

***In-vitro* Free Radicals Scavenging and Antioxidant Activity of Rice Bran Extract**

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ABSTRACT

Aim: To evaluate the antioxidant activities of the γ -oryzanol isolated from rice bran oil of Indian origin in various *in vitro* models. **Materials and Methods:** The hydrogen donating ability of the hydroalcoholic extract of γ -oryzanol was measured in the presence of 1, 1-diphenyl-2-picryl-hydrazyl (DPPH) radical. **Results:** A 100 $\mu\text{g mL}^{-1}$ of γ -oryzanol and ascorbic acid exhibited 64.56 and 64.88% inhibition, respectively and the IC_{50} values were found to be 48 and 56.50 $\mu\text{g mL}^{-1}$ for γ -oryzanol and ascorbic acid respectively. The effect of γ -oryzanol on reducing power was studied according to the reaction of Fe^{+3} to Fe^{+2} . The reducing power of the extract increased with the increasing amount of the concentration. The total polyphenolic content of the γ -oryzanol was tested using Folin-Ciocalteu reagent. It was found that γ -oryzanol contained 19.31 ± 1.79 mg of GA/g, which is significant ($p < 0.05$) when compared to gallic acid. γ -oryzanol hydrogen peroxide (H_2O_2) radical at IC_{50} values of 45 $\mu\text{g mL}^{-1}$ against the corresponding standards ascorbic acid ($\text{IC}_{50} = 35$ $\mu\text{g mL}^{-1}$). Moreover, the results were observed in a concentration dependent manner. **Conclusion:** All the above *in-vitro* studies clearly designate that the hydroalcoholic isolated γ -oryzanol has a significant antioxidant activity.

Key words: Antioxidant, DPPH, γ -oryzanol, rice bran extract

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INTRODUCTION

Oxygen free radicals and nitrogen free radicals are generated within the cells due to the various metabolic processes and mitochondrial energy production¹⁻⁵. These radicals are responsible for beneficial as well as deleterious response. Their beneficial role includes response to the noxious stimuli that is antagonizing infectious agents along with this, they are the part of cell-cell signaling processes⁶⁻⁹. In low or moderate concentration, they play vital role in mitogenic response plus they required for the activation of cyclic guanosine monophosphate (cGMP) as a second messenger¹⁰.

The destructive effect of these generated free radicals instigating potential biological damage is known as oxidative stress and nitrosative stress¹¹⁻¹⁸. This eventuates in biological systems when there is an excessive synthesis of Reactive Oxygen Species (ROS) or Reactive Nitrogen Species (RNS) on one side at the same time an insufficiency of enzymatic and non-enzymatic antioxidants on the other¹⁹⁻²⁸. This can be expressed in different way that, oxidative stress emerges from the metabolic processes that utilize oxygen and shows a disturbance in the equilibrium status of pro-oxidant or antioxidant reactions in living tissues. The excess ROS can vitiate cellular lipids, fatty acids, proteins, or DNA impeding their normal functions. This results into oxidative stress induction²⁹⁻³⁹. This induced oxidative stress can cause number of human diseases as well as in the ageing process. The insubstantial balance

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between beneficial and harmful effects of free radicals is a very key aspect of living organisms and is executed by mechanisms known as “redox regulation” (pro-oxidant-antioxidant homeostasis). The processes of “redox regulation” shields living tissues from numerous oxidative stresses formed within the cells and retain “redox homeostasis” via controlling the redox status *in-vivo*⁴⁰⁻⁴⁸.

