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Antioxidant Activity of the Water-Soluble and Alkali-Soluble Polysaccharides from Chinese Truffle *Tuber sinense*

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Abstract: To deeply explore the pharmacological activities of the Chinese truffle *Tuber sinense* (CT) in this study, three kinds of the crude polysaccharides from Chinese Truffle *Tuber sinense* (CTCP) were extracted by the method of water extracting-alcohol precipitating which include Water-soluble Polysaccharide (W-CTCP) and Alkali-soluble Polysaccharides (A-CTCPI, A-CTCPII) and then the polysaccharide content of the CTCP were determined. The antioxidant activities of the CTCPs were evaluated by reducing power, superoxide radical and hydroxyl radical free radical-scavenging assay. As the results, the polysaccharide (W-CTCP, A-CTCPI and A-CTCPII) content was 46.76, 74.05 and 57.68%, respectively. The polysaccharides all exhibited antioxidant activities *in vitro* and W-CTCP showed the strongest activity. It can be concluded that the polysaccharides extracted from CT should be explored as a potential natural antioxidant agent for use in functional foods or medicine.

Key words: Crude polysaccharide, Chinese truffle *Tuber sinense*, antioxidant, superoxide radical, medicine

INTRODUCTION

Truffle, the fruiting bodies of ascomycetous fungi belonging to the genus *Tuber* is found worldwide (Hawksworth *et al.*, 1995). Truffles are hypogeous fungi which live in symbiosis with trees and some shrubs. Previous reports have showed that truffles of Saudi Arabia contained protein, amino acids, fat, crude fiber, ash and ascorbic acid, zinc, manganese and iron (Sawaya *et al.*, 1985). Many ingredients from the fruiting-bodies of truffles have been isolated and they showed important biological activities (Gao *et al.*, 2002; Hu *et al.*, 1994).

Polysaccharide is one of the most important compounds in mushrooms. A number of reports have showed that polysaccharides have various biological activities such as antioxidant, anti-tumor, anti-inflammatory and immunomodulation (Kim et al., 2008; Mizuno et al., 1996; Wu et al., 2010). Recent research has indicated that truffle polysaccharide extracted by water extracting exhibit antioxidant activity but there are no reports on antioxidant activity of truffle polysaccharides extracted by alkaline. In present study, three kinds of crude polysaccharides from Chinese truffle *Tuber sinense* were obtained by water-extracting and alkaline-extracting and then we evaluated the antioxidant activities of the

CTCPs by reducing power, superoxide radical and hydroxyl radical free radical-scavenging assay. Understanding of the antioxidant activities of the CTCPs would allow for development of a new natural antioxidant agent for use in functional foods or medicine.

MATERIALS AND METHODS

Extraction of the water-soluble and alkali-soluble polysaccharides from Chinese truffle Tuber sinense: The fruiting bodies of Chinese truffle Tuber sinense were obtained from a commercial market at Panzhihua, Sichuan province, China. The Traditional Water Extracting-Alcohol Precipitating Method was used to extract polysaccharides from the truffle. In brief, the fruiting bodies of the truffle were crushed into powder and dried at 60°C. Then, the powder was extracted by 95% ethanol to remove the pigments, fat and inactivate enzymes. After filtering, the residues were air-dried and then boiling in distilled water for three times and 2 h for each time. By filtration, the debris was collected for the extraction of the Alkali-soluble Polysaccharide I (A-CTCPI) and the filtrate was concentrated in a rotary evaporator at 60°C. Four volumes of ethanol were added to the extract and the mixture was left standing 12 h. The precipitate was washed successively with ethanol and acetone then dried

at 50°C to obtain the Water-soluble Polysaccharide (W-CTCP). To extract A-CTCPI, the above-mentioned debris was dispersed for 12 h into 0.5 mol L⁻¹ NaOH solution under stirring at room temperature. The extraction solution was filtered and the residues were collected for the extraction of the Alkali-soluble Polysaccharide II (A-CTCPII). The extract was neutralized with hydrochloric acid then was concentrated in a rotary evaporator at 60°C. The remaining steps were the same as the preparation of W-CTCP. To extract A-CTCPII, residues collected from the extraction procedure of A-CTCPI were dispersed for 12 h into 1.0 mol L⁻¹ NaOH solution under stirring at room temperature. The extraction solution was filtered and the extract was neutralized with hydrochloric acid. The supernatant containing A-CTCPII was concentrated, ethanol precipitation and then dried. The total carbohydrate content of the water-soluble and alkalisoluble polysaccharides from Chinese truffle Tuber sinense was estimated by the Phenol-Sulfuric Acid Method (DuBois et al., 1956). The reducing sugar content was determined according to the method of Hodge and Hofreiter (Hodge and Hofreiter, 1962). The crude polysaccharide content was the subtraction of reducing sugar from total carbohydrates.

