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Research Article Quality Indicator Hypoxanthine Compared with Other Volatile Amine Indicators of Sea Foods Stored in Refrigerator

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Abstract

Background and Objective: Study evaluates the suitable chemical method for the determination of the freshness of refrigerated seafood. Hypoxanthine content (Hx) of the refrigerated seafood was analysed and it was compared with the level of volatile amines production due to bacterial spoilage. Materials and Methods: Fresh seafood such as Stolephorus commersonii, Scomberomorus koreanus and Portunus sanguinolentus were collected from the fishing harbour of Tuticorin. The fishes were analyzed initially for TMA-N, TVB-N, hypoxanthine and pH content and it was stored at -4°C for 30 days. Sampling was done for every 10 days interval and it was checked for all the above quality indicators. Results: The results showed that volatile amines (TVB-N) and (TMA-N) were not found in stored seafood till the end of 10 days storage. Changes in pH and hypoxanthine content were noticed at the initial storage period itself. Refrigeration of sea foods are common at homes and at this condition even the superior guality stored fish has hypoxanthine formation and it gradually increase during the storage period. Hx assay reflects ATP break down and ATP-related compounds and the degradation of ATP to Hx has been attributed to muscle endogenous autolytic enzymes. The enzymatic reaction occurs in the initial stage of storage at low temperature but the contents of trimethylamine-nitrogen and total volatile base-nitrogen increase progressively during the later period of spoilage. Thus, hypoxanthine is regarded as the major catabolite of adenosine triphosphate while it was formed in refrigerated situation and it is a valuable freshness marker in stored fish. Conclusion: Thus, the study reveals that Hx measurements have some advantage over TMA-N and TVB-N analyses and Hx could be used as a superior spoilage indicator in the refrigerated sea foods. This study proves that the Hx could be used as a spoilage indicator and the sea foods stored at -4°C, should be consumed within 10 days of storage period to avoid Hx accumulation.

Key words: Hypoxanthine, trimethylamine-nitrogen, total volatile base-nitrogen, refrigerated sea foods, storage period, quality assessment

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INTRODUCTION

Frozen storage is an important method for preservation of food articles and it is commonly used in meat, fish and other animal protein-based industries, because under frozen storage the quality of seafood is retained for a long period and also this method has several advantages such as insignificant alterations in the product dimensions and minimum deterioration in products color, flavor and texture¹. However, the loss of quality by seafood is unavoidable when stored in frozen state². The changes in fish muscle fibers, proteins, lipids and textural properties during frozen storage have been studied for several decades because of their economic importance³. Aberoumand and Jooyandeh⁴ found that a proper storage temperature depending on the freezing properties of different species should be maintained because of its impact on the quality of fish. Changes in quality and freshness of the fish flesh depend on the length of storage period and temperature. People prefer the various frozen methods usually to maintain the quality of seafood. On a commercial scale, freezing of fish is done at -50 to -60°C. Since the entire water content in the fish is frozen in this method, the seafood can be stored for a long period and the stored fish almost remains as good as fresh fish. In deep or quick freezing methods the fish is frozen at -20°C and it retains its physical properties and nutritive values. Freeze drying is a modified deep freezing method which completely eliminates all chances of denaturation. Apart from this, there are other methods, employed by the small scale industries and households, such as chilling and frozen storage methods using refrigerators (-4°C).

Seafood spoilage occurs rapidly after harvesting. The spoilage process begins within 12 h of their catch in high ambient temperatures of the tropics⁵. The rate of fish spoilage depends on several factors such as the species of the seafood; their fat content, size and shape; the season of catch and the nature of the fishing grounds^{6,7}. Most of the fish species degrade as a result of digestive enzymes and lipases, microbial spoilage from surface bacteria and oxidation⁸. During fish spoilage, there is a breakdown of various components and also formation of new compounds. These new compounds are responsible for the changes in odor, flavor and texture of the fish meat. Bacterial numbers and levels of decomposition products have been used as indicators of post-mortem quality of fish. These changes have been assessed using nucleotides, ATP and related compounds, biogenic amines and volatile amines (TVB-N and TMA-N) as indicators⁹.

