

Bio-Degradation of Low Density Polyethylene (LDPE) by Fungi Isolated from Marine Water

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Abstract: Fungi isolated from sea water were subjected to growth in a medium containing LDPE as the sole carbon source with and without yeast extract. Increasing fresh weight of the fungi in the medium supplemented with LDPE after regular time intervals gave the evidence that the fungi are were utilizing LDPE as the carbon source. Further confirmation of LDPE utilization was carried out by the Sturm test where the degradation was attributed to the amount of carbon dioxide evolved during the growth period. The two fungi that showed good growth in medium supplemented with LDPE proved to degrade LDPE with higher efficiency that earlier reported results. Fungi were identified as *Aspergillus* sp., LDPE degradation is a severe environmental crisis in the world and we have proved that micro-organisms can be used for bio-remediation in this line.

Key words: Bio-degradation, LDPE, *Aspergillus*, marine fungi, Sturm test, South India

INTRODUCTION

The world plastic comes from the Greek word *plastikos* which means able to be molded in to different shapes. The plastics we use today are made from inorganic and organic raw materials such as carbon, silicon, hydrogen, nitrogen, oxygen and chloride. The basic materials used for making plastics are extracted from oil, coal and natural gas.

Polyethylene (PE) is a thermoplastic polymer consisting of long chains produced by combining the ingredient monomer ethylene. The ethylene actually converts to ethane as it takes its place in a polymer and straight sections of the polymers are the same structure as the simple chain hydrocarbons. The most important polyethylene grades are HDPE, LLDPE and LDPE. LDPE is defined by a density range of 0.910-0.940 g cm⁻³. It is not reactive at room temperature, except by strong oxidizing agents. It can withstand temperatures at 80°C continuously and 95°C for a short time. It is translucent or opaque variations, it is quite flexible and tough but breakable.

LDPE has more branching (on about 2% of the carbon atoms). Its intermolecular forces are weaker, its tensile strength is lower and its resilience is higher. Its molecules are less tightly packed and less crystalline because of its side branches, its density is lower. LDPE contains chemical elements carbon and hydrogen. It shows excellent resistance to dilute and concentrated

acids, alcohols, bases and esters, good resistance to aldehydes, ketones and vegetable oils, limited resistance to aliphatic and aromatic hydrocarbons, minerals oils and oxidizing agents and poor resistance and not recommended for use with halogenated hydrocarbons. Among synthetic plastics, the most problematic one in this regard is polyethylene. The resistance of polyethylene to biological attack is related to its hydrophobicity and water repellency polyethylene considered to be inert can be bio-degraded if the right microbial strain is isolated.

The bio-degradation of polyethylene is a slow process. Micro-organisms such as bacteria, fungi and actinomycetes are involved in the degradation of both natural and synthetic plastics (Gu *et al.*, 2000). The degradation of most synthetic plastics in nature is a very slow process that involves environmental factors followed by the action of wild micro-organisms (Albertsson, 1980).

Bio-degradation of polyethylene has been studied extensively earlier (Albertsson, 1980; Breslin, 1993; Breslin and Swanson, 1993; Imam and Gould, 1990) but the results were based on polyethylene blended with starch. Lee *et al.* (1991) have reported the bio-degradation of degradable plastic polyethylene by *Phanerochaete* and *Streptomyces* species. Foust *et al.* (1997) have reported the bio-degradation of LDPE/cellulose blends by common fungi. El-Shafei *et al.* (1998) have reported the bio-degradation of disposable

polyethylene by fungi and *Streptomyces* species. A major obstacle to bio-degradation of the polyethylene is the resistance of LDPE to biological attack because of its hydrophobicity, high molecular weight and its lack of functional groups recognised by microbial enzymatic systems (Hamid, 2000). Thermally treated LDPE was proved to be bio-degradable by *Penicillium pinophilum* and *Aspergillus niger*. Kathiresan (2003) has reported isolating fungi from the mangrove soil which has the potential to degrade polyethylene materials.

In most studies, fungi were considered for the degradation of LDPE due to their ability to form hydrophobic proteins that can attach to the polymer surface (Seneviratne *et al.*, 2006; Kershaw and Talbot, 1998), their generation of degrading enzymes that are well matched to the insoluble LDPE (Shah *et al.*, 2008; Shah, 2007), the faster growth of fungal biomass compared to bacteria (Kim and Rhee, 2003) and the growth extension and penetration into other locations through the distribution of hyphae. Also, fungi survive environments with low nutrient availability, low pH and low moisture well. Yamada-Onodera *et al.* (2001) isolated a strain of fungus *Penicillium simplicissimum* YK to bio-degrade polyethylene without additives. El-Shafei *et al.* (1998) investigated the ability of fungi and streptomyces strains to attack degradable polyethylene consisting of disposed polyethylene bags containing 6% starch.

