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Thickness and Oxygen Partial Pressure Dependence on Optical Band Gap of Indium Oxide by Reactive Evaporation Method

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Abstract: Indium oxide film is deposited by reactive evaporation of indium in the presence of oxygen gas onto an unheated glass substrate. It was found that thickness of the film and partial oxygen pressure during the deposition affects the optical properties of the indium oxide thin film. We studied the optical band gap for different thickness and partial pressure keeping a constant annealing temperature. It was found that the band gap varies from 3.5-3.8 eV, as thickness of the film increased. The band gap energy had also shows the similar trend and it was also studied as a function of annealing temperature. A systematic investigation of the optical band gap as a function of thickness and oxygen partial pressure at different annealing temperature was carried out.

Key words: Reactive evaporation, transparent oxide, optical properties, thin film

INTRODUCTION

Indium oxide (In_2O_3) is a transparent conducting oxide with wide band gap n-type oxide semiconductor. It shows low resistivity which is an unusual property for a wide band gap material, compared with conventional extrinsic semiconductor. The good conductivity of oxide semiconductor film is depends on the non-stoichiometry of the film which leads to the deficiency in oxygen. Indium oxide thin films are characterized by high transparency in the visible region of the spectrum and low transmittance in the infrared region.

Transparent Conductive Oxide (TCO) material is an interesting semiconductor owing to its unique properties of both high electrical conductivity and high optical transparency (Parthiban *et al.*, 2009; Koida and Kondo, 2006). As a matrix for the generation of transparent conducting oxides, indium oxide (In_2O_3) has been the subject of numerous studies for liquid crystal displays (Falcony *et al.*, 1985), solar cell (Granqvist, 1993), sensors (Bender *et al.*, 2001), nanowire technology (Li *et al.*, 2003), light emitting diodes and other optoelectronic devices. In the present study, we have investigated the optical properties of Indium oxide thin film by reactive evaporation method on a glass substrate.

MATERIALS AND METHODS

The films of Indium Oxide (In_2O_3) were prepared by reactive evaporation technique using HINDHIVAC-

12A4D high vacuum coating unit. Resistively heated Molybdenum (Mo) source in the form of boat was used for evaporation of Indium (purity 99.999%). The evaporation was carried out in the presence of oxygen plasma. The vacuum chamber was initially evacuated to base pressure 10^{-3} Pa using diffusion pump. By adjusting the valve the chamber pressure could be varied between 0.01 and 0.1 Pa during deposition by admitting oxygen to the chamber. Optically flat glass slides were used as substrates. The cleaning procedure adopted was as follows: the films were kept in freshly prepared hydrochloric acid overnight and then washed with distilled water. The glass plates were then kept in a detergent solution and cleaned using isopropyl. Finally, the glass plates were cleaned with acetone. The thickness of these films was obtained by quartz crystal monitor. The thickness varied between 500-1200 Å keeping deposition rate 0.5 Å sec^{-1} .

The films were placed on Muffle furnace and heated for different temperature in open air to investigate annealing effect. The films thus prepared were characterized optically. A UV/Vis double beam spectrophotometer was used to study the optical characterization of these films.

RESULTS AND DISCUSSION

A typical transmission spectrum of a In_2O_3 thin film with a thickness of 743 Å was shown in Fig. 1. As can be seen from this figure, the transmission is high in the

visible region. However, it depends on the partial pressure of oxygen during deposition (Naseem *et al.*, 1988) and thickness of the film (Shah *et al.*, 2005).

From the transmission data, the absorption coefficient as a function of photon energy was calculated and using Tauc equation. we plotted the direct allowed transition for the films:

$$\alpha(h\nu) = A(h\nu - E_g)^{1/2}$$

where, E_g the transition energy is gap and $h\nu$ is the photon energy. Figure 2 shows the $(\alpha h\nu)^2$ vs $h\nu$ plots