Among the above-mentioned indicators, TMA-N and TVB-N are considered as the real quality indicators. However,

the measurement of nucleotides such as hypoxanthine (Hx) used in fish quality assessments provides several distinct advantages over other chemical tests such as trimethylamine (TMA), dimethylamine (DMA) and total volatile bases (TVB) and others for these tests basically measure only the bacterial spoilage of sea foods. But the accumulation of Hx in fish tissue determines the initial phases of autolytic enzymatic deterioration as well as later contributions caused through bacterial spoilage at storage. Hx concentration is not affected by heat processing, cold storage and irradiation and is very useful in the case of certain seafoods containing little or no trimethylamine oxide (TMAO), which renders the TMA analysis useless¹⁰.

After the death of fish, oxygen supply to the tissues ceases due to disruption of the circulatory system. Soon after rigor mortis, the mitochondrial system ceases to function. Adenosine triphosphate (ATP) is gradually depleted through the action of various ATPase enzymes. After the residual supplies of creatine phosphate are depleted, anaerobic glycolysis continues to regenerate some ATP with the accumulation of the end product, lactate¹¹. ATP is degraded at the post-mortem stage by endogenous enzymes in the fish flesh: ATP-ADP-AMP-IMP-HxR-Hx. Hypoxanthine could be broken down by xanthine oxidase. The breakdown of ATP of fish muscle during the death struggle or subsequent to it liberates inosine 5'-monophosphate (IMP). The IMP is transformed into hypoxanthine during storage-in a progressive deprivation by dephosphorylation and the action of ribonuclease enzymes. The concentration of free hypoxanthine increases in the stored muscle of most species. IMP contributes to the pleasant flavor of fresh fish. Its degradation to hypoxanthine is a factor in the progressive loss of desirable flavor and in the development of the bitter 'off' flavor of sea foods.

Hypoxanthine is one of the products of nucleotide degradation mediated by bacterial activity and is the cause of the bitter, off-flavors of sea foods¹². Studies have shown that the progress in the degradation of nucleotides varies greatly from one fish to another but is often in direct proportion to the period of preservation¹³. Any kind of quality control parameter should accumulate or disappear rapidly during spoilage. In addition, it should be absent or present in variable amounts in fresh fish. Hypoxanthine is regarded as the major catabolite of adenosine triphosphate (ATP) and it is a useful indicator of freshness because of its gradual accumulation in seafood. Hx value rapidly increases in ice storage and the content of the muscle is used for the determination of the post-mortem age in seafood. Different preservation methods such as drying, smoking, freezing, chilling, brining, fermentation and canning have been reported to extend the shelf-life of seafood. However, low-temperature storage and chemical techniques are the most common methods in the industry today for controlling water activity, enzymatic, oxidative and microbial spoilage¹⁴. Nowadays people store fish for long in refrigerating frozen condition and they are unaware of the effect of nucleotide degradation products on stored fish. Volatile amines in stored seafood samples are the preferred quality indicators, but very few studies are on Hx as an authentic quality indicator. The aim of the present study is to observe the content of hypoxanthine in seafood stored in refrigerators and compare it with the production of volatile amine at (-4°C) and to determine which method is suitable for monitoring the quality of frozen stored seafood.

MATERIALS AND METHODS

Sample collection: Fresh seafoods such as *Stolephorus commersonii, Scomberomorus koreanus* and *Portunus sanguinolentus* (crab) (Fig. 1a-c) were collected from the fishing harbour of Tuticorin during December, 2016 and transported to the laboratory in an ice bag. They were immediately washed, degutted, cleaned and stored in the freezer at -4°C (Fig. 2a-c). Hypoxanthine content of the seafoods was analyzed initially and in frozen samples at 10 days interval for 1 month.

