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Purification and Biochemical Characterization of Acid Phosphatase from *Vigna aconitifolia*

Mohammed A. Al-Omair

Department of Chemistry, College of Science, King Faisal University, Saudi Arabia

Abstract: In present study, isolation and purification of Acid Phosphatase (AP) from the shoot of *Vigna aconitifolia* were conducted. The purification processes included the enzyme precipitation by ammonium sulphate and chromatographic adsorption by DEAE-cellulose and sephadex G200. This study showed a purification of AP up to 60 folds with specific activity of 280 U mg⁻¹ protein. The optimal pH value was found to be 5.4. By studying the relationship between log V and pH, it was found that two amino acid residues namely cysteine and histidine are involved in the catalytic activity of AP. This indicate that *Vigna aconitifolia* AP is an -SH group dependent enzyme. The highest enzyme activity was recorded after 30 min of incubation in the reaction mixture. The optimal temperature for AP activity was 30°C. The activation energy was 0.44 kJ mol⁻¹. After 50°C the *Vigna aconitifolia* AP activity was decreased continuously by prolongation of incubation period. When fructose-6-phosphate and sodium phytate were used instead of p-nitrophenyl phosphate (p-NPP), they expressed 44.3 and 67.1%, respectively of the enzyme activity with p-NPP as substrate. K_m value for p-NPP was 133 mM and V_{max} 27.8 nmol min⁻¹.

Key words: Specific activity, optimum pH, optimum temperature, active sites and enzyme kinetics

INTRODUCTION

Acid phosphatases (EC 3.1.3.2) produced by both prokaryotic and eukaryotic cells and are presumed to convert organic phosphorus into available P_i (Ehsanpour and Amini, 2003; Amlabu *et al.*, 2009). Phosphate is an important molecule for cellular growth that involved in many different biological reactions (Schachtman *et al.*, 1998). The hydrolysis of phosphomonoesters by phosphatases in biological systems is an important process. This process is linked to energy metabolism, metabolic regulation and a wide variety of cellular signal transduction pathways (Allan *et al.*, 1994). The role of acid phosphatase in phosphorus metabolism has been extensively studied in prokaryotic and eukaryotic systems (Duff *et al.*, 1994). The physiological function of acid phosphatase is to provide inorganic phosphate for the cellular growing (Carswell *et al.*, 1997).

The liberation of phosphate from phosphate ester is mainly affected by phosphatase (Bewley and Black, 1994). Deficiency of phosphorus in plants leads to metabolic disorders such as a decrease in photosynthesis, respiration and retardation of growth (Garcia-Sánchez *et al.*, 1996). Phosphatases have been traditionally classified as being alkaline or Acid Phosphatase (AP) according to their optimum pH for catalytic activity above or below pH 7.0 (Barret-Lannard *et al.*, 1982; Shahbazkia *et al.*, 2009). Several intercellular acid phosphatases have been suggested to have a role in the hydrolysis of intercellular polyphosphates (Huber and Hamel, 1985).

Acid phosphatase was found to be localized in the cell walls of *Pisum sativum* (Olmos and Hellin, 1997), root cap of rice (Chen *et al.*, 1992), phloem of *Nicotiana tabacum* (Bentwood and Cronshaw, 1975) and xylem of *Phaseolus vulgaris* (Charvat and Easu, 1974). The mechanism which regulates acid phosphatase distribution and activity is unclear, although their abundance may be regulated by the level of phosphate in the environment and their activity is clearly influenced by local polyelectrolytes (Tu *et al.*, 1990). Acid phosphatase expresses its isozymes in many plants such as soybean seeds (Ferreira *et al.*, 1998); *Vigna sinensis* (Biswas and Cundiff, 1991), tea leaves (Baker and Tadakazu, 1973) and lentil seeds (Bose and Taneja, 1998).

In the present investigation, we purified and characterized the AP from the shoot of *Vigna aconitifolia* as a step towards understanding its properties. The optimum pH and temperature were determined. Moreover, the catalytic active sites were characterized to get into the AP structural functional relationship.

MATERIALS AND METHODS

Chemicals

All the chemicals obtained from Sigma and BDH chemicals LTD and stored as directed by supplier.

