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# Induction and Uncoupling of Rat Liver Mitochondria by Oral Administered Coartemether

<sup>1</sup>Godswill N. Anyasor, <sup>1</sup>Oluwadamilola M. Odunaike and <sup>2</sup>Olufunso O. Olorunsogo

Corresponding Author: Godswill N. Anyasor, Department of Biochemistry, Ben Carson School of Medicine, College of Health and Medical Sciences, Babcock University, Ilisan-Remo, Ogun State, P.M.B. 21244, Ikeja, Nigeria Tel: +2347034618325

# ABSTRACT

This study investigated the cytotoxic effect of varying doses of antimalarial coartemether (2.0, 4.0, 8.0 and 10.0 mg kg<sup>-1</sup>) in the presence of ferrous sulphate (2800 mg kg<sup>-1</sup>) for 3 days on normal rat liver mitochondrial membrane permeability transition pore opening and  $F_1F_0$  ATPase activity. Swelling was estimated spectrophotometrically under succinate energized condition. Calcium ion treated mitochondria preloaded with coartemether induced swelling in a concentration dependent manner in vitro. Swelling was amplified in the presence of ferrous. Coartemether alone and combined coartemether-ferrous also induced mitochondrial swelling in the presence of spermine. In vivo study further showed that 10.0 mg kg<sup>-1</sup> coartemether induced swelling in the presence of ferrous. Coartemether stimulated an increased activity of mitochondrial  $F_1F_0$  ATPase in a concentration dependent manner. Thus, these findings indicate that coartemether at high dose in the presence of ferrous sulphate could be an inducer of mitochondrial mega pore opening and an uncoupler of oxidative phosphorylation initiating apoptosis.

**Key words:** Coartemether, mitochondria, F<sub>1</sub>F<sub>0</sub> ATPase, rat, liver

# INTRODUCTION

Coartemether or Coartem® (artemether-lumefantrine) is currently the most viable artemisinin combination therapy specifically indicated for the treatment of acute, uncomplicated malaria infections due to *Plasmodium falciparum* in patients of 5 kg b.wt. and above (WHO, 2006; Premji, 2009).

Coartemether is a fixed-dose combination tablet of 20 mg artemether and 120 mg lumefantrine in a ratio of 1:6 (Novartis Pharma AG., 2009). One of its components, artemether is a semisynthetic chiral acetal derivative of artemisinin that interferes with parasite transport proteins, disruption of mitochondrial function, inhibits angiogenesis and modulates host immune function. While, lumefantrine is a racemic mixture of a synthetic fluorine derivative formerly known as benflumetol and is structurally related to quinine, mefloquine and halofantrine. It interferes with the conversion of heme, the toxic intermediate produced during hemoglobin break-down to non-toxic hemozoin. The accumulation of heme and free radicals results in parasite death (Byakika-Kibwika et al., 2010).

<sup>&</sup>lt;sup>1</sup>Department of Biochemistry, Ben Carson School of Medicine, College of Health and Medical Sciences, Babcock University, Ilisan-Remo, Ogun State, P.M.B. 21244, Ikeja, Nigeria

<sup>&</sup>lt;sup>2</sup>Biomembrane and Biotechnology Laboratories, Department of Biochemistry, Faculty of Basic Medical Sciences, College of Medicine, University of Ibadan, Ibadan, Nigeria

Artemether, like other artemisinin-derived compounds, acts quickly to rapidly reduce the parasite burden, while lumefantrine serves as a longer-acting agent to eliminate remaining parasites. The combination is effective in parasite strains known to be resistant to traditional antimalarials such as chloroquine (Mwesigwa et al., 2010). Artemether is largely metabolized by cytochrome P450 (CYP) 3A4/5 but also by CYP2B6, CYP2C9 and CYP2C19. Metabolism through CYP3A4 produces an active metabolite, dihydroartemisinin (DHA) that contributes substantially to its antimalarial activity (Cousin et al., 2008). Lumefantrine is metabolized primarily by CYP3A4 and then undergoes glucuronidation (Hietala et al., 2010; Mwesigwa et al., 2010).

