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Assessment of the Technological and Environmental Factors Influencing the Ultra-High-Pressure Water Jetting Treatment of the Surface of Naval Construction Steel Against Corrosion

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Abstract: This research defines the best combination of factors (water pressure, the use or non-use of washing with fresh water after process, environmental factors (i.e., temperature, humidity, etc.) and type of epoxy paint) for the high pressure water jetting treatment of naval steels to guarantee the best performance against corrosion. Results show that the epoxy/iron oxide paint had the best performance when: a water-jetting pressure of 275.8 MPa is used, followed by a low pressure freshwater cleaning. Results also show the best results when the process is developed at the highest environmental temperatures and lowest humidity.

Key words: Water-jetting, cleaning, surface preparation, coatings, flash rust, corrosion, results, freshwater

INTRODUCTION

An adequate surface preparation is resquired to improve the adhesion of marine paints, especially on metal surfaces where corrosion (rusting) resistance is important. To this end, SandBlasting (SB) have been extensively applied. However, because of its consequences on both the human health and the environment its use has been reduced (Rosenberg *et al.*, 2006). Consequently, the UltraHigh-Pressure Water-Jtting (UHPWJ) technology in the fast track to become the standard technology of the surface preparation industry to remove paints from the surface of metals. Although, the are some limitations of UHPWJ as compared to SB greater global cost, not generating a new anchor prople and forming a thin oxide layer (flash rust) at the steel surface before applying the prime (Frenzel, 2010).

Towards an improved efficiency and competitiveness of UHPWJ different studies have been developed. Special attention has been paid to advancements in the nozzle technology that increased UHPWJ productivity to similar levels as compared to SB. The automation of UHPWJ technology combined with new equipment designs significantly contributed to this end (Schmid, 2005). In

fact, introducing semiautomatic systems resulted in over twice of the productivity of handwork operated UHPWI. There is also some research developed to modify the anchor profile at the steel surface (Teimourian *et al.*, 2010; Knapp and Taylor, 1996; Calve *et al.*, 2011).

The influence of oxides and dissolved salts concentration on the adherence and durability of paints, have been determined showing that a reduction of the flash rust area below 20-30% has a negligible impact on the primer performance (Kim *et al.*, 2008). A characterization of ash rusted surfaces with pressures over 206 MPa can be found by Ault (2010, 2014) including some alternatives to reduce flash rusting. However, the best combination of the working pressure with the environmental parameters and the type of primer that ensure the best coating performance needs to be further discussed.

This study aims at defining the best combination of the technological and environmental factors (water pressure, the or not for a subsequent washing with fresh water, environmental conditions, kind of epoxy coating) that results in the best performance of naval steels in corrosive environments.

MATERIALS AND METHODS

The study was developed in a shipyard located in the Colombian Caribbean characterized by highly corrosive humid tropical weather. A two-level experimental design including four factors. Tests samples of repainted naval steel, repaired under standard conditions were subjected to a salt Spray Fog Test (SFT), following the ASTM D1654 standard and using the Taguchi L₈ orthogonal array (Bierwagen *et al.*, 2003; Yuin and Alan, 1997).

The 8 specimens of repaired naval steel, taken from a ship ongoing major repairs without changing its sample surfaces were used (Fig. 1). The specimen dimensions are: 7.5 cm length and 15 cm width. Surface preparation and painting were developed in similar environmental conditions at the shipyard. The UHPWJ machine can provide a maximum pressure of 379.2 MPa.

To assess the influence of Visible (VC) and Non-Visible (NV) contamination, temperature and relative humidity, the specimens were processed in two group, one at 7:00 a.m. and the other at 4:00 p.m.

The water pressure is of great importance in the UHPWJ given its influence in the surface contamination after the process is finished. To assess this influence two pressure levels 241.3 MPa because and 275.8 MPa. The lower pressure (i.e., 241.3 MPa) is the most frequent value used in the plant because the human operator suffers less from fatigue in this case. Figure 2 shows the cleaning of 1 specimen by UHPWJ.

After cleaning with the UHPWJ, washing the surface with fresh water at a low pressure (22 MPa) reduces the chloride concentration in surface. Implementing or not this second cleaning introduces an additional factor influencing the presence of contamination in the metal surface which impacts the coating behavior.