Plant Materials

The experimental plant used in this investigation was *Vigna aconitifolia*. Pure strain of seeds was used.

Seed Germination

Seeds of *Vigna aconitifolia* were sterilized with 0.1% sodium hypochlorite and soaked overnight in sterile water at room temperature. The wet seeds were germinated on filter papers moistened with distilled water in petri dishes until complete germination. The germinated seeds were sown in plastic pots containing 1 kg of oven dry soil. Then they were placed in a growth chamber maintained at 27/20°C light/dark (12 h) and grown for 10 days. At the end of the experiment, the shoot systems were washed with deionized water and blotted on paper towels.

Preparation of the Crude Extract of AP

Ten gram of collected shoots from 10 days old seedlings of *Vigna aconitifolia* were homogenized in a blender with 30 mL of 100 mM Tris-HCl (pH 5.0) containing 2 mM MgCl₂ and 2% polyvinylpyrrolidone (PVP). The homogenate was filtered through four layers of muslin and centrifuged at 5000 rpm for 20 min. The supernatant was filtered through a layer of absorbent cotton to get rid of fatty layer. The filtrate was used for AP purification.

Purification of AP

Ammonium sulphate were added and dissolved in a 30 mL of the crude extract, prepared as described above, to make a final concentration of 30% in the mixture. The extract was left for 1 h in a refrigerator at 4°C and then centrifuged at 10,000 rpm for 20 min in a refrigerated centrifuge (MLW WEB Medizintechnik, Leipzig GDR) at 4°C, to collect the precipitated proteins. To a 30 mL of the resulted supernatant ammonium sulphate were added in order to make a final concentration of 70%. Proteins having catalytic activity were precipitated and left to stand at 4°C for 2 h. The precipitated protein was collected by centrifugation at 10,000 rpm for 30 min at 4°C. The precipitated proteins were dissolved in 100 mM phosphate

buffer (pH 5.4) and the catalytic activity was determined. The protein solution containing AP activity was applied to column (2.5×10) containing DEAE-cellulose. The enzyme was eluted by 100 mM phosphate buffer (pH 5.4) containing gradient 0.1- 0.5 M NaCl. The rate of elution was 30 mL h⁻¹ at 4°C and 3 mL fractions were collected. The fractions with maximum AP activity were combined and applied to Sephadex G200 column (2×20 cm). The enzyme was eluted by 200 mM phosphate buffer (pH 5.4) containing gradient 0.1-0.4 M NaCl. The rate of elution was 20 mL h⁻¹ at 4°C and 2 mL fractions were collected. The fractions with maximum activity were combined and used for subsequent kinetic studies.

Acid Phosphatase Assay

The assay method of Park and van Etten (1986) for AP was modified in the present study. The AP reaction buffer was consisted of 100 mM sodium acetate (pH 5.4), 10 mM MgCl₂, 6 mM p-NPP as the substrate and protein extract. Assays were initiated by addition of substrate. The reaction was terminated by the addition of developing reagent. This reagent was prepared daily and consisted of 4 volumes freshly made 10% (w/v) ascorbic acid to 1 volume of 10 mM ammonium molybdate in 15 mM Zn-acetate (pH 5.0) solution. After addition of developing reagent, the samples were incubated for 30 min at 30°C and the A₆₃₀. To calculate activities, a standard curve in the range of 1- 150 nmol P_i was constructed.

Protein Determination

Protein concentration was measured throughout the purification steps by the method of Lowry *et al.* (1951).

RESULTS

Table 1 shows that the protein content decreased progressively throughout the different steps of purification. It should be noticed that the enzyme activity in the crude extract was small and then increased gradually throughout the purification process. The final specific activity was 280 units mg⁻¹ protein. At this final step of purification the enzyme activity was 60 fold purified. The partially purified enzyme from DEAE cellulose was applied to the column of Sephadex G200. The fractions were collected and the enzyme activity was measured in each fraction. The results are demonstrated in Fig. 1. It seems from the results that fraction no.6 contained the highest activity of AP, where the sharp peak appeared.

