

Journal of Environmental Science and Technology

ISSN 1994-7887





∂ OPEN ACCESS

Journal of Environmental Science and Technology

ISSN 1994-7887 DOI: 10.3923/jest.2017.25.34



Research Article CO₂ Absorption from its Mixture Through Super-hydrophobic Membrane Contactor

Sutrasno Kartohardjono, Kevin Antonius Sembiring, Rexy, Reza Syandika, Fauzan Ghasani and Nelson Saksono

Laboratory of Process Intensification, Department of Chemical Engineering, Faculty of Engineering, Universitas Indonesia, Kampus Baru UI, 16424 Depok, Indonesia

Abstract

Objective: The aim of this study is to evaluate the performance of super-hydrophobic hollow fiber membrane contactors to absorb CO_2 from its mixture with N_2 or CH_4 using 5 wt% diethanolamine (DEA) solution as absorbent. **Methodology:** During the experiment the absorbent flowed through lumen fibers, whilst the feed gas flowed through the shell side of the membrane contactor. **Results:** The experimental results demonstrated that the mass transfer coefficients, the fluxes and the amount of CO_2 absorbed increased with the absorbent and feed gas flow rates for CO_2 - N_2 and CO_2 - CH_4 feed gas system. The CO_2 absorption efficiency for both feed gas increased with the absorbent flow rate. The CO_2 absorption efficiency increased with feed gas flow rate for CO_2 - N_2 system, but decreased for CO_2 - CH_4 system. **Conclusion:** The overall mass transfer coefficient and the flux decreased with the number of fibers in the contactor, whilst, the amount of CO_2 absorbed and the CO_2 absorption efficiency increased.

Key words: Flux, mass transfer coefficient, membrane contactor, super-hydrophobic

Received: November 10, 2016

Accepted: December 02, 2016

Published: December 15, 2016

Citation: Sutrasno Kartohardjono, Kevin Antonius Sembiring, Rexy, Reza Syandika, Fauzan Ghasani and Nelson Saksono, 2017. CO₂ absorption from its mixture through super-hydrophobic membrane contactor. J. Environ. Sci. Technol., 10: 25-34.

Corresponding Author: Sutrasno Kartohardjono, Laboratory of Process Intensification, Department of Chemical Engineering, Faculty of Engineering, Universitas Indonesia, Kampus Baru UI, 16424 Depok, Indonesia

Copyright: © 2017 Sutrasno Kartohardjono *et al.* This is an open access article distributed under the terms of the creative commons attribution License, which permits unrestricted use, distribution and reproduction in any medium, provided the original author and source are credited.

Competing Interest: The authors have declared that no competing interest exists.

Data Availability: All relevant data are within the paper and its supporting information files.

INTRODUCTION

Carbon dioxide (CO₂) is one of the most important greenhouse gases produced by industries such as natural gas and power plant industries¹. There are many methods that can be used to capture CO₂ either form natural gas or from flue gas such as absorption through chemical and physical absorbents², absorption through solid surface³ and absorption through membrane contactor⁴. Chemical absorption in absorption column was the most established method for decades especially in petroleum and natural gas industries. However, this method has several disadvantages such as large space, high capital cost and operational problems e.g., liquid channeling, flooding, entrainment and foaming. Therefore, there is a need to develop new alternative technology to enhance the efficiency of absorption process and to reduce the effect of disadvantages⁵. Hollow fiber membrane contactor is a promising alternative and has been applied in many processes such as ammonia removal from wastewater and CO₂ absorption processes^{6,7}. Hollow fiber membrane contactor contains two channels in which absorbent liquid and gas mixture are in contact with each other without getting mixed, so an increase in the fluid velocity through the channels will not lead to the common operational problems in tower such as flooding, entrainment and foaming⁸. Therefore, gas-liquid hollow fiber membrane contactors can be an alternative substitute technology with high removal efficiency for absorption of pollutant⁹.