Processing of samples: Initially the stored samples were taken out from the refrigerator (Whirlpool of India Ltd, Gurugram, India) after 24 h for the purpose of analysis. Five individual samples from each seafood were randomly taken from the frozen blocks for the analysis. Before the analysis, the samples were defrosted, filleted and cut into pieces of 5 mm.

TVB-N and TMA-N: Levels of TVB-N and TMA-N were determined according to the procedure of Siang and Kim¹⁵ by using Conway micro diffusion unit (Bel-Art[™] F409410000, India). The extracts were prepared by mixing 2 g of the sample with 4% TCA in a 50 mL beaker and were homogenized properly. It was left for 30 min at room temperature with occasional grinding and was filtered and the filtrate was labeled and stored. Three thoroughly cleaned Conway units were taken and the edge of the outer ring of each unit was sealed using a sealing agent (Vaseline). Using a micropipette (Labline, India), 1 mL of boric acid solution was added to the inner ring of each unit. Into the outer ring of each unit, 1 mL of the sample extract was added. One milliliter of saturated K₂CO₃ solution was carefully pipetted into the outer ring of each unit and closed with a clip. The solutions in the units were then mixed gently, to prevent any solution mixing from one ring to the other. The units were placed in an incubator at 37°C for 60 min. Then the covers of the units were removed and the inner ring solution (a green color) was titrated with 0.02 N HCl



Fig. 1(a-c): Seafood samples, (a) Stolephorus commersonii, (b) Scomberomorus koreanus and (c) Portunus sanguinolentus



Fig. 2(a-c): Seafood samples in stored condition, (a) Stored *Stolephorus commersonii*, (b) Stored *Scomberomorus koreanus* and (c) Stored *Portunus sanguinolentus*

using a burette until the green color solution turned to pink. An average titrate volume of HCl was found from the results of three titrations for each sample. For each sample, the TVB-N values were calculated. A blank test was also carried out using 1 mL of 1% TCA, instead of sample extract. Trimethylamine in the sample was also determined by the Conway technique, which is similar to TVB-N determination except that prior to the addition of potassium carbonate, 1 mL of 10% neutralized formalin was pipetted into the extract with ammonia and this allows only the TMA-N to diffuse over the unit.

pH: Samples were prepared according to Woyewoda *et al.*¹⁶ using a tissue homogenizer. The pH level was monitored using a digital pH-meter (HANNA pH 123 microprocessor pH meter) in triplicate.

Hypoxanthine determination: Hypoxanthine content of the sea foods was determined according to the method of Luong *et al.*¹⁷. All the chemicals used were of analytical grade and procured from Sigma-Aldrich, USA, Reagents used for the Hx estimation:

- Perchloric acid (6%)
- Xanthine oxidase was prepared by 10 mg mL⁻¹ xanthine oxidase in a ratio of 1:5 U with 0.05 M phosphate buffer. Buffer was added very gradually to avoid loss of enzyme activity. Dilution was done immediately before use; diluted enzymes may be stored at a frozen temperature up to 6 months
- Phosphate buffer 0.05 M was prepared by dissolving 17.01 g potassium dihydrogen orthophosphate in about 250 mL distilled water. The pH was adjusted to 7.6 with 1 M NaOH and diluted to 500 mL with distilled water. A further five-fold dilution with distilled water gave a final concentration of 0.05 M
- Potassium hydroxide/phosphate buffer (pH 7.6) was prepared by dissolving 27.22 g potassium dihydrogen orthophosphate in about 250 mL water and 171 mL of 1 M sodium hydroxide was added. The pH was adjusted with orthophosphoric acid or sodium hydroxide. Then 557 mL of 1 M KOH was added and made up to 1 L with water
- Hypoxanthine standards were prepared in a 100 mL volumetric flask by dissolving 5.0 mg (0.005 g) of

hypoxanthine in 100 mL distilled water. Overnight stirring or agitation in an ultrasonic bath may be necessary to complete dissolution