Table 1: Purification of AP from *Vigna aconitifolia*

Purification steps	Total protein (mg)	Total activity (U)	Specific activity (U mg ⁻¹ protein)	Purification fold	Yield (%)
Crude extract	280.0	1400	5	1.0	100.0
NH ₄ SO ₄ (30-70 %)	57.0	648	12	2.4	46.3
DEAE-cellulose	2.5	235	94	18.8	16.8
Sephadex G-200	0.3	84	280	60.0	6.0

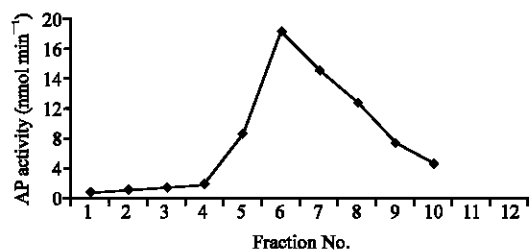


Fig. 1: Elution profile for AP from sephadex G-200 chromatography

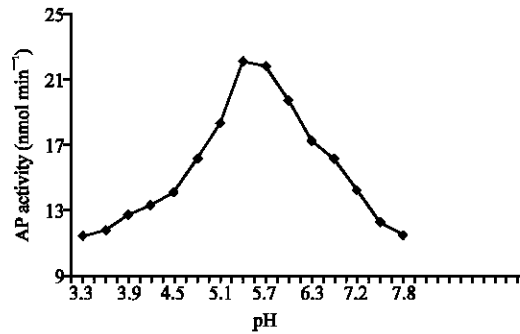


Fig. 2: Effect of pH on AP activity

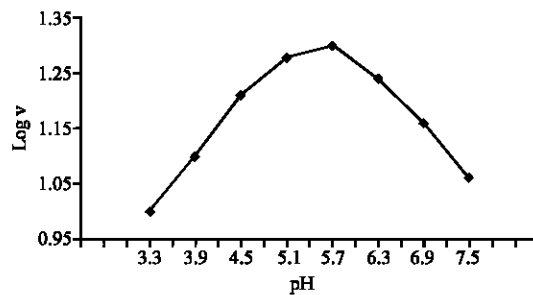


Fig. 3: Determination of the apparent dissociation constant of the groups taking part in catalytic activity of AP

On the study of pH effect on the enzyme activity, it was necessary to fix the other factors affecting the enzyme activity such as enzyme concentration, substrate concentration, $MgCl_2$ concentration, as well as the temperature under which the reaction mixture was incubated. The results are illustrated graphically in Fig. 2. It appears from the figure that the enzyme activity increased gradually from pH 3.6 to 5.4 which seems to be the optimum value after which the enzyme activity started to decline. Plotting $\log V$ against pH as shown in Fig. 3, that there are two pK_a values at 5.4 and 5.7 implicating the presence of histidyl and cysteine sulfhydryl groups (Price and Stevens, 1982).

It shows a considerable linear relationship was obtained at lower p-NPP concentrations where, a little increase at higher concentrations can be observed. Plotting the reciprocal of the reaction rate (V^{-1}) against the reciprocal of p-NPP concentration (S^{-1}) gave a straight line (Fig. 4). According to Lineweaver and Burk hypothesis, both V_{max} and K_m for AP were calculated. The values of the parameters values were $27.8 \text{ nmol min}^{-1}$ and 133 mM . Plotting $\log S$ against $\log V_0/V_{max} - V_0$ gave straight line (Fig. 5) which is known as Hill coefficient (h) whose value was 2. The Hill coefficient (h) is equal to the binding sites.

The effect of other phosphate compounds, e.g., fructose-6-phosphate and sodium phytate, on the enzyme activity were also studied (Table 2). It is apparent from the results that p-nitrophenyl phosphate is the best substrate for AP. The activity of AP enzyme with fructose-6-phosphate was 44.3% of that detected with glucose while, sodium phytate expressed 67.1% of that measured with p-nitrophenyl phosphate as a substrate.