Hydroxyl radical-scavenging assay: The hydroxyl radical-scavenging activity was measured using the method reported by Liu *et al.* (2010) with slight modification. Hydroxyl radicals were generated by Fenton reaction in the system of FeSO₄ and H₂O₂. The reaction mixture that contained polysaccharides 1 mL was incubated with phenanthroline (7.5 mM and 1 mL), distilled water (1 mL), ferrous sulfate (0.75 mM and 1 mL) and hydrogen peroxide (0.01% and 1 mL) in phosphate buffer (20 mM and pH 7.4) for 30 min at 37°C and the absorbance was read at 510 nm. The scavenging activity of hydroxyl radical was expressed using the following equation:

Hydroxyl radical-Scavenging activity (%) = $\frac{A_s - A_1}{A_0 - A_1} \times 100$

Where:

 A_s = Absorbance of the sample

A₁ = Absorbance of the control solution containing 1,10-phenanthroline, FeSO₄ and H₂O₂

A₀ = Absorbance of the blank solution containing 1,10-phenanthroline and FeSO₄

Superoxide radical-scavenging assay: The scavenging effect on superoxide radicals was determined according to the method of Li *et al.* (2011) and Ding *et al.* (2010). Briefly, 4.5 mL Tris-HCl buffer (50 mM and pH 8.2), 1 mL polysaccharide solution were mixed and incubated at 25°C

for 20 min and then 0.2 mL and 3 mM pyrogallol solution was added to initiate the reaction the change speed of absorbance (A/min) of the reactive solution was measured at 325 nm against the blank. The capability of scavenging superoxide radical was calculated using the following equation:

Superoxide radical-Scavenging activity (%) = $\frac{V_0 - V_1}{V_0} \times 100$

Where:

 V_0 = The change speed of absorbance of the control group in the superoxide radical generation system

V₁ = The change speed of absorbance of the test sample

Reducing power: The reducing power was evaluated according to the method of Deng *et al.* (2011). In brief, 1 mL polysaccharides solutions at different concentration was mixed with 2.5 mL phosphate buffer (0.02 mol mL⁻¹, pH 6.6) and 2.5 mL of ferricyanide solution (10 g L⁻¹). The mixture was incubated at 50°C for 20 min and then 2.5 mL trichloroacetic acid solution (100 g L⁻¹) 2.5 mL distilled water and 0.5 mL ferric chloride solution (0.5 ML, 1 g L⁻¹) was added. The absorbance of the mixture was read at 700 nm.

RESULTS AND DISCUSSION

Extraction of crude polysaccharides from Chinese truffle *Tuber sinense* and content analysis: The water-soluble and alkali-soluble polysaccharides from Chinese truffle *Tuber sinense* were extracted by water and alkaline. As shown in Table 1 the contents of the polysaccharides in W-CTCP, A-CTCP I and A-CTCP II were determined as 46.76, 74.05 and 57.68%, respectively. A-CTCP I showed the highest polysaccharide content while W-CTCP showed the highest reducing sugar content.