Procedure: Fifty gram of seafood samples were blended with 200 mL of perchloric acid to make protein extract which is used for hypoxanthine content determination. Ten milliliter of the perchloric acid extract was neutralized with 10 mL of KOH-buffer solution. The neutralized extracts are not stable for long periods of time under refrigeration, so prompt analysis is essential. However, neutralized extracts may be frozen at -30°C. Determination of hypoxanthine content in samples was done by preparing the following set of test tubes for each sample extracts: Tube A: 1 mL neutralized extract+2 mL buffer +2 mL water. Tube B: 1 mL neutralized extract+2 mL buffer+ 1.5 mL water+0.5 mL enzyme. If very high or low hypoxanthine concentrations are expected, the volume of extract and the volume of water are altered, so that the total volume remains at 5 mL. The samples were incubated in a water bath at 37°C for 30 min and the absorbance of the samples was measured at 290 nm.

The increase of absorbance, Abs, for each sample was calculated as follows:

$$Abs = B + A8 - A7 - A$$

Where:

- A = Absorbance from sample tube A
- A8 = Absorbance from tube 8 used in standard curve preparation
- A7 = Absorbance from tube 7 used in standard curve preparation
- B = Absorbance from sample tube B

$$Hx = H \times \frac{[V1+(0.01 \times M \times W)]}{4} \times W \times \frac{V2+V3}{V3} \times \frac{1}{G}$$

Table 1: TMA-N, TVB-N contents and pH in refrigerated sea foods

Where:

- H = Hx from standard curve
- M = Moisture content of fish expressed in percentage
- V1 = Volume (mL) of perchloric acid used in 1:4 extraction
- V2 = Volume (mL) of KOH/phosphate buffer used for neutralization
- V3 = Volume (mL) of extract neutralized by KOH/phosphate buffer
- V4 = Volume (mL) of sample extract added to test tube
- W = Weight (g) of sample used in 1:4 extraction
- G = Gram molecular weight of Hx, (136.1)

Statistical analysis: Values were expressed as mean±standard deviations. All analysis were carried out in triplicates.

RESULTS AND DISCUSSION

Post-mortem deteriorative changes in seafood samples were formerly assessed by evaluating their sensory quality, microbial population and chemical changes. Spoilage is the result of a whole series of complicated deteriorative changes brought about in dead fish tissue by their own enzymes, by bacteria and by chemical action¹⁸. The early reaction of spoilage is autolytic and bacterial enzymes become progressively more active during the later stages¹⁹.

The concentration of volatile amines and the pH values of the samples are shown in Table 1. TMA-N, TVB-N and pH increases during frozen storage, causing loss of freshness and spoilage acceleration as a result of enzymatic and microbiological activity. The initial quality of TMA-N and TVB-N depends on many factors, including fish handling before freezing and it is a spoilage index indicating the beginning of the spoilage process. Thus the limits for some economically important species are already known. In most frozen stored fish the allowable maximum quantity of TMA-N

Name of the sea foods	Days of storage				
	Initial	10	20	30	
TMA-N content (mgN/100 g)					
Stolephorus commersonnii	nil	nil	1.80±0.10	9.65±0.17	
Portunus sanguinolentus	nil	nil	0.09±0.11	5.70±0.3	
Scomberomorus koreanus	nil	0.05±0.01	1.37±0.60	7.20±0.03	
TVB-N content (mgN/100 g)					
Stolephorus commersonnii	nil	nil	6.54±0.13	22.05±0.3	
Portunus sanguinolentus	nil	nil	16.50±0.40	27.28±1.53	
Scomberomorus koreanus	nil	2.42±0.05	12.11±2.10	19.63±1.22	
pH					
Stolephorus commersonnii	6.51±1.55	6.53±1.11	6.58±0.05	6.97±2.8	
Portunus sanguinolentus	6.53±2.3	6.56±8.2	6.59±1.90	6.70±2.22	
Scomberomorus koreanus	6.85±0.5	6.87±0.9	6.93±2.10	7.05±0.6	