Still a lot has to be done to isolate the right kind of microbial strain that could promote degradation of LDPE in a shorter period of time since all the previous reports showed activity only after a minimum period of 3-4 months.

Several analytical methods have been used to test bio-degradability which includes visual observation, changes in molar mass, weight loss measurement, CO₂ evolution, clear zone formation, etc.

Under aerobic conditions, microbes use oxygen to oxidize carbon and form carbon dioxide as one major metabolic end product. Consequently, the formation of carbon dioxide (Sturm test) is a good indicator for polymer degradation and are the most often used methods to measure bio-degradation in laboratory tests. This test has long been used to evaluate the degradability of diverse substances and chemicals in water (OECD guidelines) and has now been adapted to applications in non-water soluble polymeric materials like the trapping of the CO₂ gas in KOH solution and performing the gravimetric analysis with the help of Barium chloride solution as suggested by Muller *et al.* (1992).

MATERIALS AND METHODS

Sea water sample was collected from Kovalam coast-off the Bay of Bengal, 500 m away from shore at the depth of 5 m. Sample was inoculated in malt extracts soyapeptone broth. After observation of visible growth, the inoculum was transferred to soyapeptone medium having LPDE in the powdered form (0.50 g/100 mL).

Preparation of LDPE powder: LDPE sheets were cut into bits and immersed in xylene. It was boiled for 15 min as xylene dissolved the LDPE film and the residue was crushed while it was warm by using band gloves. The LDPE powder so obtained was washed with ethanol to remove residual xylene and allowed to evaporate to remove ethanol. The powder was dried in hot air oven at 60°C over night (Shah *et al.*, 2009).

Fungal colonizing studies: The colonizing capacity of the fungi on LDPE film was studied in big sized petri-plate. Briefly, Mineral Salt Medium (MSM) containing the following salts in 1 L distilled water: K₂HPO₄, 1 g; KH₂PO₄, 0.2 g; NaCl, 1 g; CaCl₂.2H₂O, 0.002 g; boric acid, 0.005 g; (NH₄)₂SO₄, 1 g; MgSO₄.7H₂O, 0.5 g; CuSO₄.5H₂O, 0.001 g; ZnSO₄.7H₂O, 0.001 g; MnSO₄.H₂O, 0.001 g and FeSO₄.7H₂O, 0.01 g was aseptically poured into petriplates. LDPE sheets were cut into small pieces 2×2 cm of similar weight, disinfected with 70% ethanol for 30 min and transferred to sterile distilled water for 20 min. Five LDPE sheets of same weight were placed in petriplates containing the minimal salt medium without yeast extract and the petriplates were inoculated with 5 similar sized colonies of fungi using the cork borer. The petriplates were incubated at room temperature and results were observed after 1 week to 10 days.

Quantification of CO₂-modified Sturm test: About 100 mL capacity autoclavable plastic containers were used for the study. The set up was arranged as shown in Fig. 1. Separate setup was maintained with un-inoculated MSM supplemented with LDPE powder. After the stipulated time (48 h) the KOH solution (1 M) that had trapped the CO₂ liberated by the inoculant as gravimetrically quantified using barium chloride. The dissolved carbon dioxide present in the medium was also estimated using Titration method. Briefly, sample (25 mL) was taken in a conical flask and 0.05 mL of 0.1 N Thiosulphate solution was added. After the addition of 2 drops of methyl orange indicator, this solution was titrated against 0.02 sodium hydroxide solution. End point was the change in color from orange red to yellow.

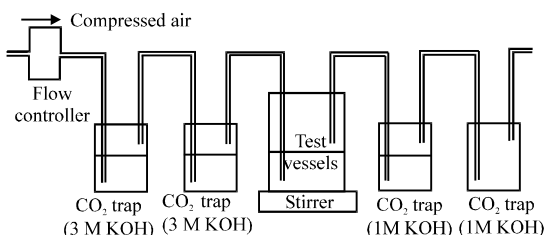


Fig. 1: Modified Sturm test

was added and titration continued till a pink color developed. Volumes of the titrant used were noted and the amount of CO₂ calculated using the equation:

$$\text{Amount of CO}_2 = \frac{A \times B \times 50 \times 1000}{V}$$

Where:

A = mL of NaOH titrant

B = Normality of NaOH

V = mL of the sample

Separate quantification was performed for test as well as control. Fungi that were used for the study were subjected to detailed macroscopic and microscopic analysis to prove their identity. This was done by growing the fungi in a slide culture (Muller *et al.*, 1992).

RESULTS AND DISCUSSION

Two fungal isolates SB and SD were observed to grow in the medium supplemented with paraffin wax and sub-sequently in the medium supplemented with LDPE powder. These isolates were further studied for colonization. From the colonization studies, there was an increase in the fresh weight of the fungal isolates observed after 7 and 17th day of the experiments (Table 1, Fig. 2 and 3).