The CO₂ absorption through the membrane contactor is generally focused on the use of porous membranes, which are suffered if the membrane pores are wetted by the solvent¹⁰. The wetting of membrane pores significantly affected the mass transfer coefficients as the resistance of membrane phase increasing^{11,12}. It was reported that the membrane wetting can cause a significant drop of CO₂ flux for both physical and chemical absorptions up to 90% when water used as the absorbent¹³. In this case, an assembly of ultrathin films incorporating carbonic anhydrase can be applied to microporous hydrophobic membrane to reduce pore wetting¹⁴. Therefore, the selection of the liquid absorbents is an important measures to prevent the wetting of the membrane¹⁵. In addition, the high temperature of absorbent can reduce the mass transfer coefficient in the membrane contactor¹⁶.

The other important element of the membrane contactor is membrane material, which should have important properties such as high hydrophobicity, high surface porosity, low mass transfer resistance as well as resistance to various chemical liquids¹⁷. The high hydrophobicity property is

needed to prevent the wetting of membrane pores that can increase the overall mass transfer coefficient and therefore, the choice of membrane material is critical to the efficiency of the process¹⁸. There are many membrane materials that can be utilized as contactor for gas-liquid operation such as popypropylene (PP)⁹, polyvynilidinefluoride (PVDF)¹⁹ and polyvynilchloride (PVC)²⁰. The membrane can be incorporated with other material such as graphene nano sheets²¹ or prepared as super-hydrophobic membrane²² to improve the hydrophobicity of the membrane. This study aims to evaluate the performance of super hydrophobic hollow fiber membrane contactor to absorb CO₂ using water and diethanolamine (DEA) solution as absorbents. The flow rates of absorbents as well as the number of fibers in the membrane contactor will be observed to see their effects on CO₂ absorption processes in the contactor²³.

MATERIALS AND METHODS

Figure 1 shows the schematic diagram of CO₂ absorption from its mixture with N₂ or CH₄ using a membrane contactor, which has already reported in the literature⁷. The super hydrophobic hollow fiber membrane contactors used were supplied by PT GDP Filter Bandung. The feed gas (CO₂:N₂ = 13:87 and CO₂:CH₄ = 36:64) and DEA were purchased from BOC Gases and Merck, respectively. The super-hydrophobic fiber membrane used is polypropylene-based with outer and inner diameter of about 525 and 235 µm, respectively. Hollow fiber membrane contactors used for CO₂-N₂ mixture were 6 and 25 cm in diameter and length, respectively, which consist of 1000, 2000 and 5000 fibers (FA1000, FA2000 and FA5000). Meanwhile, hollow fiber membrane contactors used for CO₂-CH₄ mixture were 8 and 25 cm in diameter and length, respectively, which consist of 2000, 4000 and 8000 fibers (FB2000, FB4000 and FB8000). During the experiment, the absorbents of DEA solution (5% v in water) flowed through the lumen side of the membrane fiber, whilst the feed flowed through the shell side of the membrane contactor. The absorbents and gas flow rates were measured using liquid flow meter Krohne and mass flow meter Sierra Top Trak Instruments, respectively. Meanwhile, inlet and outlet gas compositions were measured using gas chromatography Bruker Scion 436-GC.

The overall mass transfer coefficient, K_{OVL} , will be calculated by Wang *et al.*²⁴:

$$K_{OVL} = \left(\frac{Q_{Gin}}{A_{m}}\right) ln\left(\frac{C_{in}}{C_{out}}\right)$$
(1)