*Values were expressed as Mean±standard deviations

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Name of the sea foods	Hypoxanthine content (mg/100 g)					
		Days of storage				
	Initial	10	20	30		
Stolephorus commersonnii	1.45±0.09	3.38±0.01	3.94±2.1	7.05±0.05		
Portunus sanguinolentus	0.98±1.12	5.11±0.17	6.19±0.11	7.81±0.09		
Scomberomorus koreanus	2.33±0.04	3.05±0.1	5.72±1.23	6.47±0.15		

* Values were expressed as mean±standard deviations

is 10-15 mgN/100 g and the TVB-N acceptability limits are 30-35 mgN/100 g wet weight²⁰. TMA-N concentration in fresh fish is low and it increases progressively in later stages of spoilage and therefore, not a suitable indicator of quality of fish stored in ice for less than 10 days²¹. But most of the researchers use the volatile amines production in stored fishes as a primary quality indicator. TVB-N values of fish species increased during the storage period but it was observed only after 10 days. In the present study, initially the TMA-N and TVB-N content of stored frozen seafood was below the detectable level and after 10 days storage it gradually increased to 0.05 mg N/00 g (TMA-N) in Scomberomorus koreanus. A TVB-N content of 2.42 mg N/100 g was found in the frozen samples of Scomberomorus koreanus after 10 days. Amplified levels of both TMA-N and TVB-N formation were observed in stored seafood after 20 days. The present study agrees with the results of Simat et al.²², who stated that volatile amines production occurred in fishes stored for long during the later stages of their storage. Kamal et al.²³ reported that TVB-N increased from 5.60-27.20 mg/100 g in frozen Hilsa fish at -20°C throughout 75 days and it was detected after 15 days. Some other researchers found that in pike fish TVB-N value was 4.19 at the end of 17 days but increased to 14.90 at the end of 7 months storage at -18°C²⁴. This lower increase in volatile amines might have resulted from lower storage temperature. Amegovu et al.²⁵ reported that freezing inhibits bacterial activity and so is expected to inhibit TMA accumulation too. Orak and Kayisoglu²⁶ observed that little change occurred at -5°C and none at -12 and -20°C in frozen hake (Merluccius merluccius). El-Deen and El-Shamery²⁷ found that TMA value of banana shrimp (Penaeus merguiensis) did not change during the 6 months storage.

In the present study, an increase in pH was observed during the storage period of the seafood samples. These results are in accordance with Erkan and Ozden²⁸, who stated that the increase was due to an increase in volatile bases from the decomposition of nitrogenous compounds by endogenous or microbial enzymes. Obemeata et al.19 observed that the increase in pH was higher in stored sample of Tilapia at -4°C, indicating that the biochemical and microbial changes are occurring faster in the stored fish at -4°C. Pawar²⁹ reported a slightly increased pH in Catla catla from 6.50-6.79 when stored at chilled temperature of -2 to -4°C. The change in pH of fish muscle is usually a good index for quality assessment. The increase in pH is caused by the enzymatic degradation of fish muscle. The highest pH value was obtained in Scomberomorus koreanus. Lakshmanan et al.³⁰ reported that there was a decline in pH after 36 weeks in whole, gutted and fillet frozen rock cod, *Epinephelus* species at -20°C. In catfish the muscle pH values after frozen storage at various time intervals were not significantly different from fresh muscle pH³¹. In the present study, a slight increase was observed in seafood throughout the 30 days of storage.