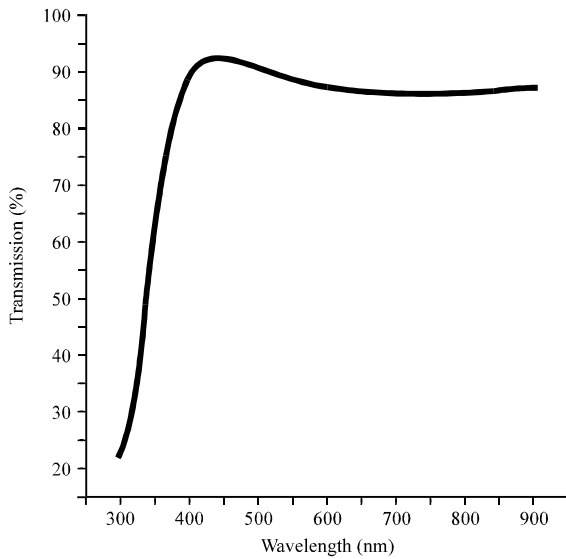


Fig. 1: Transmission of an In_2O_3 film of thickness 743 Å

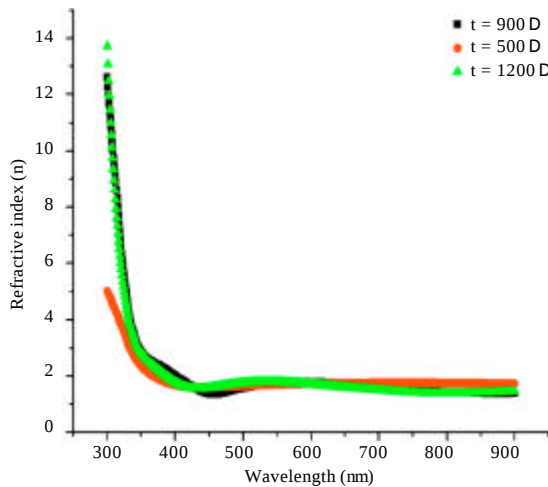


Fig. 2: Plot of $(\alpha h\nu)^2$ vs. $h\nu$ for various thicknesses

for various thicknesses. Extrapolation of linear region of such a curve gives the value of the band gap.

The refractive index n of the film was calculated using the formula 10. Figure 3 shows the variation of refractive index with wavelength:

$$n_f = \left[\frac{n_s(2 - T_{\min}) + 2n_s(1 - T_{\min})^{1/2}}{T_{\min}} \right]^{1/2}$$

Influence of thickness on optical properties: The influence of the film thickness on the optical band gap was studied for films deposited 500-1200 Å thicknesses, with annealing temperature variation 200-400°C. Figure 4 shows the variation of band gap energy as a function of thickness. The calculated band gap energy as a function of thickness, with different annealing temperature are listed in Table 1. The results showed that the optical band gap was increased from 3.58-3.74 by increasing the film thickness and it decreases with increase in annealing temperature. These results are consistent with other published results such as results of

Table 1: Optical band gap dependence on thickness and annealing temperature

Temp. (°C)	500 Å	700 Å	743 Å	900 Å	1200 Å
200	3.66	3.68	3.71	3.74	-
300	3.62	3.63	3.65	3.69	3.71
400	3.58	3.60	3.62	3.63	3.65

Table 2: Optical band gap dependence on oxygen partial pressure and annealing temperature 4

Temp. (°C)	3E-5 mbar	5E-5 mbar	7E-5 mbar	9E-5 mbar	19E-5 mbar
300	3.59	3.61	3.63	3.65	3.71
360	3.58	3.60	3.62	3.64	3.69
400	3.56	3.58	3.60	3.62	3.66