Fig. 1: Schematic diagram of experimental set up, 1: Feed gas, 2: Mass flow meter, 3: Super-hydrophobic hollow fiber membrane contactor, 4: Mass flow meter, 5: Absorbent reservoir, 6: Flow meter, 7: Absorbent reservoir, 8: Gas chromatography

where, Q_{Gin} , A_m , C_{in} and C_{out} are inlet gas flow rate, membrane surface area and CO₂ concentration in the inlet and outlet gas, respectively. Meanwhile, CO₂ flux through the membrane contactor, J and CO₂ absorption efficiency and R (%) were calculated by:

$$J = (Q_{iGin} - Q_{iGout}) \times \frac{RT}{Am}$$
(2)

$$R(\%) = \frac{100(Q_{iGin} - Q_{iGout})}{Q_{iGin}}$$
(3)

where, Q_{iGin} , Q_{iGout} , T and R are the CO₂ flow rates in the inlet and outlet gas, temperature and gas constant, respectively.

RESULTS AND DISCUSSION

Based on the experiments, there are three mass transfer resistances in the process of CO_2 absorption through the hollow fiber membrane contactors, namely: Mass transfer resistance in the gas phase in the shell side of the membrane contactors; mass transfer resistance in the pores of the membrane fiber and mass transfer resistance in the absorbents phase in the lumen fibers. Therefore, the equation of mass transfer coefficient in the membrane contactors can be written as:

$$\frac{1}{K_{\rm OV}} = \frac{H}{k_{\rm G}} + \frac{H}{k_{\rm m}} + \frac{1}{k_{\rm L}}$$
(4)

where, K_{ov} , k_G , k_m , k_L and H are overall mass transfer coefficient, mass transfer coefficient in gas, membrane and liquid phases and Henry's constant, respectively.

The effect of the absorbent flow rate and the number of fiber in the membrane contactors at the same feed gas flow rate on the overall mass transfer coefficient is shown in Fig. 2. Figure 2 shows that, for the CO₂-N₂ and CO₂-CH₄ system, the mass transfer coefficient increase with increasing the absorbent flow rate in the lumen fiber in the membrane contactor. The mass transfer resistance in the hollow fiber membrane contactor is usually dominated in the liquid phase²⁵. An increase in the absorbent flow rate will increase the turbulence in the liquid boundary layer that will reduce the mass transfer resistance in the liquid phase and based on Eq. 4, will increase the overall mass transfer coefficient^{26,27}. The increase in the mass transfer coefficient with the absorbent flow rates indicated that the membrane was not wetted by the absorbent²⁴. The increase in the overall mass transfer coefficient with increasing solvent flow rate in the membrane contactors also reported by Kim and Yang²⁸. In this study, the overall mass transfer coefficient, K₁, increase from 0.00034-0.00039 and 0.00023-0.00027 cm sec⁻¹ for CO₂-N₂ and CO2-CH4 system, respectively, if the absorbent flow rate increase from 100-500 mL min⁻¹. Meanwhile, Kim and Yang²⁸ reported that the overall mass transfer coefficient, K₁, increase from 0.0014-0.0025 cm sec⁻¹ if the absorbent flow rate increase from 10-130 mL min⁻¹ for CO₂-N₂ (40:60) system using amine solution as absorbent.

The numbers of fibers in the contactors have the same effect for both feed gas on the overall mass transfer coefficient



Fig. 2: Effects of absorbent flow rate, Q_L, on the overall mass transfer coefficient, K_{OVL}, at the membrane contactor consists of 1000, 2000 and 5000 fibers for CO₂-N₂ system and 2000, 4000 and 8000 fibers for CO₂-CH₄ system

at the same absorbent and feed gas flow rate. The increase in the number of fibers in the contactor will reduce the absorbent flow rate in the individual fiber at the same overall absorbent flow rate, which will reduce the overall mass transfer coefficient²⁹. On the other hand, the increase in the number of fibers in the contactor will increase the contact area for the mass transfer, which led to the increase in the overall mass transfer coefficient. Figure 2 shows that the overall mass transfer coefficient for both feed gas decrease with increasing the number of fibers in the contactor at the same absorbent and feed gas flow rate, indicating that the effect of the absorbent flow rate more dominant than the effect of the surface area for contact. Figure 2 also shows that the overall mass transfer coefficient for CO₂-N₂ system is higher than for CO₂-CH₄ system due to the decrease in the CO₂ inlet concentration. The similar phenomena also shown by Wang et al.²⁴, who reported that the overall mass transfer coefficient decrease with increasing the CO₂ concentration in the feed gas using 2 M Na₂CO₃ aqueous solution as absorbent.