Shortly after the harvest, chemical and biological changes take place in dead fish due to enzymatic breakdown of major fish molecules³². Ozyurt et al.³³ stated that textural quality was reduced by the action of autolytic enzymes during the early stages of deterioration but it did not produce the characteristic spoilage off-odors and off-flavors. This indicates that autolytic degradation can limit shelf-life and product quality even with relatively low levels of spoilage organisms³⁴. Most of the impact is on the textural quality in addition to the production of hypoxanthine and formaldehyde. The digestive enzymes cause extensive autolysis which results in meat softening and rupture of the belly wall. It also results in the draining out of the blood water which contains both protein and oil³⁵.

The freshness index was determined on the basis of the results of the content in muscle tissue of ATP as well as the products of its breakdown viz. ADP, AMP, IMP, inosine and hypoxanthine³⁶. This index is a good indicator of meat freshness although it is specific to particular species³⁷. During the initial stage of storage hypoxanthine (Hx) begins to accumulate; volatile amines are absent initially but they can be increased during later stage of the storage period. The hypoxanthine content of the collected seafood samples initially and during the storage is presented in Table 2. In the present study initially Stolephorus commersonnii fish sample showed the Hx level of 1.45 mg/100 g and it increased to

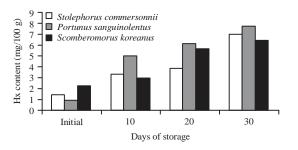


Fig. 3: Hypoxanthine content during the storage

7.05 mg/100 g during 30 days of storage at -4°C. In the case of *Scomberomorus koreanus* sample Hx content was 2.33 and it increased to 6.47 mg/100 g and in *Portunus sanguinolentus*, Hx was 0.98 in fresh crab and it increased to 7.581 mg/100 g during 30 days of storage (Fig. 3). An increase in ATP levels triggers the onset of rigor mortis and the accompanying nucleotide degradation.

Initially, the Hx concentration was noticed during the storage period but the TMA-N and TVB-N were absent in all the samples. Hx concentration increases with progressive spoilage of the sea foods. Ozogul and Balikci³⁸ reported that the acceptable limit of Hx in sea food was 4 mg/100 g whereas the unacceptable limit was 6-9 mg/100 g for consumption. However actual levels of Hx have been found to be dependent on genus, sampling procedures, seasonal differences and environmental factors³⁹.

Barnett et al.40 reported IMP concentration in very fresh coho salmon less than 24 h of post-harvest to be greater than 7.0 mg/100 g. However, there was a decrease of IMP content in the coho salmon during storage, followed by a corresponding increase in hypoxanthine. After 9 days of storage, the changes in nucleotides appeared to affect the organoleptic attributes of the iced salmon in such a way as to cause loss of normal cooked odor and flavor characteristics due to increases of Hx. Sofyan⁴¹ reported that increases in Hx value were noticed in Kembung fish (Rastrelliger neglectus) stored at 2-5°C for 12 days. Ozogul et al.42 reported that the concentration of Hx increased gradually from the initial storage time for herring kept under modified atmospheric packaging (MAP) in iced storage conditions. In the present study, Hx increases almost linearly with storage time, indicating higher freshness index for frozen sea foods.

In the present study, Hx content was at the acceptable limit in all the three sea foods up to 20 in days storage in the refrigerator but after 20 days storage, all were found to be at the unacceptable limit. Greene and Bernatt-Byrne⁴³ studied Hx values in Pacific cod and pollock during storage. Accumulation of Hx in Pacific cod (*Gadus macrocephalus*) fillets was slower than that in Atlantic cod (*Gadus morhua*), but similar to that in North Sea cod (*Gadus callarias*). The variation in initial nucleotide contents is associated with differences among species, seasons, catching gear, stress during fish death, water temperature and the time lapse between catch, slaughter and storage⁴⁴. Massa *et al.*⁴⁵ reported that at the beginning the Hx concentration increased during storage until the values leveled off at 4.7-5.0 M g⁻¹. The results of the present study are in concordance with those of the above-mentioned study.