Strenuous exercise causes oxygen utilization and that results into disturbance of intracellular redox regulation. The mitochondrial electron transport chains, polymorphonuclear, along with xanthine oxidase have been recognized as crucial sources of intracellular free radical generation centers during exercise. Reactive oxygen species constitute a significant defilement to the cellular antioxidant defense mechanism, like decreased storage of antioxidant such as vitamins and glutathione and increased tissue propensity to oxidative damage⁴⁹⁻⁵². Nevertheless, enzymatic and nonenzymatic antioxidants have signified great familiarization to acute and chronic exercise. The important balance between pro-oxidants and antioxidants propose that supplementation of antioxidants may be useful for physically active individuals under certain physiological conditions by imparting a larger protective margin⁵³⁻⁵⁷. Research from human and animal studies proved that strenuous physical exercise may produce a state wherein the antioxidant defense mechanisms of several tissues are affected by excess reactive oxygen. A wide range of physiological and dietary antioxidants exploit in concert to elude such a stress. Regular physical activity in combination with dietary supplement that provide adequate supply of a combination of appropriate antioxidants may be anticipated to yield desirable outcome⁵⁸.

Ramarathnam in 1989 first identified and reported that the components of rice bran as flavonoid, isovitexin, α -tocopherol and γ -oryzanol possess high antioxidant activity⁵⁹. γ -oryzanol is composed of mixture of phytosteryl ferulates and comprises of cycloartenyl ferulates, 24-methylenecycloartenyl ferulate and campesterol ferulate. This mixture has been studied for antioxidant activity and this effect has been achieved as it enables donation of electron and destroys the action of reactive oxygen species and reactive nitrogen species. γ -oryzanol blocks pyrogallol autoxidation moreover, γ -oryzanol is more effective than the synthetic antioxidants against hydroperoxide formation. Also, reports suggested that γ -oryzanol inhibits the cholesterol

oxidation and prevents tissues through stabilization of cellular metabolism preventing disease formation⁶⁰. But the reports of the γ -oryzanol isolated from Indian rice bran extract are less about its antioxidant property. So, the main objective of the present study was to isolate γ -oryzanol from Indrayani rice bran of Indian origin and to study the antioxidative potencies, scavenging activities against DPPH radical, reducing power and estimation of polyphenol contents.

MATERIALS AND METHODS

Chemicals: Standard γ -oryzanol was procured from Oryza Oil and Fat Chemical Co. Ltd. Japan. Toluene, ethyl acetate, methanol and other analytical grade reagents and chemicals were used in the study are purchased from Merck Specialties Private Limited, Mumbai, India. Double distilled water filtered via 0.45 μ filter paper was utilized in the research work. Microliter syringe (Hamilton, Bonaduz, Switzerland) was acquired from Anchrome Enterprises (I) Pvt. Ltd Mumbai, India. Precoated Thin Layer Chromatographic (TLC) silica gel plates were purchased from Merck, Bangalore, Karnataka, India, of Kieselgel 60, F-254, 0.2 mm. Precoated glass silica gel plates were utilized for preparative TLC. High-Performance Thin-Layer Chromatography (HPTLC) spectra analysis was carried out on a Linomat V (Camag, Muttenz, Switzerland). Butylated hydroxytoluene (BHT), L-ascorbic acid, gallic acid, 2, 2-diphenyl-1-picrylhydrazyl (DPPH), potassium ferricyanide, 2-deoxyribose, trichloroacetic acid (TCA) and ferric chloride were obtained from Sigma-Aldrich Co. LLC. Mumbai, India.

Extraction and isolation of γ -oryzanol Rice Bran Oil (RBO): The extraction of rice bran oil was carried out according to previously reported method⁶¹. Shortly, rice bran of Indrayani variety of rice was acquired from Radhe Shyam Poha mill, Pune. This bran was kept in a refrigerator before its use. Direct solvent extraction method was applied for getting crude rice bran oil from 1 kg rice bran. Petroleum ether (60-80) was used as the solvent for extraction and the solvent was properly kept for 24 h for getting more yields. This procedure was performed with same rice bran with fresh petroleum ether three times for getting more yields. These obtained three fractions were mixed with each other to get one petroleum fraction. The resultant fraction then concentrated on rotary evaporator to get oil. The percentage yield was 10.4% of petroleum ether extract.