Similar to the mass transfer coefficient, the flux of CO₂ through the membrane fibers increase with increasing the absorbent flow rate in the membrane contactor due to the turbulence and surface renewal effects^{26,27,30} as shown in Fig. 3. Several researchers also demonstrated the same trend where the flux increases with increasing solvent flow rate^{16,31}. In this study, the flux, J, increase from 8.3×10^{-6} to 8.8×10^{-6} mmol cm⁻² sec⁻¹ and 2.9×10^{-6} to 3.2×10^{-6} mmol cm⁻² sec⁻¹ for CO₂-N₂ and CO₂-CH₄ system, respectively, if the absorbent flow rates increase from

100-500 mL min⁻¹. Gong *et al.*³¹ reported that the flux of CO₂ increase from 1.6-1.9 mmol m⁻² sec⁻¹ if the absorbent of 1 wt% MEA and 9 wt% MDEA solution flow rate increase from 20-50 mL min⁻¹. Meanwhile, Yan *et al.*¹⁶ reported that the flux of CO₂ increase from 2.2-3.0 mol m⁻² h⁻¹ if the absorbent of 1 M Potassium Glycinate (PG) velocity increase from 0.21-0.56 m sec⁻¹.

Figure 3 also shows that the flux of CO_2 through the membrane fibers for both feed gas, at the same absorbent and feed gas flow rate decrease with the increase in the number of fibers in the contactor, indicating that the effect of the absorbent flow rate more dominant than the effect of the surface area for contact. As for the overall mass transfer coefficient, the flux for CO_2 -N₂ system are higher than for CO_2 -CH₄ system due to the decrease of CO_2 concentration in the inlet gas²⁴.

Figure 4 and 5 show the effect of the absorbent flow rate on the CO₂ absorption efficiency, R (%) and the amount of CO₂ absorbed for both feed gas at the various numbers of fibers in the membrane contactors. The CO₂ absorption efficiency and the amount of CO₂ absorbed increase with increasing the absorbent flow rate in the membrane contactor due to increasing turbulence the absorbent boundary layer^{24,26,27,30}. The CO₂ absorption efficiency and the amount of CO₂ absorbed also increase with the increase in the number of fibers in the membrane contactor due to more contact surface area for absorption. The same results also reported by Gong *et al.*³¹ and Kim and Yang²⁸. In this study, the CO₂



Fig. 3: Effects of absorbent flow rate, Q_L, on the flux, J, at the membrane contactor consists of 1000, 2000 and 5000 fibers for CO₂-N₂ system and 2000, 4000 and 8000 fibers for CO₂-CH₄ system



Fig. 4: Effects of absorbent flow rate, Q_L, on the absorption efficiency, R (%) and amount of CO₂ absorbed at the membrane contactor consists of 1000, 2000 and 5000 fibers for CO₂-N₂ system

absorption efficiency, R (%), increases from 93.9-96.2 and 54.2-81.4% for CO_2 - N_2 and CO_2 - CH_4 system, respectively, if the absorbent flow rates increase from 100-500 mL min⁻¹. Gong *et al.*³¹ showed that the CO_2 absorption efficiency increase from 65-85% if the absorbent of 1 wt% MEA+9 wt% MDEA solution increase from 20-50 mL min⁻¹. Meanwhile, Kim and Yang²⁸ showed that the CO_2 absorption efficiency increases from 85-99% if the absorbent of 4 wt% MEA solution increase from 15-115 mL min⁻¹.