Hx concentrations in hybrid striped bass irradiated with different radiation doses of 2.0, 3.0 and 20 kGy were similar to those of non-irradiated samples⁴⁶. Brown *et al.*⁴⁷ stored rockfish and silver salmon under modified atmosphere and the level of Hypoxanthine was very erratic in modified atmosphere-stored samples and hence determined Hx cannot be eliminated at various types of storage. Since the Hx content in fish depends on species and quality of storage, it can be used as a freshness indicator of stored fishes⁴⁸. Irradiation and MAP are good storage methods for seafoods, but Hx was not reduced using the above-mentioned methods and based on the present study it was concluded that Hx was not eliminated in refrigerated (-4°C) sea foods also.

The increase in Hx production with time is related to increase in autolytic enzymes⁴⁹. FDA⁵⁰ indicated that Hx levels in fish relate to the length of storage and also to the amount of autolytic enzymes in the flesh. During the comparative storage study on specimens of cold-smoked salmon with reduced autolytic enzymes (with short storage) and those with normal autolytic enzymes (with long storage), it was found that 68% of the Hx in normal samples with long storage samples was due to enzymatic activity. The present study suggests that a month in the refrigerator is a prolonged storage period which increases the production of Hx and this increased Hx production, due to the nucleotide degradation of the sea foods, is responsible for the enzymatic worsening. Amos⁵¹ reported that the first step in the degradation of adenosine triphosphate to Hx is governed by autolytic processes and the next step partly by bacterial enzymes. Duun and Rustad⁵² reported that hypoxanthine can be formed by the autolytic decomposition of nucleotides and that it can be formed in fresh sterile fish muscles and the physical storage condition was one of the dependable factors for this. These inferences agree with the results of the present study.

Abbas *et al.*⁵³ reported that in most of the cases the enzymes involved in the breakdown of adenosine triphosphate (ATP) to inosine monophosphate (IMP) are believed to be autolytic whereas the conversion of IMP to inosine (Ino) and then to hypoxanthine (Rx) is believed to be mainly due to spoilage bacteria although Hx has been shown to accumulate slowly in sterile fish tissue. The factors which

have been shown to affect the nucleotide breakdown pattern include species, the temperature of storage and physical disruption of tissues. In addition, since nucleotide breakdown reflects the combined action of autolytic enzymes and bacterial action, the types of spoilage bacteria would no doubt affect the nucleotide patterns⁵⁴. The present study correlates with the above findings.

Generally refrigeration storage freezes only 75% of the water present in the samples. This does not include the water which is chemically bound to specific sites such as carbonyl and amino groups of proteins and hydrogen bonding⁵⁵. Ocano-Higuera *et al.*⁵⁶ reported on the variation of the proportion of water which is converted to ice in the muscle tissue of fish against temperature. The final quality depends on the condition of the fish at the time of freezing as well as on other factors including freezing/cold storage temperature, freezing rate and distribution. Therefore, based on the above reports the present study suggests that the bound water containing protein and lipid under enzymatic degradation (proteolysis and lipolysis) leads to the production of Hx in partially frozen sea foods.

The most crucial factor which affects the quality of refrigerated sea foods in the present study is slow freezing. Slow freezing, in comparison to fast freezing, results in the formation of large ice crystals, which damages the walls of the cells and causes denaturation of the protein. On the other hand, denaturation also depends on the concentration of enzymes and other compounds present in them⁵⁷. The enzymatic reactions can still continue in refrigerated fish from the beginning of the storage at a temperature of $-4^{\circ}C^{58}$ which involve a number of other metabolic activities such as glycolysis, nucleotide degradation and proteolysis⁵⁹. These endogenous enzymatic activities cause intrinsic chemical and physical changes. Although enzymatic activities are slow, they can still support microbial growth and metabolism⁶⁰.