There was considerable increase in the weight of the isolate SB whereas there was only a substantial increase in the weight of the isolate SD.

Table 2 shows the value for CO₂ evolution from the degradation of LDPE sample by the fungal isolates. The isolate SD was found to evolve about 4 g L⁻¹ of CO₂ in 1 week and isolate SB evolved around 3.8 g L L⁻¹ of CO₂. From the slide culture technique based on the microscopic morphology, we identified the isolate, SB as *Aspergillus versicolor* and SD as *Aspergillus* sp. (Fig. 4 and 5). The list of pollutants which pose environmental and health hazard and are tough for bio-degradation is a long one and includes solvents, wood preservative chemicals, pesticides, synthetic fibres, plastics, polyethylene, etc.



Fig. 2: LDPE film showing colonization of hyphae



Fig. 3: Microscopic view of stained fungal hyphae colonizing the LDPE film

Table 1: Colonization studies on LDPE films

Fungal isolate	7 days (g)	17 days (g)
SB	0.001	0.023
SD	0.014	0.018

Table 2: Quantification of carbon dioxide evolution after degradation

Fungal isolate	Amount of CO ₂ (1 week) g/L
SB	3.8913
SD	4.1594

The present study deals with the isolation of plastic degrading fungi from the marine water samples. Low density polyethylene strips were used for the study. Fungi that degraded paraffin wax were isolated first and their potential to grow in medium supplemented with LDPE as sole carbon source was assayed thereafter so as to study their bio-degradation potential of LDPE. According to Hiroyuki *et al.* (1978), it is anticipated that synthetic oligomer assimilating bacteria can also degrade corresponding polymers.

According to them, paraffin wax which contained most abundant species of C₂₈-C₃₂ can be regarded as lower homologues of polyethylene. Hadad *et al.* (2005)



Fig. 4: Microscopic morphology of isolate SB

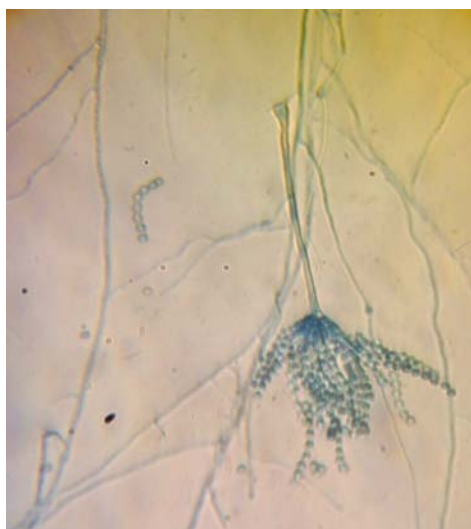


Fig. 5: Microscopic morphology of isolate SD

had isolated polyethylene degrading bacteria only from a grow up of bacteria isolates that had been initially screened for utilization of a mixture of liquid waxes. Colonization studies with the fungi yielded mediocre results with only marginal increase in the fresh weight of fungi after 7 and 17th day.

The CO₂ evolution test gave a valid data about the degradation rate. Isolate SB showed a good degradation rate of 77% followed by isolate SD that had a degradation rate of 83%, thus proving its efficiency as the bio-degrading agent.

Researchers obtained very good results from the CO₂ evolution test (Sturm test) when compared with the studies done by Ali *et al.* (2009) who reported a concentration of about 10 g/L of carbon dioxide after a period of 30 days. Shah *et al.* (2008) also reported carbon dioxide concentration of about 1.85 g/L after a 30 days period of growth of a fungal strain of *Fusarium* sp., on

LDPE films. Shah *et al.* (2009) also performed this test with a consortium of microbes consisting of both bacterial and fungal isolates but he reported a concentration of 1.85 g/L. All these reports do not carry the initial weight of the LDPE supplied in the medium nonetheless, the values obtained by us are far more than those that are reported. These findings corroborate the fact that even though colonization of the surface by microbe is an essential factor for metabolism of the substratum, it is not necessarily correlated with bio-degradation efficiency (Hadad *et al.*, 2005). LDPE can be bio-degradable if the right micro-organism is isolated.

CONCLUSION

Researchers have proved that the hydrophobic LDPE film can act as a substratum for some groups of micro-organisms which formed a biofilm on the LDPE film. The isolates also grew on minimal medium containing only LDPE in the powdered form as the carbon source even without any nitrogen source. We have also proved that the LDPE can be totally degraded into carbon dioxide which brings us closer to the fulfillment of the objective of isolating a micro-organism that can completely degrade the recalcitrant polyethylene if the right conditions are provided.

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