The effect of the feed gas flow rate and the number of fibers in the membrane contactor at the same absorbent flow rate on the overall mass transfer coefficients is shown in Fig. 6. Figure 6 shows that, for the CO_2 - N_2 and CO_2 - CH_4 system, the mass transfer coefficient increase with increasing the feed gas flow rate in the shell side of the membrane contactor. The increase of the feed gas flow rate can provide more CO_2 to be absorbed by the DEA solution which led to an increase in the overall mass transfer coefficients³². The same trend also



Fig. 5: Effects of absorbent flow rate, Q_L, on the absorption efficiency, R (%) and amount of CO₂ absorbed at the membrane contactor consists of 2000, 4000 and 8000 fibers for CO₂-CH₄ system



Fig. 6: Effects of feed gas flow rate, Q_G, on the overall mass transfer coefficient, K_{OVL}, at the membrane contactor consists of 1000, 2000 and 5000 fibers for CO₂-N₂ system and 2000, 4000 and 8000 fibers for CO₂-CH₄ system

reported by Wang *et al.*³² using 2M DEA solution as absorbent for CO_2 - N_2 system (20/80). In this study, the overall mass transfer coefficients, K_L , increase from 0.00009-0.00037 cm sec⁻¹ and 0.00023-0.00032 cm sec⁻¹ for CO_2 - N_2 and CO_2 - CH_4 system, respectively, if the feed gas flow rates increase from 120-260 mL min⁻¹ and from 170-340 mL min⁻¹, respectively. Meanwhile, Wang *et al.*³² reported that the overall mass transfer coefficients increase from 0.00016-0.00025 m sec⁻¹ if the feed gas velocities increase from 0.03-0.09 m sec⁻¹.

The numbers of fibers in the contactors have the same effect for both feed gas on the overall mass transfer coefficient at the same absorbent and feed gas flow rates. The increase in the number of fibers in the contactor will reduce the residence time of the gas in the membrane contactor, which led to the decrease in the overall mass transfer coefficient. On the other hand, the increase in the number of fibers in the contactor will increase the contact area for the mass transfer, which led to the increase in the overall mass transfer coefficient. Figure 5 shows that the overall mass transfer



Fig. 7: Effects of feed gas flow rate, Q_G, on the flux, J, at the membrane contactor consists of 1000, 2000 and 5000 fibers for CO₂-N₂ system and 2000, 4000 and 8000 fibers for CO₂-CH₄ system

coefficients for both feed gas decrease with increasing the number of fibers in the contactor at the same absorbent and feed gas flow rates, indicating that the effect of the feed gas flow rate more dominant than the effect of the surface area for contact.

The impact of the feed gas flow rate and the number of fibers in the membrane contactor at the same absorbent flow rate on the flux is demonstrated in Fig. 7. Similar to the overall mass transfer coefficient, the flux increase with increasing the feed gas flow rate in the shell side of the membrane contactor due to more CO₂ that can be absorbed by the absorbent solution^{16,32}. Similar result was also shown by Keshavarz et al.¹² and Zhang et al.³³. In this study, the flux of CO₂, J, increase from 2.7×10^{-6} to $8.8\!\times\!10^{-6}$ mmol cm^{-2} sec^{-1} and $2.6\!\times\!10^{-6}$ to 3.9×10^{-6} mmol cm⁻² sec⁻¹ for CO₂-N₂ and CO₂-CH₄ system, respectively, if the feed gas flow rates increase 120-260 and 170-340 mL min⁻¹, respectively. Meanwhile, Keshavarz et al.¹² and Zhang et al.³³ demonstrated that the flux of CO₂ increase from 0.0003-0.0008 mol m^{-2} sec⁻¹ if the feed gas flow rate increase from 0.03-0.09 m sec⁻¹ for CO₂-N₂ system (20/80) using 2 M DEA solution as absorbent. Figure 7 also shows that the flux for both feed gas decrease with increasing the number of fibers in the contactor, at the same absorbent and feed gas flow rates, indicating that the effect of the feed gas flow rate more dominant than the effect of the surface area for contact.