This collected rice bran oil was kept in a refrigerator before its use. From this, 3 mL oil was mixed with 5 mL absolute ethanol. Ethanol-oil mixture was further vigorously mixed by using cyclo-mixer and kept aside for 5 min. The supernatant was collected. The same procedure was repeated four times, so that 25 mL of supernatant was obtained. All the ethanol fractions were mixed and concentrated on rotary evaporator. The resultant concentrate was used for HPTLC spotting, isolation and *in-vitro* antioxidant study of γ -oryzanol. Oil (lower layer) remained in the test tube was dissolved in 5 mL ethyl acetate to verify the complete extraction of γ -oryzanol.

Chromatography: A Camag High-pressure Thin-Layer Chromatography (HPTLC) system equipped with an automatic TLC sampler (Linomat 5) with TLC scanner 3 and integrated software Win-Cats version 4 was used for the analysis. The HPTLC was implemented on a pre-coated silica gel HPTLC 60F254 (20×10 cm) plate for the quantification of γ -oryzanol in concentrated fractions. The samples and the standards were spotted on the plate as 6 mm wide bands. The linear ascending development was performed in a camag twin trough chamber (20×10 cm), which was pre-saturated with 20 mL mobile phase with toluene: ethyl acetate: Methanol (15.0:1.7:3.3, v/v/v) for 15 min, at room temperature. The length of the chromatogram run was kept up to 80 mm. Subsequent to the development; the TLC plate was dried by using current of air and with a hair dryer. Quantitative evaluation of the plate was carried out in the absorption reflection mode at 317 nm using Win-Cats software. Isolation of active constituent from concentrated fraction was carried out by preparative HPTLC.

***In-vitro* antioxidant activity of 24-mCAF isolated from rice bran oil**

Free radical scavenging activity in 1, 1-diphenyl-2-picrylhydrazyl radical (DPPH) assay:

The antioxidant activity using the DPPH assay was evaluated in terms of hydrogen donating or radical-scavenging capability using the stable radical DPPH. The DPPH solution of 0.1 mM was prepared in ethanol and from this 1 mL was added to γ -oryzanol solution in ethanol at varying concentrations ranging from 10-100 $\mu\text{g mL}^{-1}$. These test solutions are kept at 37°C for incubation for 30 min. After 30 min of incubation, the absorbance of sample solutions was measured at 517 nm using ethanol as blank. Ascorbic acid was used as positive control for

the study. Lower absorbance of reaction mixture stipulates greater free radical scavenging activity. To get exact scavenging capacity of γ -oryzanol following mathematical formula has been used:

$$\text{DPPH scavenged (\%)} = (A_{\text{control}} - A_{\text{test}} / A_{\text{control}}) \times 100$$

where, A_{control} indicates the absorbance of control reaction while A_{test} is for the absorbance in the presence of γ -oryzanol. The antioxidant capacity of the γ -oryzanol was manifested as IC_{50} . Here, IC_{50} value was described as the concentration of γ -oryzanol in $\mu\text{g mL}^{-1}$ that inhibits the formation of DPPH radicals by 50%.

Reducing power assay: Various concentrations of γ -oryzanol (10-100 $\mu\text{g mL}^{-1}$) of 1 mL were mixed with 2.5 mL of 0.2 M sodium phosphate buffer of pH 6.6 plus 2.5 mL of 1% potassium ferricyanide. The mixture was kept for incubation at 50°C for 20 min. After incubation, 2.5 mL of 10% trichloroacetic acid (w/v) was added followed by centrifugation at 3000 rpm for 10 min. The upper layer 2.5 mL was taken and mixed with 2.5 mL deionised water plus 0.1% of 0.5 mL of ferric chloride was added. The absorbance of final mixture was analyzed on UV spectrophotometer at 700 nm by using sodium phosphate buffer having pH 6.6 with 0.2 M strength as blank solution. Higher absorbance at 700 nm denotes higher reducing power. The assay was performed in triplicate. Reducing power is indicated in Ascorbic Acid Equivalent (ASE) with unit ASE/mL that shows amount of ascorbic acid expressed in milli-Moles those reducing power is the same than that of 1 mL of sample.