The highly reactive endoperoxide moiety in artemisinins is thought to be crucial for their mode of action but the exact mechanism remains controversial (Del Pilar Crespo *et al.*, 2008). Several models have been proposed including a Fenton-type reaction where artemisinins generate reactive oxygen species and carbon-centered radical molecules that modify proteins of *Plasmodium* parasites. Other studies suggest that artemisinin inhibit the Ca<sup>2+</sup>-dependent SERCA-like ATPase PfATP6 upon activation by Fe<sup>2+</sup> from hemoglobin. Another mechanism is the disruption of the mitochondrial membrane potential as suggested from data of yeast model (Li *et al.*, 2005).

Mitochondrial respiration depends on the flow of electrons through four oligomeric respiratory complexes that comprise the electron transport chain. The energy released by electron flow through the respiratory complexes is conserved in an electrochemical potential consisting of a proton gradient and membrane potential produced by the coupled translocation of protons through the inner mitochondrial membrane at Complexes I, II and IV. Energy stored in the electrochemical potential is coupled to ATP synthesis by translocation of protons into the mitochondrial matrix through complex V (ATP synthase) (Brand and Nicholis, 2011).

The list of described inducers of the Ca<sup>2+</sup>-dependent mitochondrial mega channel (Membrane Permeability Transition (MPT) pore, is long and includes many different chemical and physical factors all synergistic to Ca<sup>2+</sup>(Gunter and Pfeifer, 1990; Zoratti and Szabo, 1995). As a result of MPT pore opening intramitochondrial solutes of molecular mass lower than 1.5 kDa equilibrate with those in cytosol (Zoratti and Szabo, 1995). This is accompanied by the activation of mitochondrial respiration, the loss of ions accumulated in the matrix and high amplitude swelling of mitochondria. Recent findings suggest MPT involvement in either programmed mitochondrial destruction (Zorov *et al.*, 1992; Brand and Nicholis, 2011) and hence, in mitochondrial selection (Skulachev, 1996) or in programmed cell death (Zamzami *et al.*, 1995). The mitochondrial contribution to apoptosis opens a vast field for investigating new mitochondrial related diseases.

Several studies have tested coartemether and its components in a complete range of acute and subchronic animal toxicology studies, including reproductive toxicology, genotoxicity and juvenile animal studies (Raji et al., 2005; Efferth and Kaina, 2010; Onyesom and Agho, 2011). Mechanistic neurotoxicity studies were performed in both rats and dogs to evaluate functional and histopathologic changes (Oyemitan et al., 2007; Cousin et al., 2008; Ajibade et al., 2011). Despite these findings there is a paucity of large-scale clinical trials suitable to detect rare but significant toxicity especially when artemisinin combine therapy is prescribed alongside with ferrous in severe anemia conditions. Although, Mpiana et al. (2007) have reported that endoperoxide lactone based drugs could form complexes with heme and hemin. Therefore, attempts had been made to further investigate the effect of cytotoxic effect of coartemether on mitochondrial integrity.

#### MATERIALS AND METHODS

Chemicals: Coartemether tablets (Norvatis, Pharma AG, Switzerland), ferrous sulphate tablets (Pharmacy unit, University College of Health, Ibadan, Nigeria), mannitol, sucrose (BDH Chemicals

Ltd; Pools, England), HEPES (May and Baker Lab; USA), EGTA, bovine serum albumin (Sigma Chemical Co; USA), spermine (Research Biochemical, USA) and all other reagents used were of analytical grade.

Animal: Male albino Wistar rats (120-150 g) were obtained from Preclinical Animal House, Physiology Department, University of Ibadan, Ibadan, Nigeria. The animals were maintained in cages acclimatized for two weeks in accordance to good laboratory animal care practice at the departmental animal house. Tap water and commercial pelleted feed were provided under standard conditions of temperature 28±2°C and a 12 h light/dark cycle.

Experimental design: Oral coartemether tablets were dissolved in sunflower oil as vehicle and administered orally to the test groups using an oral dosing needle for 3 days in accordance to WHO recommendation for treatment of uncomplicated malaria. Animals were assigned randomly into six groups of three rats each. Group I: untreated normal rats; group II: (control) normal rats were given oral sunflower oil (1 cm³), group III-VI were coadministered coartemether (2.0, 4.0, 8.0 and 10.0 mg kg⁻¹ body weight) and ferrous sulphate (4.3 mg kg⁻¹ b.wt.) respectively. The animals were euthanized 24 h after fasting overnight by cervical dislocation at the end of treatment. Subsequently, the rat liver was excised, trimmed of excess tissue and subjected through the standard protocol for mitochondria isolation. Mitochondria isolated from Group I were preloaded with varying concentrations of coartemether (600, 1200, 2400 and 3000 μg mL⁻¹) and ferrous sulphate (1280 μg mL⁻¹) to assess the *in vitro* effects on mitochondria MPT.