Figure 8 and 9 show the effect of the feed gas flow rate on the CO₂ absorption efficiency, R (%) and amount of CO₂ absorbed at the various numbers of fibers in the membrane contactors. The amount of CO₂ absorbed for both feed gas increase with increasing the feed gas flow rate in the membrane contactor due to increasing turbulence the absorbents boundary layer^{24,26,27,30} as shown in Fig. 8 and 9. The amount of CO₂ absorbed also increase with the increase in the number of fibers in the membrane contactor due to more contact surface area for absorption. The same results also reported by Yan et al.¹⁶ using 1 M PG or 1 M MEA as absorbents. However, both feed gas system show the different phenomena for the CO₂ absorption efficiency, R (%). The CO₂ absorption efficiency increases with increasing feed gas flow rate for CO2-N2 system and decreases with increasing feed gas flow rate for CO₂-CH₄ system. The decrease in the CO₂ absorption efficiency with increasing gas flow rate is also shown by Yan et al.¹⁶. In this study, the CO₂ absorption efficiency for CO₂-CH₄ system decrease from 72.7-64.0% if the gas flow rate increase from 170-340 mL min⁻¹ using 5% DEA solution as absorbent. Meanwhile, Yan et al.¹⁶ showed that the CO₂ absorption efficiency for CO₂-N₂-O₂ system decrease from 94-65 and 89-54% if the gas velocity increase from 0.21-0.56 m sec⁻¹ using 1 M PEG and 1 M MEA as absorbents, respectively.



Fig. 8: Effects of feed gas flow rate, Q_G, on the absorption efficiency, R (%) and amount of CO₂ absorbed at the membrane contactor consists of 1000, 2000 and 5000 fibers for CO₂-N₂ system



Fig. 9: Effects of feed gas flow rate, Q_G, on the absorption efficiency, R (%) and amount of CO₂ absorbed at the membrane contactor consists of 2000, 4000 and 8000 fibers for CO₂-CH₄ system

CONCLUSION

High hydrophobicity is one of the important properties of the membrane for gas-liquid contact process, which is required to avoid pores membrane wetting so that can increase the overall mass transfer coefficient. This study utilized polypropylene super hydrophobic hollow fiber membrane contactor to absorb CO_2 from its mixture with N_2 or CH_4 using DEA solution (5% vol in water) as absorbent. In the experiments, the absorbent flowed through the lumen of fibers, while the feed gas flowed through the shell side of the contactor. The experimental results showed that the mass transfer coefficients, the fluxes and the amount of CO_2 absorbed increased with increasing the absorbent and feed gas flow rates. The CO_2 absorption efficiency for both feed gas increased with increasing the absorbent flow rate. However, the increase in the feed gas flow rate gives different effect to the CO_2 absorption efficiency for CO_2 - CH_4 system.

The increased in the feed gas flow rate will increase the CO₂ absorption efficiency for CO₂-N₂ system, but will decrease the CO₂ absorption efficiency for CO₂-CH₄ system. The numbers of fibers in the contactors gave the same effects for both feed gas system in terms of the overall mass transfer coefficient, the flux, the amount of CO₂ absorbed and the CO₂ absorption efficiency. The overall mass transfer coefficient and the flux decreased with the increase in the number of fibers in the contactor, whilst, the amount of CO₂ absorbed and the CO₂ absorption efficiency increased.

ACKNOWLEDGMENT

The authors acknowledge financial supports for this work from the Universitas Indonesia through contract No. 1178/UN2.R12/HKP.05.00/2016.