Determination of total phenolic content: A 0.5 mL of sample i.e., 24-mCAF was mixed with 0.5 mL of Folin and Ciocalteu's phenol reagent. Above mixture was mixed thoroughly and was kept aside for three minutes. After 3 min, 3 mL of 2% sodium carbonate solution was added to the mixture. Final volume at 10 mL of mixture was made by adding sufficient amount of distilled water. This final mixture was allowed to stand for 2 h with intermediate shaking. The absorbance of the blue colored solution was read at 760 nm. Total phenolic content of 25-mCAF was performed in triplicate. The concentration of total phenolic compounds in the gallic acid was measured in terms of μg of gallic acid equivalent using an equation acquired from the standard gallic acid graph.

$$\text{Absorbance} = 0.0008 \times \text{gallic acid } (\mu\text{g})$$

Hydrogen peroxide scavenging assay: The assay was done by adding 0.1 mL of 1 mM of EDTA, 0.01 mL of 10 mM FeCl_3 , 0.1 mL of 10 mM H_2O_2 and 0.36 mL of 10 mM deoxyribose along with 1.0 mL of different dilutions of 24-mCAF ($10\text{--}100\ \mu\text{g mL}^{-1}$) dissolved in ethanol plus 0.33 mL of phosphate buffer of pH 7.4 of strength 50 mM and 0.1 mL of ascorbic acid in sequence. The mixture was kept for incubation at 37°C aside for 1 h. for the development of pink color to the incubated mixture was taken 1 mL and mixed with 1.0 mL of 10% TCA and 1.0 mL of 0.5% TBA (in 0.025 M NaOH containing 0.025% BHA). The absorbance of the final mixture was read at 532 nm. The hydroxyl radical scavenging activity of the compound is reported in terms of % inhibition of deoxyribose degradation and is calculated as per following formula:

$$\text{H}_2\text{O}_2 \text{ scavenged (\%)} = (A_{\text{cont}} - A_{\text{test}}) / A_{\text{cont}} \times 100$$

where, "A cont" is absorbance of the control reaction whereas "A test" is the absorbance in the presence of the sample of the compound.

Statistical analysis: Experimental results were expressed in Mean \pm SD of three measurements. The values $p < 0.05$ recorded were considered as significant.

RESULTS AND DISCUSSION

Preparative HPTLC of petroleum extract: The method of extraction of phytochemicals from rice bran is the most significant route in the progress of pharmaceutical development of any plant species which is identified or described to have medicinal value. Preparative HPTLC is one of the simplest, robust and precise and yet highly steadfast procedure in the desire of collecting pure isolated compound from the rice bran extract. The rice bran extract revealed the brown fluorescent band and light brown band at R_f 0.72 for γ -oryzanol under the TLC scanner at 317 nm. The clear, bright fluorescent band was confirmed by comparing with the bands of standard γ -oryzanol and having same R_f value was scraped and dissolved in absolute ethanol and filtered. Ethanol was evaporated under vacuum and γ -oryzanol was isolated. The same procedure was repeated to get sufficient quantity of γ -oryzanol.

In-vitro antioxidant activity of isolated compound:

A wide range of *in-vitro* schemes have been used to evaluate radical scavenging ability and antioxidant activity of a compound. Antioxidant capacity is broadly applied as a criterion for medicinal bioactive components^{62,63}. Different artificial entities have been utilized like 1, 1-diphenyl-2-picrylhydrazyl (DPPH), OH radicals scavenging activity, Fe^{3+} - Fe^{2+} reducing activity, ferrous ions (Fe^{2+}) chelating activity, H_2O_2 scavenging activity. In this study, the antioxidant activity of the γ -oryzanol was studied. The antioxidant activity of the γ -oryzanol has been studied in a succession of *in-vitro* tests namely: DPPH, free radical scavenging, reducing power, scavenging of superoxide anion radical-generated non-enzymatic system, hydrogen peroxide scavenging and metal chelating activities.