**Mitochondrial isolation:** Rat liver mitochondria isolated by conventional differential centrifugation in a buffer containing 210 mM mannitol, 70 mM sucrose, 5 mM HEPES (pH 7.4) and 1 mM EGTA (Schneider and Hogeboom, 1950); EGTA was omitted in the final wash solution. Protein content was estimated by Folin-Ciocalteu method using Bovine Serum Albumin as standard (Lowry *et al.*, 1951).

Assessment of mitochondrial swelling: Mitochondrial swelling was assessed according to the method of Lapidus and Sokolove (1993). Changes in absorbance of mitochondria were monitored at 540 nm in a 6405 Jenway UV-visible spectrophotometer. Mitochondria (0.4 mg mL<sup>-1</sup>) were suspended in a medium containing 210 mM mannitol, 70 mM sucrose, 5 mM HEPES-KOH (pH 7.4), 0.8 μM rotenone and 5 mM succinate. Swelling was triggered by Ca<sup>2+</sup> while spermine serves as an inhibitor.

Assessment of mitochondrial  $F_1F_0$  ATPase activity: Mitochondria isolated from untreated normal rats in another set of experiment were preloaded with varying concentrations of 10.0, 20.0, 30.0, 40.0, 50.0 and 60.0 mg mL<sup>-1</sup> coartemether to assess the effects on mitochondrial  $F_1F_0$  ATPase activity. Mitochondrial adenosine triphosphatase or  $F_1F_0$  ATPase activity was determined by a modified method of Lardy and Wellman (1953). Each reaction vessel contained 65 mM tris-HCl (pH 7.4), 1 mM ATP and 25 mM sucrose. The reaction was started by the addition of mitochondrial fraction (0.4 mg mL<sup>-1</sup>) vortex for 30 min at 25°C. The reaction was stopped by the addition of 8 mL of 10% trichloroacetic acid to each test tube then centrifuged at speed of 3000 g. The deproteinized supernatant was kept for phosphate determination.

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Determination of inorganic phosphate: This was performed according to the method described by Fiske and Subbarow (1925) modified by Bababunmi and Bassir (1972). 0.4 mL of perchloric acid was added to 5.0 mL of the deproteinized supernatant in a test tube. This was followed by addition of 0.4 mL of 5% ammonium molybdate and 0.2 mL of a 0.2% freshly prepared solution of ascorbic acid. The tube was thoroughly mixed, gently shaken and allowed to stand for 20 min. A standard solution of potassium dihydrogen phosphate (0.2 mg Pi per 5 mL) was similarly treated. The intensity of the blue colour which developed was read at 680 nm using a spectrophotometer. Water blank was used to set the instrument at zero.

#### Calculation:

$$P_{i} \; (\text{mg mL}^{-i}) = \frac{\text{Reading of test} \times 0.02 \times 1}{\text{Reading of standard} \times 0.001}$$

$$P_{i} \ \ \text{released mole} = \frac{Pi \ mg \ \ mL^{-1} \times 1000}{Molecular \ weight \ of \ Pi \times 1}$$

Mole P; released mL<sup>-1</sup> mitochondrial protein is given by the expression:

$$\frac{\text{Mole } P_i \text{ released}}{\text{mg mitochondrial protein} \times 30}$$

#### RESULTS

Triggering agent (Ca²+) induced MPT or swelling in a succinate energized rat liver mitochondria under a normal respiration sucrose-phosphate buffer. However, mitochondrial membrane was intact without any observable swelling in the absence of a triggering agent (Fig. 1). *In vitro* study showed that mitochondria preloaded with 600, 1200, 2400, 3000 µg mL<sup>-1</sup> coartemether induced opening of the MPT pore in a concentration dependent manner with minimal and maximal swelling inductions at 600 and 3000 µg mL<sup>-1</sup> coartemether, respectively. Conversely, ferrous sulphate at 1280 µg mL<sup>-1</sup> did not induced mitochondria swelling (Fig. 1). However, coartemether preloaded mitochondria in the presence of ferrous showed a significantly large amplitude (p<0.05) of swelling in a concentration dependent manner (Fig. 2).