REFERENCES

- Rahbari-Sisakht, M., A.F. Ismail, D. Rana and T. Matsuura, 2012. A novel surface modified polyvinylidene fluoride hollow fiber membrane contactor for CO₂ absorption. J. Membrane Sci., 415-416: 221-228.
- Mandal, B.P., A.K. Biswas and S.S. Bandyopadhyay, 2003. Absorption of carbon dioxide into aqueous blends of 2-amino-2-methyl-1-propanol and diethanolamine. Chem. Engin. Sci., 58: 4137-4144.
- Zhao, G., B. Aziz and N. Hedin, 2010. Carbon dioxide adsorption on mesoporous silica surfaces containing amine-like motifs. Applied Energy, 87: 2907-2913.
- Zhao, L., E. Riensche, R. Menzer, L. Blum and D. Stolten, 2008. A parametric study of CO₂/N₂ gas separation membrane processes for post-combustion capture. J. Membrane Sci., 325: 284-294.
- Lv, Y., X. Yu, J. Jia, S.T. Tu, J. Yan and E. Dahlquist, 2012. Fabrication and characterization of superhydrophobic polypropylene hollow fiber membranes for carbon dioxide absorption. Applied Energy, 90: 167-174.
- Kartohardjono, S., G.M. Damaiati and C.T. Rama, 2015. Effects of absorbents on ammonia removal from wastewater through hollow fiber membrane contactor. J. Environ. Sci. Technol., 8: 225-231.
- Kartohardjono, S., R. Darmawan, M.F. Karyadi and N. Saksono, 2016. CO₂ absorption through super-hydrophobic hollow fiber membrane contactors. J. Environ. Sci. Technol., 9: 214-219.
- Saidi, M., 2017. Mathematical modeling of CO₂ absorption into novel reactive DEAB solution in hollow fiber membrane contactors; kinetic and mass transfer investigation. J. Membrane Sci., 524: 186-196.

- Hosseinzadeh, A., M. Hosseinzadeh, A. Vatani and T. Mohammadi, 2017. Mathematical modeling for the simultaneous absorption of CO₂ and SO₂ using MEA in hollow fiber membrane contactors. Chem. Eng. Process.: Process Intensification, 111: 35-45.
- Yu, X., L. An, J. Yang, S.T. Tu and J. Yan, 2015. CO₂ capture using a superhydrophobic ceramic membrane contactor. J. Membr. Sci., 496: 1-12.
- 11. Lu, J.G., Y.F. Zheng and M.D. Cheng, 2008. Wetting mechanism in mass transfer process of hydrophobic membrane gas absorption. J. Membr. Sci., 308: 180-190.
- Keshavarz, P., J. Fathikalajahi and S. Ayatollahi, 2008. Analysis of CO₂ separation and simulation of a partially wetted hollow fiber membrane contactor. J. Hazard. Mater., 152: 1237-1247.
- Zhang, H.Y. R. Wang, D.T. Liang and J.H. Tay, 2008. Theoretical and experimental studies of membrane wetting in the membrane gas-liquid contacting process for CO₂ absorption. J. Membr. Sci., 308: 162-170.
- Yong, J.K.J., G.W. Stevens, F. Caruso and S.E. Kentish, 2016. *In situ* layer-by-layer assembled carbonic anhydrase-coated hollow fiber membrane contactor for rapid CO₂ absorption. J. Membrane Sci., 514: 556-565.
- Rongwong, W., R. Jiraratananon and S. Atchariyawut, 2009. Experimental study on membrane wetting in gas-liquid membrane contacting process for CO₂ absorption by single and mixed absorbents. Sep. Purif. Technol., 69: 118-125.
- 16. Yan, S.P., M.X. Fang, W.F. Zhang, S.Y. Wang, Z.K. Xu, Z.Y. Luo and K.F. Cen, 2007. Experimental study on the separation of CO₂ from flue gas using hollow fiber membrane contactors without wetting. Fuel Process. Technol., 88: 501-511.
- 17. Mansourizadeh, A., 2012. Experimental study of CO₂ absorption/stripping via PVDF hollow fiber membrane contactor. Chem. Eng. Res. Des., 90: 555-562.
- Goyal, N., S. Suman and S.K. Gupta, 2015. Mathematical modeling of CO₂ separation from gaseous-mixture using a Hollow-Fiber Membrane Module: Physical mechanism and influence of partial-wetting. J. Membr. Sci., 474: 64-82.
- Lin, S.H., P.C. Chiang, C.F. Hsieh, M.H. Li and K.L. Tung, 2008. Absorption of carbon dioxide by the absorbent composed of piperazine and 2-amino-2-methyl-1-propanol in PVDF membrane contactor. J. Chinese Inst. Chem. Eng., 39: 13-21.
- Fashandi, H., A. Ghodsi, R. Saghafi and M. Zarrebini, 2016. CO₂ absorption using gas-liquid membrane contactors made of highly porous poly(vinyl chloride) hollow fiber membranes. Int. J. Greenhouse Gas Control, 52: 13-23.
- Wu, X., B. Zhao, L. Wang, Z. Zhang, H. Zhang, X. Zhao and X. Guo, 2016. Hydrophobic PVDF/graphene hybrid membrane for CO₂ absorption in membrane contactor. J. Membrane Sci., 520: 120-129.
- 22. Lu, K.J., J. Zuo and T.S. Chung, 2016. Tri-bore PVDF hollow fibers with a super-hydrophobic coating for membrane distillation. J. Membrane Sci., 514: 165-175.