Two different two of the most used epoxy paints in shipbuilding industry (i.e., one with iron oxide and another with polyamide) were tested to assess their performance against corrosion after the UHPWJ treatment.

To define the best combination of factors that guarantee the best physicochemical performance against corrosion, a Taguchi $L_{\rm 8}$ orthogonal array is used (Table 1). This design allows to measure at two levels the effect of up to seven control factors including noise factors. The control factors are placed in the first four columns and 5 repetitions for each test sample allowed the variability caused by the presence of a noise factor to be minimized. Furthermore, two stages were implemented first, the levels of the control factors that minimize the variability were defined using the highest signal/noise

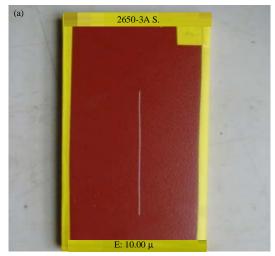




Fig. 1: Specimens readies for corrosion test; a) Coated with epoxy iron oxide and b) Coated with polyamide



Fig. 2: UHPWJ surface cleaning

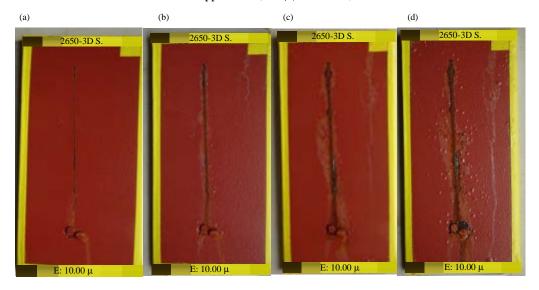


Fig. 3: Evolution of corrosion during the SFT test: a) 70 horas; b) 150 horas; c) 300 horas and d) 500 horas

Table 1: Factors and levels								
Factors	Description	Level 1	Level 2					
A	Coating type	Iron oxide	polyamide					
В	Pressure amplied	241.3 MPa	275.8 MPa					
C	Freshwater	"Yes"	"No"					
D	Surface preparation time	7:00 a.m.	4:00 p.m.					

Table 2	2: Lg orthogo	onal array for	r the experii	nental study		
<u>A</u>	В	С	D	e	e	e
1	1	1	1	1	1	1
1	1	1	2	2	2	2
1	2	2	1	1	2	2
1	2	2	2	2	1	1
2	1	2	1	2	1	2
2	1	2	2	1	2	1
2	2	1	1	2	2	1
2	2	1	2	1	11	2

ratio (which are achieved at the levels that maximize signal-to-noise); second, the control factor levels that maximize the mean value were defined. The best combination of factors result from the levels that minimize the variability and maximize the mean. The statistical processing was developed in the MINITAB 141 Software. Table 2 shows the " $L_{\rm s}$ " orthogonal array for the control factors and their levels. Columns 6-8 indicate the random error.

The response variables evaluate the progress of corrosion: the formation of blisters because of corrosion, the loss of adhesion along a cut line or any other issue of a particular epoxy paint resulting from the attack of corrosion by means of Salt Fog Test (SFT) during 50, 150, 300 and 500 h (Schmid, 2005) performed under ASTM D1654 standard. This tests assess the physicochemical behavior of steel with an atomizer spray a fog with a concentration of 5% NaCl by weight at 35°C over the steel specimens. Corrosion is evaluated at 10 levels (improving for 0-10) according the SFT test. Figure 3 shows the

evolution of the coating failure caused by the accelerated corrosion in one specimen painted with epoxy iron oxide.

RESULTS AND DISCUSSION

Table 3 shows the (L_8) orthogonal array of the statistical assessment of results from the SFT. In the assessment, the parameters with higher value have more influence as show in Fig. 4a where factor A (the type of paint) has the highest influence with level 1 (iron oxide-type paint) minimizing the variability. Factors B-D showed lower impacts on the variability. The combination of factors that minimize the variability is (A, B, C, D) = (1, 2, 1, 2).

The assessment of the factors affecting the mean, Fig. 4b shows the Factor A (the paint type) has the highest impact with Level 1 (iron oxide-type paint) maximizing the mean. Factors B-D have little influence on the mean. The combination of factors that maximizes the mean is (A, B, C and D) = (1, 2, 1 and 2).

From the experimental results, the best combination of factors for an optimum UHPWJ surface preparation is to start at 4.00 pm using a water pressure of 275.8 MPa in UHPWJ operation and washing afterwards with freshwater. Additionally, the physicochemical performance of naval steel exposed to corrosive environments improves with the use of iron oxide-based paints.