Spermine inhibited swelling induced at low concentration of coartemether (600  $\mu g$  mL<sup>-1</sup>) and ferrous while high concentration of coartemether (3000  $\mu g$  mL<sup>-1</sup>) and ferrous induced mitochondria swelling in the presence of spermine (Fig. 3).

In the animal study, co-administered ferrous sulphate  $(4.3~{\rm mg~kg^{-1}})$  and coartemether at  $2.0,~4.0,~8.0~{\rm mg~kg^{-1}}$  did not induce mitochondria MPT both in the absence  ${\rm Ca^{2+}}$ . However, there was significantly high (p<0.05) amplitude of liver mitochondria swelling in the animals co-administered with 10 mg kg<sup>-1</sup> coartemether and ferrous sulphate  $(4.3~{\rm mg~kg^{-1}})$  (Fig. 4).

The result also revealed that varying concentrations of coartemether (10, 30, 50 and  $60 \text{ mg mL}^{-1}$ ) elevated the mitochondrial  $F_1F_0$  ATPase activity in a concentration dependent manner with minimal (104.4 mol  $P_i$ /mg protein/ min) and maximal (146.4 mol  $P_i$ /mg protein/ min) activities at 10 and 60 mg mL<sup>-1</sup>, respectively (Table 1).

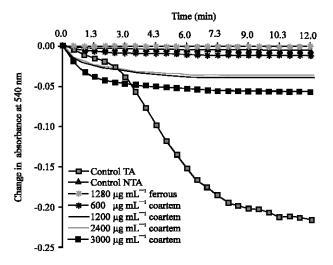


Fig. 1: Change in absorbance (540 nm) for 12 min by ferrous sulphate (Fe<sup>2+</sup>), varying concentrations of coartemether on mitochondrial membrane permeability transition pore in the energized by sodium succinate, TA: Triggering agent, NTA: Non triggering agent

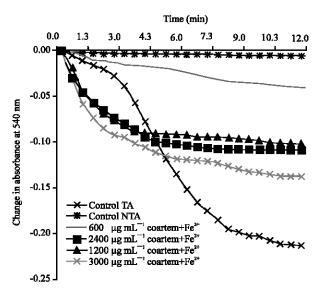


Fig. 2: Change in absorbance (540 nm) for 12 min by varying concentrations of coartemether in the presence of ferrous sulphate on mitochondrial membrane permeability transition pore energized by sodium succinate, TA: Triggering agent, NTA: Non triggering agent

Table 1: The Hydrolysis of ATP by mitochondrial  $F_1F_0$  ATP ase by varying concentrations of coartemether

Concentration (mg mL <sup>-1</sup> )	$mg \; P_i  released \; (mL^{-1})$	$Mole P_i released$	$Mole\ P_i\ released\ (mL^{-1})$
10	17.0	125.2	104.3
20	19.0	139.4	116.2
30	19.6	144.1	120.0
40	20.7	152.2	126.8
50	23.5	172.9	144.1
60	23.9	175.7	146.4

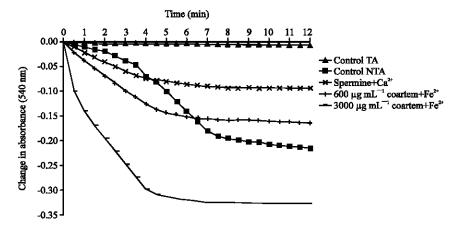


Fig. 3: Change in absorbance (540 nm) for 12 min by two concentrations of coartemether (lowest and highest) combined with ferrous sulphate on mitochondria permeability transition pore in the presence of spermine energized by sodium succinate, TA: Triggering agent, NTA: Non triggering agent