Hydroxyl radical-scavenging activity: As shown in Fig. 1 W-CTCP was found to have stronger scavenging activity of hydroxyl radical than A-CTCP I and A-CTCP II. W-CTCP exhibited hydroxyl radicals scavenging activity in a concentration-dependent manner. Scavenging effects of W-CTCP were $13.71 \sim 83.92\%$ at concentration of $0.1-2.4 \, \mathrm{mg \, mL^{-1}}$. A-CTCP I and A-CTCP II showed a weak

Table 1: Determination of the content of polysaccharides from Chinese truffle *Tuber sinense*

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Sample name (%)	W-CTCP	A-CTCP I	A-CTCP II
Reducing sugar content	5.73	0.00	0.00
Polysaccharide content	46.76	74.05	57.68
The total sugar content	52.49	74.05	57.68

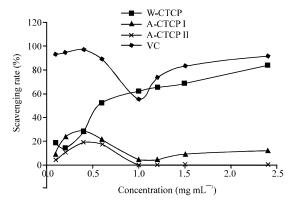


Fig. 1: Hydroxyl radicals scavenging activity of polysaccharide from CT

scavenging ability and both of them have higher scavenging activities on hydroxyl radical at low concentration (<1 mg $\rm mL^{-1}$) than at higher concentration (1~2.4 mg $\rm mL^{-1}$).

Superoxide radicals scavenging activity: The three polysaccharides were found to have the ability to scavenge superoxide radical at concentration between 0 and 2.4 mg mL⁻¹ (Fig. 2). These results indicated that the water-extracting polysaccharide had a stronger superoxide radical-scavenging activity than the alkali-extracting polysaccharides. The superoxide radicals scavenging effects of the polysaccharides increased with increasing concentration. W-CTCP showed good superoxide anion scavenging activities (45.9%) at the concentration of 2.4 mg mL⁻¹ while A-CTCP I and A-CTCP II were only 13.5 and 35.98% at the same concentration, respectively.

Reducing power: A higher absorbance of the reaction mixture indicated greater reducing power. As shown in Fig. 3, the reducing power of the polysaccharides increased with increasing concentration between 0 and 0.6 mg mL⁻¹. Though the reducing power of all samples was low in the test concentration, the water-extracting polysaccharide had a better reducing power than alkaliextracting polysaccharides at the same concentration. The reducing power of W-CTCP, A-CTCP I and A-CTCP II at 0.6 mg mL⁻¹ were 0.332, 0.07 and 0.078, respectively which were much weaker than those of ascorbic acid.

The oxidative damage, associated with reactive oxygen species is believed to be involved in many illnesses such as diabetes mellitus, arterisclerosis and cancer (Halliwell and Gutteridge, 1984). It is important to suppress the reactive oxygen species for treatment of the illness listed above. Polysaccharides as one of the most

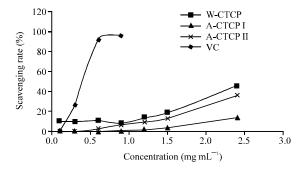


Fig. 2: Superoxide radicals scavenging activity of polysaccharide from CT

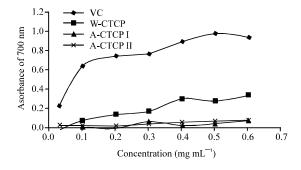


Fig. 3: Reducing power of polysaccharides from CT

important components of mushroom have been reported to have various bioactivies such as antioxidant, antitumor and immunodulatory activites (Li et al., 2011; Kim et al., 2008; Mizuno et al., 1996). For exmaple, the polysaccharides from Ganoderma atrum, Dictyophora indusiata, Grifola frondosa, Hericium erinaceus and Tricholoma giganteum, Tricholoma matsutake exhibited the antioxidant activity (Mau et al., 2002; Ding et al., 2010; Chen et al., 2008). In present study, three kinds of polysaccharides from Chinese truffle Tuber sinense were extracted by water and alkaline and several in vitro assays were applied to evaluate the antioxidant potential of the polysaccharides. W-CTCP exhibited excellent antioxidant activities and it is consistent with the reports of Jin-Zhong etal.(2011).Alkaline extracting polysaccharides also exhibited antioxidant acitivity but compared with W-CTCP, A-CTCP I and A-CTCP II had weaker antioxidant activities. The differences in antioxidant activity may be due to different structure and components. Meanwhile, the results also suggested that the pH of extraction solvent played an important role on bioactivity of polysaccharides. But the exact mechanism of the antioxidant activity of the polysaccharides from truffle has not been clarified and need further research.

CONCLUSION

This study shows that three kinds of polysaccharides from Chinese truffle all exhibited antioxidant activity *in vitro* and W-CTCP have highest antioxidant activity among them. Polysaccharides of Chinese truffle could be explored as a potential natural antioxidant agent for use in functional foods or medicine.

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