Inhibition of DPPH radical: With this method, it was promising to conclude the antiradical power of an antioxidant by measuring of a decrease in the absorbance of DPPH^\cdot at 517 nm. Color change from purple to yellow, the absorbance decreased when the DPPH^\cdot was scavenged by an antioxidant during donation of hydrogen to form a stable DPPH^- molecule. In the radical form, this molecule had an absorbance at 517 nm, which vanished after acceptance of an electron or hydrogen radical from an antioxidant compound to become a stable diamagnetic molecule. Table 1 demonstrate a significant ($p < 0.05$) reduce in the concentration of DPPH radical due to the scavenging ability of isolated γ -oryzanol and the standard ascorbic acid as a reference compound, conferred the maximum activity at all concentrations. A $100\ \mu\text{g mL}^{-1}$ of γ -oryzanol and ascorbic acid exhibited 64.56 and 64.88% inhibition, respectively and the IC_{50} values were found to be 48 and $56.50\ \mu\text{g mL}^{-1}$ for γ -oryzanol and ascorbic acid, respectively.

Table 1: DPPH radical scavenging activity of γ -oryzanol and ascorbic acid

Concentration ($\mu\text{g mL}^{-1}$)	Inhibition (%) (Extracts)	Inhibition (%) (Standard)
10	22.42 \pm 0.257	33.14 \pm 0.150
20	27.14 \pm 0.144	48.93 \pm 0.076
40	44.96 \pm 0.160	62.93 \pm 0.117
60	50.10 \pm 0.163	70.58 \pm 0.139
80	61.32 \pm 0.175	76.87 \pm 0.127
100	64.46 \pm 0.122	81.53 \pm 0.125

Data is presented as the percentage of DPPH radical scavenging, Mean \pm SD

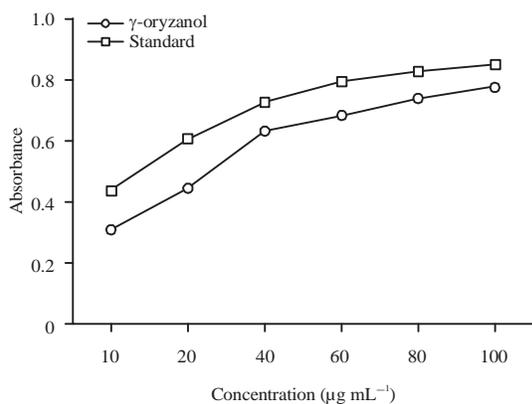


Fig. 1: Reducing power of γ -oryzanol isolated from rice bran oil. Each value is expressed as Mean \pm SD (n = 3)

Reducing ability: The reducing capability of a compound generally depends on the presence of reductants which have been demonstrating antioxidant potential by breaking the free radical chain and donating a hydrogen atom. The presence of reductants in γ -oryzanol causes the reduction of the Fe^{3+} /ferricyanide complex to the ferrous form. Hence, the Fe^{2+} can be monitored by measuring the production of Perl's Prussian blue at 700 nm. Figure 1 shows the reductive capabilities of the " γ -oryzanol" extract compared to ascorbic acid. The results on reducing power reveals the electron donor properties of γ -oryzanol thus neutralizing free radicals by forming stable products. The outcome of the reducing reaction is to conclude the radical chain reactions that may otherwise be very damaging.

Amount of total phenolic compounds: Phenolic compounds are identified as powerful chain breaking antioxidants. Phenols are very significant plant constituents because of their scavenging capacity owing to their hydroxyl groups. The phenolic compounds may provide directly to antioxidant action. It is recommended that polyphenolic compounds have inhibitory effects on mutagenesis and carcinogenesis in humans, when ingested up to 1 g daily from a diet rich in fruits and vegetables. The total phenolic content expressed in terms of Gallic Acid Equivalent (GAE) and was found to be $19.31 \pm 1.79 \text{ mg of GA g}^{-1}$. The total phenolic contents were calculated using the