The results of variance analysis are shown in Table 4 and 5, mainly concluding that factor "A" has the highest effect on the response variable with a p<5% which confirms the results from Fig. 4.

The performance of anticorrosive coating applied to steels is strongly influenced by the surface preparation,

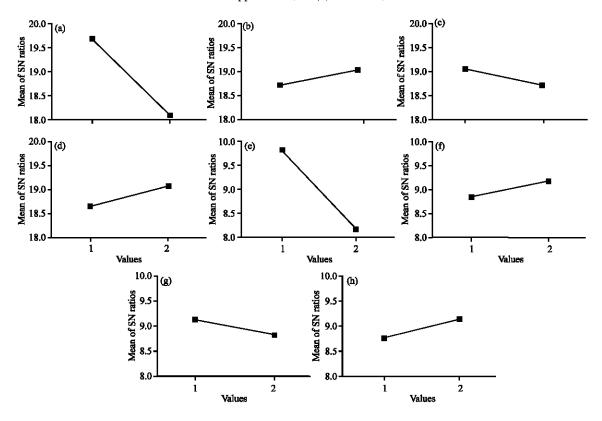


Fig. 4: Factors and interaction; a-d) S/N ratios and e-h) The median of the factors

Table	3: Statistic	al assessmer	nt of the res	ults from th	e SFT								
A	В	С	D	e	e	eSFC			μ	S/N			
1	1	1	1	1	1	1	10	10	10	10	10	10	20.00
1	1	1	2	2	2	2	10	10	10	10	10	10	20.00
1	2	2	1	1	2	2	10	10	10	10	10	10	20.00
1	2	2	2	2	1	1	10	10	10	10	10	10	20.00
2	1	2	1	2	1	2	10	7	7	7	7	7.6	17.36
2	1	2	2	1	2	1	10	8	8	7	7	8.0	17.84
2	2	1	1	2	2	1	10	8	7	7	7	7.8	17.60
2	2	1	2	1	1	2	10	9	9	9	9	9.2	19.25

Table 4: Analysis of the S/N ratio variance									
Source of									
variation	df	Seq SS	Adj. SS	Adj. MS	F-values	p-values			
A	1	7.8559	7.8559	7.8559	25.88	0.015			
В	1	0.3353	0.3353	0.3353	1.10	0.370			
C	1	0.3353	0.3353	0.3353	1.10	0.370			
D	1	0.5671	0.5671	0.5671	1.87	0.265			
Residual error	3	0.9106	0.9106	0.3035	-	-			
Total	7	10.0042	-	-	-	-			

optimal environmental conditions and a good primer. The
percentage of flash-rust concentration is lower for higher
water pressure in UHPWJ and the surface preparation
taking place in the afternoon with higher temperature and
lower moisture. On the other hand, the chloride
concentration slightly decreases (from a mean value of
4.3-3.2 $\mu Lg/cm^2$) when a freshwater washing at low
pressure is applied after the UHPWJ. Although, chloride
concentrations are acceptable for paint application

Source of						
variation	df	Seq. SS	Adj. SS	Adj MS	F-values	p-values
A	1	6.845	6.845	6.8450	31.35	0.011
В	1	0.245	0.245	0.2450	1.12	0.367
C	1	0.245	0.245	0.2450	1.12	0.367
D	1	0.405	0.405	0.4050	1.85	0.266
Residual error	3	0.655	0.655	0.2183	-	-
Total	7	8.395	-	-	-	-

according to NACE No 5/SSPC-SP 12. However, this reduction guarantees a lower absorption of humidity by osmosis through the coating and consequently, less superficial rust is produced.

Table 3 shows that epoxy-based coating with iron oxide has the best performance against corrosion. However, the epoxy paint with polyamide also showed good performance for the same combination of factors: water pressure of 275.8 MPa, subsequent

washing with fresh water and afternoon environmental conditions with highest temperature and lower humidity.

CONCLUSION

The best combination factors affecting the performance of the anticorrosion coating in naval steels to repair ships was defined. Conclusion points to the use of iron oxide based epoxy paints, highest water pressure and washing at low pressure afterwards. The humidity of the base material is not a troubling factor because evaporation is almost immediate under the work conditions.

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