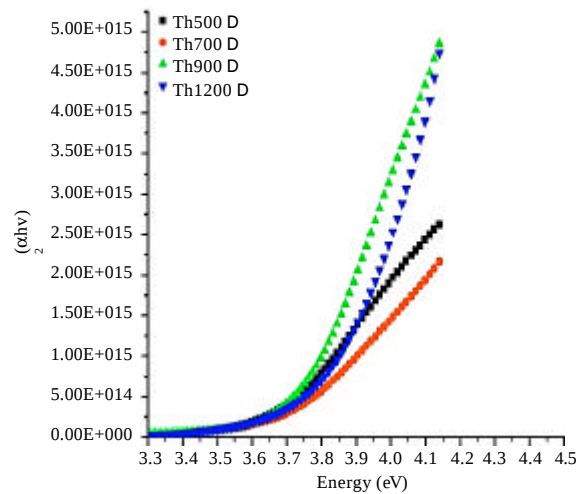


Fig. 3: Plot of refractive index vs. wavelength for various thicknesses

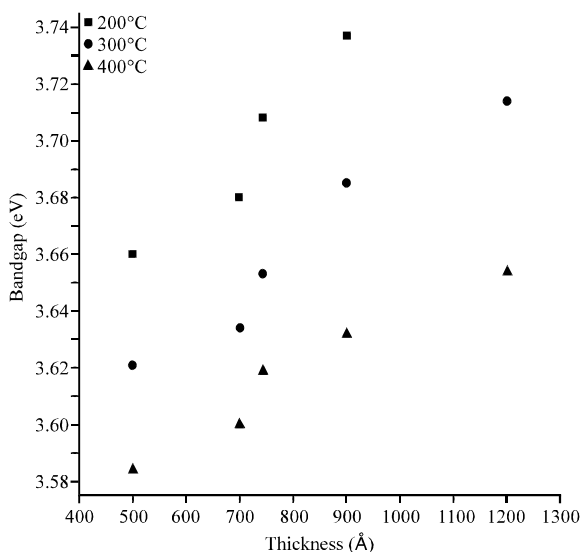


Fig. 4: Plot of E_g vs. thickness for various annealing temperature

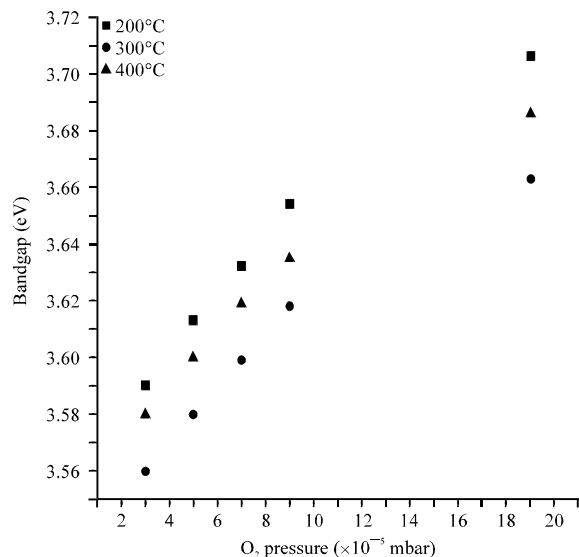


Fig. 5: Plot of E_g vs. oxygen partial pressure for various annealing temperature

Benoy *et al.* (2009) who attribute the increase in the band gap as a function of thickness was due to grain size growth and composition changes taking place in the sample.

Influence of oxygen partial pressure: Figure 5 shows the plot of $(\alpha h\nu)^2$ vs. $h\nu$ for indium oxide film of thickness 743 Å deposited at different oxygen partial pressures and at different annealing temperature. Extrapolation of the linear region of the graph gives the band gap.

The values are given in Table 2. These values are somewhat higher than the bulk value about 3.5 eV. This shift may be attributed to the effect which is observed in many conducting transparent oxides. The band gap is found to increase with oxygen partial pressure. When the oxygen partial pressure is increased, transformation of InO to In₂O₃ is accelerated.

CONCLUSION

Indium oxide thin film was deposited by reactive evaporation of indium in oxygen. We found that the thickness, annealing temperature and oxygen partial pressure dependence on the optical band gap of indium oxide film. We also had shown that the transmittance of annealed film was above 95% in visible region.

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REFERENCES

- Bender M., N. Katsarakis, E. Gagaoudakis, E. Hourdakos, E. Douloufakis, V. Cimalla and G. Kiriakidis, 2001. Dependence of the photoreduction and oxidation behavior of indium oxide films on substrate temperature and film thickness. *J. Applied Phys.*, 90: 5382-5387.
- Benoy, M.D., E.M. Mohammed, M.S. Babu, P.J. Binu and B. Pradeep, 2009. Thickness dependence of the properties of indium tin oxide films prepared by activated reactive evaporation. *Braz. J. Phys.*, 39: 629-632.
- Falcony, C., J.R. Kirtley, D.J. Dimaria, T.P. Ma and T.C. Chen, 1985. Electroluminescence emission from indium oxide and indium-tin-oxide. *J. Applied Phys.*, 58: 3556-3558.
- Granqvist, C.G., 1993. Transparent conductive electrodes for electrochromic devices: A review. *Applied Phys. A*, 57: 19-24.
- Koida T. and M. Kondo, 2006. High-mobility transparent conductive Zr-doped In₂O₃. *Applied Phys. Lett.*, vol 89,
- Li, C., D. Zhang, B. Lei, S. Han, X. Liu and C. Zhou, 2003. Surface treatment and doping dependence of In₂O₃ nanowires as ammonia sensors. *J. Phys. Chem. B*, 107: 12451-12455.

- Naseem, S., I.A. Rauf, K. Hussain and N.A Malik, 1988. Effect of oxygen partial pressure on the properties of reactively evaporated thin films of indium oxide. *Thin Solid Films*, 156: 161-171.
- Parthiban S., E. Elangovan, K. Ramamurthi, R. Martins and E. Fortunato, 2009. High near-infrared transparency and carrier mobility of Mo doped In₂O₃ thin films for optoelectronics applications. *J. Applied Phys.*, vol. 106, 10.1063/1.3224946.
- Shah, M.R., M.K. Alam, M.R. Karim and M.A. Sobhan, 2005. Study of optical properties of indium oxide thin films. *Institute for Computational and Mathematical Engineering*.