AP activity was measured at different periods of incubation (10-60 min). The results of these measurements are shown in Fig. 6. These results clearly show that the enzyme activity

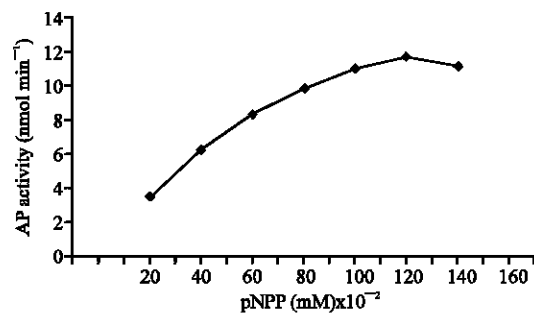


Fig. 4: Effect of pNPP concentration on AP activity

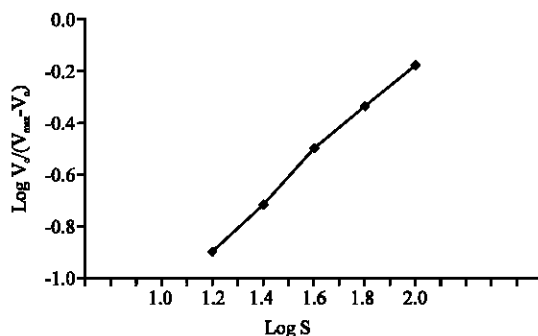


Fig. 5: Hill plot for AP

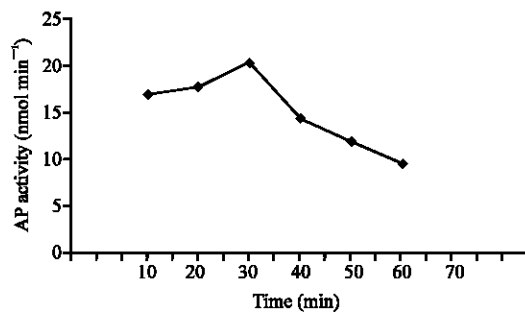


Fig. 6: Effect of incubation time on AP activity

Table 2: Effect of different substrates on AP activity

Substrate	Enzyme activity (nmol min ⁻¹)	Activity (%)
p-nitrophenyl phosphate	16.7	100.0
fructose-6-phosphate	7.4	44.3
Sodium phytate	11.2	67.1

increased slowly from 10 up to 30 min at which highest enzyme activity was recorded and after which the activity declined gradually. Therefore, an incubation period of 30 min was chosen for AP mixtures throughout the present study.

The activity of AP was studied in a temperature range of 10-80°C. The results are illustrated in Fig. 7. The optimal temperature for AP was 30°C. Any further increase after the optimal temperature leads to a noticeable reduction in the activity. There is an apparent

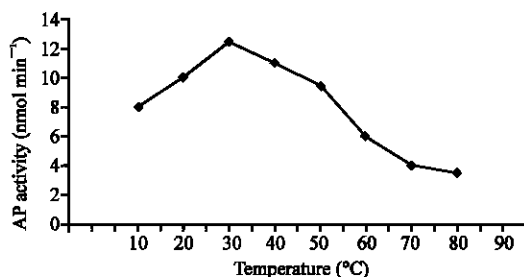


Fig. 7: Effect of temperature on AP activity

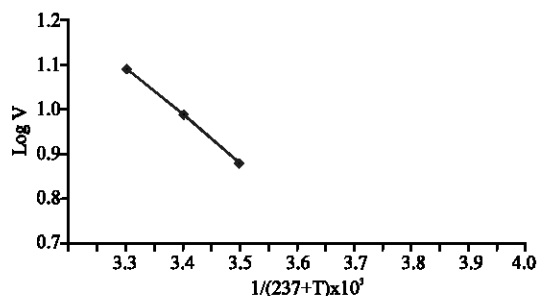


Fig. 8: Arrhenius plot for AP

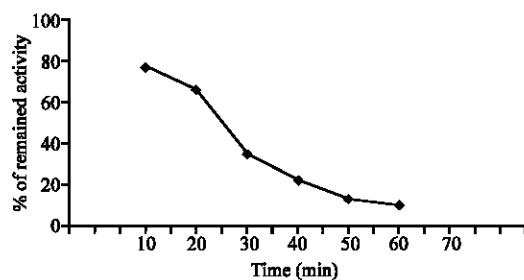


Fig. 9: Rate of heat inactivation of AP

linearity between the enzyme activity and the temperature in the range 10-80°C and the highest enzyme activity is recorded at 30°C, which seems to be the optimum. After 30°C the activity declined gradually. When the logarithm of the reaction rate is plotted versus the reciprocal of the absolute temperature a straight line was obtained (Fig. 8) and the activation energy was calculated and its value was 0.44 kJ mol⁻¹.

