

Radiometric Survey of River Osun-Osogbo in Osun State of Nigeria

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Abstract: Survey for radiometric minerals has become important over the last few decades because of the demand for alternative to present common source of energy. Radiometric survey is one of the geophysical techniques in use in exploration for geothermal energy, which is generated mainly from the decay of long-lived radioactive isotopes. Ten fresh rock samples were collected from Osun-Osogbo river. This is to determine their radioactive heat production and the contribution of each radionuclide content. The radiogenic heat production was determined in the laboratory using NaI(Tl) gamma-ray spectrometer. The results shows that the contribution and rate of heat production of ⁴⁰K, ²³⁸U and ²³²Th in the samples vary significantly with lithology. (rock type) and rock samples OS4 (mica) is associated with high heat production uranium.

Key words: Radiogenic heat, radionuclide, gamma-ray spectrometer, photopeaks, geothermal energy

INTRODUCTION

The Earth's internal heat derives from several sources but there are 2 main sources. One of the sources is the cooling of the Earth since its early history, when internal temperature were much higher than they are now. The other source is the heat produced by the decay of long-lived radioactive isotopes. This is the main source of the Earth's internal heat, which in turn, powers all geodynamic processes (Philip, 2005). The Earth is constantly losing heat from its interior, which is many times larger than the energy lost by other means, such as the changes in Earth's rotation and energy released in earthquakes. The interior of the Earth is losing heat viz geothermal flux at a rate of about 4.4×10^3 W, which is equivalent to 1.4×10^{21} Jyr⁻¹.

Geophysical methods play a key role in geothermal exploration. The geophysical surveys are directed at obtaining indirectly, from shadow depth, the physical parameters of the geothermal systems. The various geophysical techniques in use in exploration for geothermal energy include subsurface (Shallow) temperature measurement (Lachenbruch and Sass, 1977; Kintzinger, 1956; Lee, 1977; LesSchach and Lewis, 1983; Ranmingwong *et al.*, 2000); Geochemical thermometric method (Sigvaldason, 1973; Rajver, 2000). Electrical methods (Bandwell and MacDonald, 1965; Anderson and Johnson, 2000; Pertamina, 1997; Tripp and Ros, 1997); Magnetotelluric methods (Johnson, 1992;

Ushijima *et al.*, 2000); Gravity method (Johnson, 1995; Sumintadireje *et al.*, 2000) Aeromagnetic and magnetic surveys (Reynolds *et al.*, 1990; Salem *et al.*, 1999, 2000); Sismic Method (Keller, 1981; Rajver *et al.*, 1996) and Radioactive Method (Pasquale *et al.*, 1997; Louden and Mareschal, 1996). Each of these methods has its own advantages and disadvantages. Some lack the maturity under difficult conditions while others become less useful for deep exploration because of lack of sensitivity. Considering the limitations of the various methods, it is probably necessary to use an integrated geophysical approach employing a wide variety of techniques.

Non renewable energy sources like the fossil fuels (Crude oil, natural gas, coal, coke, etc) are daily depleting fast. As a matter of expediency, there is a strong need to replace non renewable energy sources with alternative renewable sources. This is the strong impetus for this current study on geothermal exploration.

In this study, we use the radioactive method, which involves measuring the concentration of radioactive elements: Potassium (⁴⁰K), Uranium (²³⁸U) and Thorium (²³²Th), using Gamma-ray spectrometer. The gamma-ray spectrometry method is widely use in Earth's Sciences for the determination of naturally occurring radioactive materials. Heat produced by radioactive decay in rocks is of the fundamental importance in understanding the thermal history of the Earth and interpreting the continental heat flux data (Chiozzi *et al.*, 2000, 2007).

Theory: Energy released by short-lived radioactive isotopes may have contributed to the initial heating, but the short-lived isotopes would be consumed quite early. The heat generated by long-lived isotopes has been an important heat source during most of Earth's history.

In order to be a significant source of heat a radioactive isotope must have a half-life comparable to the age of the Earth, the energy of its decay must be fully converted to heat and isotope must be sufficiently abundant. The main isotopes that fulfill these conditions are ^{238}U , ^{235}U , ^{232}Th and ^{40}K . The isotope ^{235}U has a shorter half-life than ^{238}U and release more energy in its decay.

The heat Q , produced by radioactivity in a rock that has concentrations C_u , C_{Th} and C_K , respectively, of these elements is

$$Q = 0.00348C_K + 95.2C_U + 25.6C_{Th} \text{ (Rybach } et al., 1988) \quad (1)$$

Heat can be transported by three process: Conduction, convection and radiation; conduction and convection require the presence of a material; radiation can pass through space or a vacuum. Conduction is the most significant process of heat transport in solid materials and thus it is very important in the crust and Lithosphere. However, it is an inefficient form of heat transport and when the molecules are free to move, as in fluid or gas, the process of convection becomes more important. Although the mantle is solid from the standpoint of the rapid passage of seismic waves, the temperature is high enough for mantle to act as a viscous fluid over long time intervals (Philip, 2005) consequently, convection is more important form of heat transfer than conduction in mantle. Convection is also the most important form of heat transport in the fluid core.

MATERIALS AND METHODS

Ten fresh rock samples were collected from different location of Osun-Osogbo river in Osun State, Nigeria. The rock samples were crushed to fine grains to minimize self-absorption and to have geometry and matrix. Each sample was carefully packed in a 391 g plastic container, sealed and weighed. They were then left for thirty days in order for gaseous members of Uranium and Thorium series reach secular equilibrium before counting.