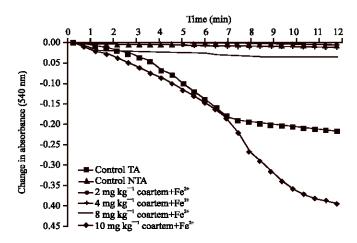


Fig. 4: Change in absorbance (540 nm) for 12 min *in vivo* effect of co-administered Fe<sup>2+</sup> (4.3 mg kg<sup>-1</sup>) and varying dose of coartemether on mitochondria permeability transition pore in the absence of a triggering agent energized by sodium succinate, TA: Triggering agent, NTA: Non triggering agent

#### DISCUSSION

In the present study, rat liver mitochondria preloaded with varying concentrations of coartemether in vitro induced mitochondrial Membrane Permeability Transition (MPT) pore in a concentration dependent manner. This supports the theory implicating mitochondrial MPT pore formation (swelling) as the possible mode of action for artemisinin-related compounds (Eckstein-Ludwig et al., 2003; Wang et al., 2010). Studies had shown that the parasiticidal activity of artemether resides on the peroxyl ring of dihydroxyartemisinin metabolite (Krishna et al., 2004; Efferth and Kaina, 2010). The cleavage of an endoperoxide bridge generates free radicals (Cumming et al., 1997). Free radicals have been associated with mitochondrial MPT pore opening (Del Pilar Crespo et al., 2008). Mitochondrial membrane permeabilization results in the release of

cytochrome c that trigger apoptosis by caspase-cascade pathways, consequently leading to cell death (Galluzi et al., 2009). Furthermore, the addition of varying concentrations of coartemether to the calcium ion preloaded mitochondria in the presence of ferrous sulphate further increased the swelling of mitochondria. Previous studies reported that iron bioactivates artemisinin into a free radical through an iron-mediated cleavage (Dhingra et al., 2000; Balint, 2001; Noori et al., 2004). This study also showed that coartemether in the presence of iron could serve as a cytotoxic agent against cancerous cells. Other studies had proved that the combined administration of dihydroxyartemisinin and iron retarded the growth rate of tumor (Moore et al., 1995).

The induction of mitochondrial MPT by coartemether at high concentrations in the presence of a natural inhibitor spermine *in vitro* suggests that the cytotoxic effect of coartemether could be at high concentrations.

Animal study showed that coartemether at the therapeutic dose range of 2-4 mg kg<sup>-1</sup> b.wt. did not induce mitochondrial swelling in the absence of a calcium ion. This seems to be in agreement with Norvatis Pharma claim on the clinical safety of the therapeutic dose range (Novartis Pharma AG., 2009). This observation might have resulted from the action of CYP3A4 isoenzyme metabolizing and degrading the total amount of drug that reaches the receptor/active site to elicit a pharmacological response (Cousin *et al.*, 2008; Mwesigwa *et al.*, 2010). However, swelling was induced at high dose of 10 mg kg<sup>-1</sup> b.wt. coartemether.

Further studies indicated that coartemether elevated the hydrolysis of ATP to ADP and inorganic phosphate by mitochondrial  $F_1F_0$  ATPase in a concentration dependent manner. This was determined spectrophotometrically by the increase in concentration of released inorganic phosphate  $P_i$ . This observation could also be accounted through the induction of mitochondrial MPT by coartemether *in vitro* which have already compromised the intactness of the mitochondria. The mitochondrial  $F_1F_0$  ATPase or ATP synthase is known to harnesses the proton gradient generated during the transfer of electron along the respiratory chain and couples it to the oxidative phosphorylation of ADP and inorganic phosphate to produce ATP required for diverse biochemical and cellular functions (Vinogradov, 2000; Nelson and Cox, 2008). The collapse in mitochondrial electrochemical gradient could result in the hydrolysis of ATP by  $F_1F_0$ -ATPase for the proton gradient recovery. This turns  $F_1F_0$ -ATPase into a consumer rather than being a producer of ATP in failing cells (Zablockaite *et al.*, 2007; Nelson and Cox, 2008).

#### CONCLUSION

This study indicates that coartemether at high dose in the presence of iron could have a profound cytotoxic effect on mitochondrial membrane permeability transition pore and it also could serve as an uncoupler of mitochondria respiration. However, additional studies are required to probe further into the effect of coartemether on cytochrome c release and mechanisms associated with apoptosis to gain more insight into the toxicological pathway.