The rate of heat inactivation of AP activity was studied by plotting the percentage of the remaining activity against time at 50°C. The results are illustrated in Fig. 9. It could be considered that AP from *Vigna aconitifolia* was labile above 30°C and its activity declined after 10 min since the remaining activity was 76% which decreased to 66.3% after 20 min. After 50 min the enzyme activity was approximately abolished.

The enzyme activity was measured at different concentrations of the purified AP (Fig. 10) in volumes ranging from 0.1 to 0.7 mL of the enzyme solution. The results indicated

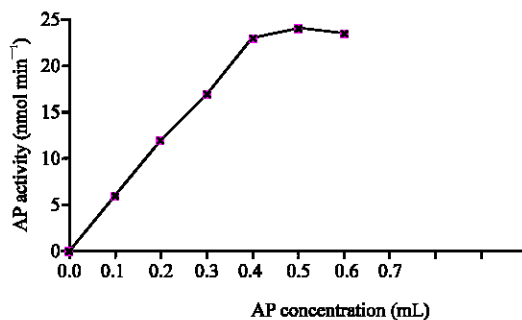


Fig. 10: Effect of AP concentration on the reaction activity

that by increasing the enzyme concentration, there was a corresponding increase in the rate of reaction and a linear relationship was obtained.

DISCUSSION

The technique adopted for acid phosphatase purification from the *Vigna aconitifolia* shoot included precipitation of the crude extract by ammonium sulphate and using DEAE-cellulose and Sephadex G200 columns. The purification technique of acid phosphatase showed that the specific activity was 280 U mg⁻¹ protein. This value is higher than that reported for AP from *Stichopus japonicus* and *Trichoderma harzianum* (Zhu *et al.*, 2009; Leitão *et al.*, 2010) and is similar to that reported for AP from tobacco cells (Okuda *et al.*, 1987). However, this value was higher than those reported for the acid phosphatase from seeds of sunflower, maize endosperm cultures, seeds of *Vigna sinensis* (Park and van Etten, 1986; Miernyk, 1992; Tapan *et al.*, 1996).

The optimum pH of the purified acid phosphatase was 5.4. This value is consistent with that reported for maize endosperm (Miernyk, 1992) and is higher than pH 5 reported for acid phosphatase from germinating seeds of *Vigna sinensis* (Biswas and Cundiff, 1991). Moreover, the obtained pH 5.4 for acid phosphatase from *Vigna aconitifolia* shoot is lower than the values obtained for the enzyme from lupin (*Lupinus albus*) seeds (Zheng and Duranti, 1995; Tapan *et al.*, 1996); sunflower seeds (Park and van Etten, 1986); roots of *Trifolium* (Julie *et al.*, 1999; Zhang and McManus, 2000) and cotton seedlings (Bhargava and Sacher, 1987).

Generally, enzymes are sensitive to changes in pH. The pH can influence the enzyme activity in a number of ways. Firstly, it can change the ionization of the enzyme substrate complex. Secondly, it can change the ionization of various groups of the enzyme molecule which may affect the affinity of the enzyme for the substrate. Thirdly, it changes the ionization of the substrate which may affect the binding of the substrate to the enzyme. Fourthly, at extreme pH it can bring about changes in protein structure (Palmer, 1985).

The present results showed that acid phosphatase has a broad substrate specificity with highest activities with p-nitrophenyl phosphate and fructose-6-phosphate while sodium phytate showed a little activities compared to p-nitrophenyl phosphate. These results are in harmony with the findings of other workers (Penheiter *et al.*, 1997; Wannet *et al.*, 2000). Plotting V⁻¹ versus S⁻¹ gave a linear relationship from which K_m value for acid phosphatase was calculated. The obtained value of K_m of acid phosphatase for p-nitrophenyl phosphate was 133 mM which is higher than 35 mM reported for the enzyme from maize endosperm (Miernyk, 1992).

The optimal incubation time for acid phosphatase was 30 min. Similar incubation time (30 min) was reported for acid phosphatase from tobacco cells (Okuda *et al.*, 1987).

The optimal temperature of acid phosphatase was 30°C. A higher value was 37°C which was reported for the enzyme from cotton seedlings (Bhargava and Sachar, 1987).