Natural radionuclide of relevance for the radiogenic heat production are mainly ^{40}K and gamma-ray emitting nuclei in decay series of ^{238}U and ^{232}Th Gamma radiation analysis allows various gamma emitter to be distinguished and the quantitative content of potassium, Uranium and thorium to be calculated. Concentration of ^{40}K , ^{238}U and

^{232}Th are determined in the laboratory through spectrometry of emitted gamma rays using a cylindrical NaI(Tl) scintillator. The detector used is 7.6 by 7.6 cm NaI(Tl) detector (model No 802 series) by Canberra Inc. the gamma rays, which interact with the scintillator are converted into quanta of visible light, which can be detected with a photomultiplier produces voltages pluses with height proportional to the energy of the gamma rays. These pluses are amplified and fed to a multichannel analyzer (Canberra series 10 multichannel analyzer). All the samples were counted for 18000 sec, as this was considered adequate for measurement of the low activity of the samples. The efficiency and quantitative calibration of the apparatus was determined using a standard material prepared from Rocketdyne laboratories. California, USA. The photopeak area values were converted into concentration in Bqkg^{-1} and then later to part per million (ppm). These concentrations in ppm were used for determination of the radiogenic heat production was calculated using Rybach (Eq. 1) where C_U , C_{Th} and C_K are concentration in ppm of uranium, thorium and potassium, respectively. Multiplying the radiogenic heat production values by the rock density gives the radiogenic heat generated in cubic meter of the rock (Wm^{-3}).

RESULTS AND DISCUSSION

The total heat production represents the summation of the three isotopes for each sample and is a comprehensive parameter to reflect the rate of radiogenic heat rock samples (Table 1).

The radiogenic heat contribution of the isotopes ranges from 0.04 to $3.35 \times 10^{-6} \text{ Wm}^{-3}$ for ^{40}K , 0.13- $5.26 \times 10^{-6} \text{ Wm}^{-3}$ for ^{238}U and 0.04- 2.75 Wm^{-3} for ^{232}Th (Table 2). Overall result shows that uranium has the highest radiogenic heat contribution, which is in support of (Stacey, 1994; Stacey and Loper, 2007) results.

Mica (OS4) has the highest radiogenic heat production with the highest contribution from the isotopes. This is in support of the earlier work carried out by Alabi (2007) on rock samples from three different rivers of different location in Osun State. It is also observed that Mica is associated with high radiogenic heat contribution from uranium. Mica is closely followed by amphibolites, however, Mica has greater radiogenic heat production than amphibolites with approximately multiple of three folds.

Quartz is expected to have the highest radiogenic heat but it is found to ranked ninth, this is likely due to geological location because radiogenic heat contributions are unequal and different in the ocean and continent (Bott *et al.*, 1982).

Table 1: Isotopic concentrations and radiogenic heat production in samples collected from Osun-Osogbo River

Sample code	Lithology	Concentration (ppm)			Heat production (10^{11}Wkg^{-1})			
		^{40}K	^{238}U	^{232}Th	Potassium	Uranium	Thorium	Total
OS1	China clay	3428.72	0.79	0.87	11.93	74.81	22.29	109.03
OS2	Glauconite	462.07	0.17	0.30	1.61	15.96	7.71	25.28
OS3	Shale	2007.75	0.05	0.77	6.99	4.71	9.82	31.52
OS4	Mica	34370.56	1.97	3.83	119.61	187.41	98.08	405.10
OS5	Pegmatite	24764.86	0.50	0.15	86.18	47.85	3.80	137.83
OS6	Muscovite	31529.80	0.22	0.10	109.72	21.30	2.65	133.83
OS7	Quartz	1208.74	0.20	0.05	4.21	19.31	1.30	24.82
OS8	Synite	728.18	0.38	1.00	2.53	35.99	82.84	121.36
OS9	Amphibolite	28778.73	0.34	1.26	100.15	32.13	32.21	164.49
OS10	Mica	8504.87	0.81	0.24	29.60	77.03	6.02	112.65

Table 2: Radiogenic heat production in samples collected in micro watts per meter cube

Sample code	Lithology	Density (Kgm^{-3})	Heath production (10^6Wm^{-3})			
			Potassium	Uranium	Thorium	Total
OS1	China clay	2750	0.33	2.06	0.61	3.00
OS2	Glauconite	2700	0.04	0.43	0.21	0.68
OS3	Shale	2710	0.19	0.13	0.54	0.86
OS4	Mica	2800	3.35	5.25	2.75	11.35
OS5	Pegmatite	2740	2.36	1.31	0.10	3.77
OS6	Muscovite	2700	2.96	0.58	0.07	3.61
OS7	Quartz	2900	0.12	0.56	0.04	0.72
OS8	Synite	2700	0.07	0.97	2.24	3.28
OS9	Amphibolite	2640	2.64	0.85	0.85	4.34
OS10	Mica	2800	0.83	2.16	0.17	3.16
		Total	12.89	14.30	7.58	34.77

CONCLUSION

The concentration and radiogenic heat production by the 3 isotopes for each sample vary significantly. Highest concentration and heat contribution is recorded in Mica. Mica has greater heat production than Quartz, which is contrary to expectation; this might be as a result of geological location. Thus, Mica rock samples from the river produce more heat than any other rock type and they are associated with high heat contribution from uranium isotope.

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