# REFERENCES

Ajibade, A.J., P.B. Fakunle and P.D. Shallie, 2011. Some histological observations and microstructural changes in the Nissl substances in the cerebellar cortex of adult Wistar rats following artesunate administration. Curr. Res. Neurosci.

Bababunmi, E.A. and O. Bassir, 1972. Effects of aflatoxin B1 on the swelling and adenosine triphosphatase activities of mitochondria isolated from different tissues of the rat. Febs Lett., 26: 102-104.

- Balint, G.A., 2001. Artemisinin and its derivative: an important new class of antimalarial agents. Pharmacol Ther., 90: 261-265.
- Brand, M.D. and D.G. Nicholis, 2011. Assessing mitochondrial dysfunctions in cells. Biochem. J., 435: 297-312.
- Byakika-Kibwika, P., M. Lamorde, H. Mayanja-Kizza, C. Merry, B. Colebunders and J.P. van Geertruyden, 2010. Update on the efficacy, effectiveness and safety of artemether-lumefantrine combination therapy for treatment of uncomplicated malaria. Ther. Clin. Risk Manag., 6: 11-20.
- Cumming, J.N., Ploypradith, P. and G.H. Posner, 1997. Antimalarial activity of artemisinin (qinghaosu) and related trioxanes: Mechanism(s) of action. Adv. Pharmacol., 37: 253-297.
- Del Pilar Crespo, M., T.D. Avery, E. Hanssen, E. Fox and T.V. Robinson *et al.*, 2008. Artemisinin and a series of novel endoperoxide antimalarials exert early effects on digestive vacuole morphology. Antimicrobial Agents Chemother., 52: 98-109.
- Dhingra, V., K.V. Rao and M.L. Narasu, 2000. Current status of artemisinin and its derivatives as antimalarial drugs. Life Sci., 66: 279-300.
- Eckstein-Ludwig, U., R.J. Webb, I.D.A. van Goethem, J.M. East and A.G. Lee *et al.*, 2003. Artemisinins target the SERCA of *Plasmodium falciparum*. Nature, 424: 957-961.
- Efferth, T. and B. Kaina, 2010. Toxicity of the antimalarial artemisinin and its derivative. Crit. Rev. Toxicol., 40: 405-421.
- Fiske, C.H. and Y. Subbarow, 1925. The colorimetric determination of phosphorus. J. Biol. Chem., 66: 375-400.
- Galluzi, L., E. Morselli, O. Kepp and G. Kroemer, 2009. Targeting post mitochondrial effectors of apoptosis for neuroprotection. Biochem. Biophys. Acta Bioenergetics, 1787: 402-413.
- Gunter, T.E. and D.R. Pfeifer, 1990. Mechanisms by which mitochondria transport calcium. Am. J. Physiol., 258: 755-786.
- Hietala, S.F., A. Martensson, B. Ngasala, S. Dahlstrom and N. Lindegardh *et al.*, 2010. Population pharmacokinetics and pharmacodynamics of artemether and lumefantrine during combination treatment in children with uncomplicated falciparum malaria in Tanzania. Antimicrob. Agents Chemother., 54: 4780-4788.
- Krishna, S., A.C. Uhlemann and R.K. Haynes, 2004. Artemisinins: Mechanism of action and potential for resistance. Drug Resistance Updates, 7: 233-244.
- Lapidus, R.G. and P.M. Sokolove, 1993. Spermine inhibition of the permeability transition of isolated rat liver mitochondria: An investigation mechanism. Arch Biochem. Biophys., 306: 246-253.
- Lardy, H.A. and H. Wellman, 1953. The catalyst effect of 2,4 dinitrophenol on adenosinetriphosphate hydrolysis by cell particles and soluble enzymes. J. Biol. Chem., 201: 357-370.
- Li, W., W. Mo, D. Shen, L. Sun and J. Wang *et al.*, 2005. Yeast model uncovers dual roles of mitochondria in action of artemisinin. PLoS Genet., Vol. 1.
- Lowry, O.H., N.J. Rosebrough, A.L. Farr and R.J. Randall, 1951. Protein measurement with the Folin phenol reagent. J. Biol. Chem., 193: 265-275.
- Moore, J.C., H. Lai, J.R. Li, R.L. Ren, J.A. McDougall, N.P. Singh and C.K. Chou, 1995. Oral administration of dihydroartemisinin and ferrous sulfate retarded implanted fibrosarcoma growth in the rat. Cancer Lett., 98: 83-87.
- Mpiana, P.T., B.K. Mavakala and Y. Zhi-Wu, 2007. Interaction of artemisinin based antimalarial drugs with hemin in water-DMSO mixture. Int. J. Pharmacol., 3: 302-310.