Plotting $1/T+273$ against $\log V$ resulted in a straight line in case of acid phosphatase which is in agreement with Arrhenius plot (Palmer, 1985). From this relationship the activation energy was calculated. The value of activation energy for acid phosphatase was 9.44 kJ mol^{-1} .

In addition to the enzyme and substrate, many enzymes require, a non-protein substances for the catalytic reaction to proceed. These substances are known as cofactors and are usually known as activators, which probably affect the conformation of the enzyme producing a more catalytically active state, or interact with the substrate.

CONCLUSION

In conclusion, the characterized acid phosphatase from the shoot of *Vigna aconitifolia* would be useful for adapting organic phosphorus into available P_i that involved in many different biological reactions.

REFERENCES

- Allan, A.C., M.D. Fricker, J.L. Ward, M.H. Beale and A.J. Trewavas, 1994. Two Transduction pathways mediate rapid effects of abscisic acid in *Commelina* guard cells. *Plant Cell*, 6: 1319-1328.
- Amlabu, E., A.J. Nok and A.B. Sallau, 2009. Purification and biochemical characterization of lysosomal acid phosphatases (EC 3.1.3.2) from blood stream forms, *Trypanosoma brucei brucei*. *Parasitol. Int.*, 58: 238-242.
- Baker, J.E. and T. Tadakazu, 1973. Acid phosphatase in plant tissues. I. Changes in activity and multiple forms in tea leaves and tomato fruit during maturation and senescence. *Plant Cell Physiol.*, 14: 459-471.
- Barret-Lennard, E.D., A.D. Robson and H. Greenway, 1982. Effect of phosphorus deficiency and water deification phosphatase activity from wheat leaves. *J. Exp. Bot.*, 33: 682-693.
- Bentwood, B.J. and J. Cronshaw, 1975. Biochemistry and cytochemical localization of acid phosphatase in the phloem of *Nicotiana tabacum*. *Planta*, 130: 97-104.
- Bewley, J.D. and M. Black, 1994. *Seeds: Physiology of Development and Germination*. 2nd Edn., Plenum Press, New York.
- Bhargava, R. and R.C. Sacher, 1987. Induction of acid phosphatase in cotton seedlings: Enzyme purification, subunit structure and kinetic properties. *Phytochemistry*, 26: 1293-1297.
- Biswas, T.K. and C. Cundiff, 1991. Multiple forms of acid phosphatase in germinating seeds of *Vigna sinensis*. *Phytochemistry*, 30: 2119-2125.
- Bose, S.K. and V. Taneja, 1998. Induction of a germination specific, low molecular weight, acid phosphatase isozyme with specific phosphotyrosine phosphatase activity in lentil (*Lens esculenta*) seeds. *Biochem. Biophys. Res. Commun.*, 250: 629-634.
- Carswell, M.C., B.R. Grant and W.C. Plaxton, 1997. Disruption of the phosphate-starvation response of oilseed rape suspension cells by the fungicide phosphonate. *Planta*, 203: 67-74.
- Charvat, I.D. and K. Easu, 1974. An ultrastructural study of acid phosphatase localization in *Phaseolus vulgaris* xylem by the use of an azo-dye method. *J. Cell Sci.*, 19: 543-561.