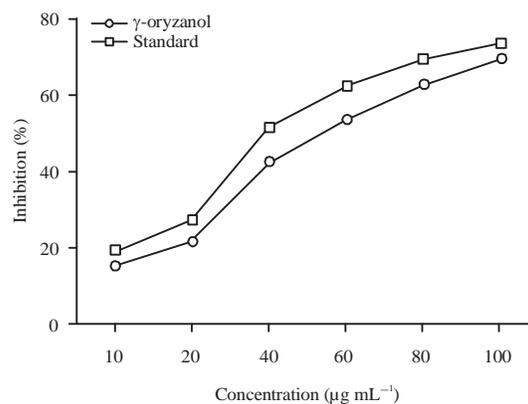


Fig. 2: Inhibition of hydroxyl radical by γ -oryzanol isolated from rice bran oil. Each value is expressed as Mean \pm SD (n = 3)

following linear equation based on the calibration curve of gallic acid:

$$y = 0.0025X + 0.013, R^2 = 0.964$$

where, A is absorbance and X is amount of gallic acid in μg .

Inhibition of hydroxyl radical: The highly reactive hydroxyl radicals can originate oxidative damage to DNA, lipids and proteins. The effect of γ -oryzanol on the inhibition of free radical-mediated deoxyribose damage was evaluated by means of the Iron (II)-dependent DNA damage assay. The Fenton reaction creates hydroxyl radicals (OH) which deplete DNA deoxyribose, using Fe^{2+} salts as an important catalytic component. Oxygen radicals may assault DNA either at the sugar or the base, giving production of large number of products. Figure 2 shows the effect of the γ -oryzanol on the Iron (II) dependent deoxyribose damage. The γ -oryzanol was capable of sinking DNA damage at all concentrations. ($\text{IC}_{50} = 53 \mu\text{g mL}^{-1}$). Ascorbic acid, used as a standard was highly effective in inhibiting the oxidative DNA damage, showing an $\text{IC}_{50} = 38.50 \mu\text{g mL}^{-1}$.

Inhibition of H_2O_2 radical: Hydrogen peroxide itself is not very reactive, but it can sometimes be toxic to cells since it may give rise to hydroxyl radical in the cells. Therefore, removing H_2O_2 as well as O_2^- is very significant for protection of food systems. The scavenging ability of γ -oryzanol on hydrogen peroxide is shown in Fig. 3 and compared with that of ascorbic acid as standard. γ -oryzanol at $45 \mu\text{g mL}^{-1}$ and α -tocopherol

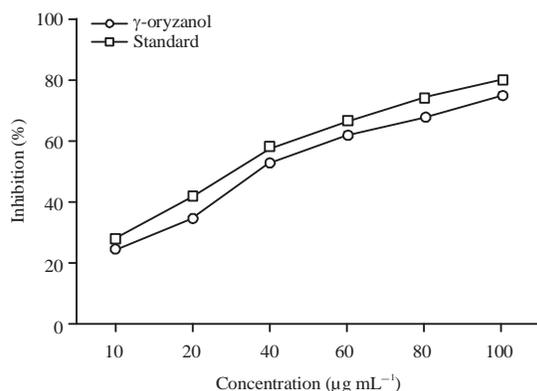


Fig. 3: Inhibition of H_2O_2 radical by γ -oryzanol isolated from rice bran oil. Each value is expressed as Mean \pm SD (n = 3)

at $35 \mu\text{g mL}^{-1}$ exhibited 50% scavenging activity, respectively, on H_2O_2 radical. There was statically significant correlation between γ -oryzanol value and standard ($p < 0.05$).

CONCLUSION

According to data obtained from the present study, γ -oryzanol isolated from Indian rice bran oil extract was found to be an effective antioxidant in different *in-vitro* assay including reducing power, DPPH radical. Hydrogen peroxide scavenging and metal chelating activities when it is compared to standard antioxidant compounds such as BHT; γ -oryzanol plays equal role as natural antioxidant. Based on the discussion above, it can be applicable for minimizing and preventing lipid oxidation in food products, retarding the formation of toxic oxidation products, maintaining nutritional quality and lengthening the shelf life of foods and pharmaceuticals.

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