- Mwesigwa, J., S. Parikh, B. McGee, P. German and T. Drysdale *et al.*, 2010. Pharmacokinetics of artemether-lumefantrine and artesunate and amodiaquine in Children in Kampala, Uganda. Antimicrob. Agents Chemother., 54: 52-59.
- Nelson, D.L. and M.M. Cox, 2008. Lehninger Principles of Biochemistry. 5th Edn., W.H. Freeman and Co., New York, pp. 707-772.
- Noori, S., G.A. Naderi, Z.M. Hassan, Z. Habibi, S.Z. Bathaie and S.M.M. Hashemi, 2004. Immunosuppresive activity of a molecule isolated from *Artemisia annua* on DTH responses compared with cyclosporine A. Int. Immunopharmacol., 4: 1301-1306.
- Novartis Pharma AG., 2009. Coartem® prescribing information. Novartis Pharma AG, May 2009. http://www.coartem.com/.
- Onyesom, I. and J.E. Agho, 2011. Changes in serum glucose and triacylglycerol levels induced by the co-administration of two different types of antimalarial drugs among some >malarial patients in edo-delta region of Nigeria. Asian J. Sci. Res., 4: 78-83.
- Oyemitan, I.A., E.O. Iwalewa, O.E. Ukponmwan, R.O. Akomolafe and O.M. Daniyan, 2007. The involvement of serotonin in artemether-induced behavioural activities. J. Biol. Sci., 7: 575-578.
- Premji, Z.G., 2009. Coartem<sup>®</sup>: The journey to the clinic. Malaria J., Vol. 8. 10.1186/1475-2875-8-S1-S3.
- Raji, Y., I.O. Osonuga, O.S. Akinsomisoye, O.A. Osonuga and O.O. Mewoyeka, 2005. Gonadotoxicity evaluation of oral artemisinin derivative in male rats. J. Medical Sci., 5: 303-306.
- Schneider, W.C. and G.H. Hogeboom, 1950. Intracellular distribution of enzymes. V. Further studies on the distribution of cytochrome c in rat liver homogenate. J. Biol. Chem., 183: 123-128.
- Skulachev, V.P., 1996. Role of uncoupled and non-coupled oxidations in maintenance of safely low levels of oxygen and its one-electron reductants. Q. Rev. Biophys., 29: 169-202.
- Vinogradov, A.D., 2000. Steady-state and pre-steady kinetics of mitochondrial F<sub>1</sub>F<sub>0</sub> ATPase. Is ATP synthase a reversible molecular machine. J. Exp. Biol., 203: 41-49.
- WHO, 2006. Guidelines for Treatment of Malaria. World Health Organization, Geneva.
- Wang, J., L. Huang, J. Li, Q. Fan, Y. Long, Y. Li and B. Zhou, 2010. Artemisinin directly targets malarial mitochondria through its specific mitochondrial activation. PLoS ONE, Vol. 5, 10.1371/journal.pone.0009582.
- Zablockaite, D., V. Gendviliene, I. Martisiene and J. Jurevicius, 2007. Effect of oxidative phosphorylation uncoupler FCCP and F1F0-ATPase inhibitor oligomycin on the electromechanical activity of human myocardium. Adv. Med. Sci., 52: 89-93.
- Zamzami, N., P. Marchetti, M. Castedo, C. Zanin, J.L. Vayssiere, P.X. Petit and G. Kroemer, 1995.
  Reduction in mitochondrial potential constitutes an early irreversible step of programmed lymphocyte death in vivo. J. Exp. Med., 181: 1661-1672.
- Zoratti, M. and I. Szabo, 1995. The mitochondrial permeability transition. Biochim. Biophys. Acta, 1241: 139-176.
- Zorov, D.B., K.W. Kinnally and H. Tedeschi, 1992. Voltage activation of heart inner mitochondrial membrane channels. J. Bioenerg. Biomembr., 24: 119-124.