Shortly after capture, the dead fish undergoes chemical and biological changes due to enzymatic breakdown of major molecules⁶¹. Getu *et al.*⁶² stated that autolytic enzymes reduce textural quality during early stages of deterioration but did not produce the off-odors and off-flavors characteristic of spoilage. So researchers are focusing more attention on the volatile amines and bacterial spoilage produced at the later stages rather than on the initial end product (Hx). This indicates that autolytic degradation can limit shelf-life and product quality even with relatively low levels of spoilage organisms. Tys and Pieters⁶³ reported that the autolytic changes occur in chilled/frozen stages. Most of the impact is on textural quality along with the production of hypoxanthine and formaldehyde. The digestive enzymes cause extensive autolysis which results in meat softening, rupture of the belly wall and draining out of the blood water which contains both protein and oil⁶⁴.

Under improper or inadequate storage conditions proteolysis is responsible for degradation of proteins and is followed by a process of solubilization⁶⁵. At the same time, peptides and free amino acids can also be produced as a result of autolysis of muscle proteins, which lead towards the spoilage of fish meat as an outcome of microbial growth at later stage and production of hypoxanthine at the initial stage of storage⁶⁶. Aoki and Ueno⁶⁷ reported that postmortem degradation of mackerel muscle tissue was due to the activity of Cathepsin L. enzyme liberated from lysosomes during aging for 7 days at -4°C. The enzymatic hydrolysis of fats by lipases is termed lipolysis (fat deterioration). During this process, lipases split the glycerides forming Hx initially and later it is converted to free fatty acid which is responsible for the common off flavor, frequently referred to as rancidity and reducing the oil quality⁶⁸. The present study clearly indicates that the presence of hypoxanthine could be used as a quality indicator instead of the analysis of free fatty acids, other volatile amines and the microbial spoilage, for Hx is produced at the initial stages of storage and our results correlate with the previous studies made by other researchers.

Arannilewa *et al.*⁶⁹ investigated the effect of frozen storage on the chemical, microbiological and sensory profile of tilapia fish (*Sarotherodun galiaenus*). They reported decrease in the values of protein and fat and increase in nucleotide degradation products. Miladi *et al.*⁷⁰ reported that *Salmonella typhimurium* and *Anasakis simplex* (nematode) can be killed by freezing of fish but the nucleotide breakdown was not reduced in refrigerated storage.

Al-Jasser and Al-Jasass⁷¹ stated that freezing and cold storage are efficient methods of fish preservation but they do not improve product quality. It is necessary to preserve the fish at 0°C after the catch as its spoilage is very rapid⁷². Humaid and Jamal⁷³, put forward two possibilities for storing fresh fish at low temperatures: cooling at -1 to +4°C, which inhibits the growth of microorganisms; and freezing at -18 to -30°C, which completely stops the growth of bacteria and nucleotide degradation. Seafood under prolonged refrigeration is highly prone to nucleotide breakdown and Hx formation.

CONCLUSION

The present study was carried out to determine Hx as the best spoilage indicator when compared with the other commonly used indicators (TMA-N and TVB-N).This study clearly shows that Hx is formed at the initial stage of storage period though the rate of formation varies from species to species. So, it is concluded that the nucleotide degradation product is increased in refrigerated product and can only be used as a temporary method of preservation. Therefore, for long time storage of seafood the household refrigerator is not recommended.

SIGNIFICANCE STATEMENT

This study discovers the content of hypoxanthine in seafood stored in refrigerators and compare it with the production of volatile amine at (-4°C) and to determine which method is suitable for monitoring the quality of frozen stored seafood and that can be beneficial for seafood consumers and exporters. This study will help the researcher to uncover the critical areas of best quality indicators on the post harvest processing of seafood's under storage that many researchers were not able to explore. Thus a new theory on quality indicators of stored food may be arrived at.

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