborne and dietary pathways, some studies show the relationship of Cd level in fish tissues with the age and size of fish. Ciardullo *et al.*⁷⁶, Farkas *et al.*⁷⁷ and Giguere *et al.*⁷⁸ show that the age of the fish a potentially confounding factor when studying Cd bioaccumulation because Cd concentrations in liver and kidney increase with the age.

CONCLUSION

Cadmium and other toxic heavy metal in water impair feral populations of freshwater fish by altering their reproductive and physiological behaviors and abilities which ultimately affect the environmental permanence and fish diversity. There has been an increasing interest in the utilization of fishes as bioindicators of the integrity of aquatic environmental systems^{79,80,81}. The measurement of bioaccumulation and total protein in fish tissues may prove to be useful in biomonitoring of exposure to aquatic pollutants. In the developing world, especially in South East Asia, fish is one of the main sources of protein and provides a significant contribution to the diet of the rural communities. With increased urbanization, industrialization and human population, there has been a rapid increase in the municipal waste water (sewage water and industrial effluents) and use inorganic agricultural fertilizers (phosphate fertilizers) which in turn has intensified the environmental pollution. As a result, metal bioaccumulation is a major route by which increased levels of the pollutants are transferred across food chains. Affinity of freshwater fish species to Cd is diverse (Table 1). This may be due to different reasons such as physiological, biological or genetic changes. Cadmium residues accumulated in different organs are proportional to those present in the environment. Both waterborne and dietary pathways are responsible for uptake of Cd from their environments. At the same time, different species of fish show disparities in Cd accumulation even in the same environment. Similarly, the effect of Cd can also influence by abiotic environmental factors. As far as fish tissues are concerned the liver shows higher affinity towards Cd rather than kidneys and gills (Table 1). Accumulation tendency of Cd by the muscle tissue which consists of mainly the edible parts of a fish is comparatively low according to the studies carried out by many authors.

FUTURE DIRECTIONS

Metal accumulation and their effects on fish are more diverse and complex. Moreover, the dynamic nature of aquatic systems makes it more difficult to understand. Because of time to time changes in the type and strength of pollution, the so called effect of Cd on freshwater fish may be more challenging to explain. However, individual close investigations are required especially on commercial and threaten fish species under different water quality conditions because the amount of absorption of Cd and assembling depends on ecological, physical, chemical and biological condition and physiology of organisms. The understanding of the interaction between water quality conditions and Cd in fish as well as the occurrence of the others metals and interactions among various metals with Cd are not enough to close the gaps between Cd accumulation in freshwater fish and possible environmental impacts. Similarly, comprehensive understanding of the species-level temporal changes of Cd accumulation in freshwater fish in their natural environments help to protect them from the future threats. Further investigations are necessary to understand the genetic influence on Cd bioaccumulation. Identification of genetic reasons is useful to improve more resistant species in the future. Knowledge of the form of Cd accumulation is a prerequisite to understand why freshwater fish accumulate trace metals (Cd) to such different body concentrations.

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