- Chen, W.S., Y.F. Huang and Y.R. Chen, 1992. Localization of acid phosphatase in root cap of rice plant. *Bot. Bull. Acad. Sin.*, 33: 233-239.
- Duff, S.M.G., G. Sarath and W.C. Plaxton, 1994. The role of acid phosphatases in plant phosphorus metabolism. *Physiol. Plant.*, 90: 791-800.
- Ehsanpour, A. and F. Amini, 2003. Effect of salt and drought stress on acid phosphatase activities in alfalfa (*Medicago sativa* L.) explants under *in vitro* culture. *Afr. J. Biotechnol.*, 2: 133-135.
- Ferreira, C.V., J.M. Granjeiro, E.M. Taga and H. Aoyama, 1998. Soybean seed acid phosphatases: Unusual optimum temperature and thermal stability studies. *Biochem. Biophys. Res. Commun.*, 242: 282-286.
- García-Sánchez, M.J., J.A. Fernández and F.X. Niell, 1996. Photosynthetic response of P-deficient *Gracilaria tenuistipitata* under two different phosphate treatments. *Physiol. Plant.*, 96: 601-606.
- Huber, A.L. and K.S. Hamel, 1985. Phosphatase activities in relation to phosphorus nutrient in *Nodularia spumigena* (Cyanobacterales). *Hydrobiologia*, 123: 81-88.
- Julie, E.H., A.E. Richardson and R.J. Simpson, 1999. Phytase and acid phosphatase activities in extracts from roots of temperate pasture grass and legum seedlings. *Aust. Plant Physiol.*, 26: 801-809.
- Leitão, V.O., R.C. de Melo Lima, M.H. Vainstein and C.J. Ulhoa, 2010. Purification and characterization of an acid phosphatase from *Trichoderma harzianum*. *Biotechnol. Lett.* 10.1007/s10529-010-0264-2
- Lowry, D.H., N.J. Rosenbrough, A.L. Far and R.J. Randal, 1951. Protein measurement with folin phenol reagent. *J. Biol. Chem.*, 193: 265-275.
- Miemyk, J.A., 1992. Purification and characterization of the major acid phosphatase isozyme by maize endosperm cultures. *Phytochemistry*, 31: 2613-2616.
- Okuda, S., J. Kaneko, T. Ogawa, T. Yamaguchi, K. Izaki and H. Takahashi, 1987. Increase in enzyme activities in embryonic axes of soybean seeds during germination. *Agric. Biol. Chem.*, 51: 109-113.
- Olmos, E. and E. Hellin, 1997. Cytochemical localization of ATPase plasma membrane and acid phosphatase by cerium-based method in a salt-adapted cell line of *Pisum sativum*. *J. Exp. Bot.*, 48: 1529-1535.
- Palmer, T., 1985. *Understanding Enzymes*. Harwood Publisher, Chichester, UK.
- Park, H.C. and R.L. van Etten, 1986. Purification and characterization of homogeneous sunflower seed acid phosphatase. *Phytochemistry*, 25: 351-357.
- Penheiter, A.R., S.M.G. Duff and G. Sarath, 1997. Soybean root nodule and acid phosphatase. *Plant Physiol.*, 114: 597-604.
- Price, N.C. and L. Stevens, 1982. *Fundamentals of Enzymology*. Oxford University Press, Oxford, UK.
- Schachtman, D.P., R.J. Reid and S.M. Ayling, 1998. Phosphorus uptake by plants: From soil to cell. *Plant Physiol.*, 116: 447-453.
- Shahbazkia, H.R., S. Sharifi and B. Shareghi, 2009. Purification and kinetic study of bone and liver alkaline phosphatase isoenzymes in the dog. *Comp. Clin. Pathol.*, 19: 81-84.
- Tapan, K., M. Promo and B. Biswas, 1996. Purification of acid phosphatase I from germinating seeds of *Vigna sinensis*. *Phytochemistry*, 41: 1457-1458.
- Tu, S.I., J.R. Lavanaugh and R.T. Roswell, 1990. Phosphate uptake by excised maize roots tips studies by *in vivo* 31P nuclear magnetic resonance spectroscopy. *Plant Physiol.*, 93: 778-784.

- Wannet, W.J.B., R.W. Wassenaar, H.J.M. M. Jorissen, C. Drift and H.J.M. Camp, 2000. Purification and characterization of an acid phosphatase from the commercial mushroom *Agaricus bisporus*. *Antonie van Leeuwenhoek*, 77: 215-222.
- Zhang, C. and M.T. McManus, 2000. Identification and characterisation of two distinct acid phosphatases in cell walls of roots of white clover. *Plant Physiol. Biochem.*, 38: 259-270.
- Zheng, Y. and M. Duranti, 1995. Molecular properties and thermal secretion of lupin seed acid phosphatase. *Phytochemistry*, 40: 21-22.
- Zhu, B.W., J.W. Yu, Z. Zhang, D.Y. Zhou, J.F. Yang, D.M. Li and Y. Murata, 2009. Purification and partial characterization of an acid phosphatase from the body wall of sea cucumber *Stichopus japonicus*. *Process Biochem.*, 44: 875-879.