- Choi, W., G. Kim and K. Lee, 2012. Influence of the CO₂ absorbent monoethanolamine on growth and carbon fixation by the green alga *Scenedesmus* sp. Bioresour. Technol., 120: 295-299.
- Wang, D., W.K. Teo and K. Li, 2004. Selective removal of trace H₂S from gas streams containing CO₂ using hollow fibre membrane modules/contractors. Sep. Purif. Technol., 35: 125-131.
- 25. Wang, R., D.F. Li and D.T. Liang, 2004. Modeling of CO₂ capture by three typical amine solutions in hollow fiber membrane contactors. Chem. Eng. Process., 43: 849-856.
- Dindore, V.Y., D.W.F. Brilman, P.H.M. Feron and G.F. Versteeg, 2004. CO₂ absorption at elevated pressures using a hollow fiber membrane contactor. J. Membr. Sci., 235: 99-109.
- Franco, J., D. de Montigny, S. Kentish, J. Perera and G. Stevens, 2008. A study of the mass transfer of CO₂ through different membrane materials in the membrane gas absorption process Sep. Sci. Technol., 43: 225-244.

- 28. Kim, Y.S. and S.M. Yang, 2000. Absorption of carbon dioxide through hollow fiber membranes using various aqueous absorbents. Sep. Purif. Technol., 21: 101-109.
- 29. Mansourizadeh, A. and A.F. Ismail, 2009. Hollow fiber gas-liquid membrane contactors for acid gas capture: A review. J. Hazard. Mater., 171: 38-53.
- Wu, J. and V. Chen, 2000. Shell-side mass transfer performance of randomly packed hollow fiber modules. J. Membrane Sci., 172: 59-74.
- 31. Gong, Y., Z. Wang and S. Wang, 2006. Experiments and simulation of CO_2 removal by mixed amines in a hollow fiber membrane module. Chem. Eng. Process., 45: 652-660.
- 32. Wang, R., H.Y. Zhang, P.H.M. Feron and D.T. Liang, 2005. Influence of membrane wetting on CO₂ capture in microporous hollow fiber membrane contactors. Sep. Purif. Technol., 46: 33-40.
- Zhang, H.Y., R. Wang, D.T. Liang and J.H. Tay, 2006. Modeling and experimental study of CO₂ absorption in a hollow fiber membrane contactor. J. Membrane Sci., 279: 301-310.