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## Protochlorophyllide Spectral Forms

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**Abstract:** This study summarized recent results on POR and plastid development in order to find an explanation for the existence of Pchlde spectral forms. This review has summarized many researches about the Pchlde spectral forms and their phototransformability in different higher plants which have been published. Chlorophyll (Chl) is the most important pigment on the Earth. Each spring millions of tons Chl are formed during bud break and leaf development. The sun light needed for photosynthesis is captured by Chl and transformed to chemical energy. In the Biosynthesis of chlorophyll (Chl) begins with the synthesis of  $\delta$ -aminolevulinic acid (ALA) from glutamic acid. Chl biosynthetic pathway, a light dependent enzyme protochlorophyllide oxidoreductase (POR) catalyses a key light-driven reaction, trans addition of hydrogen across the C-17-C-18 double bond of the Chl precursor, protochlorophyllide (Pchlde), that triggers a profound transformation in plant development. Pchlde is spectrally heterogeneous and exist in different spectral forms having slightly different absorption and fluorescence peaks. The identified Pchlde spectral forms can be sorted into three groups. The first group is designated as short-wavelength forms. This group with fluorescence in the 625-646 nm spectral region has a heterogeneous nature and is made by four components. The second group includes the long-wavelength Pchlde forms with emission maxima between 652 and 657 nm. The third group is found in the extreme red region (670-730 nm) of the fluorescence emission spectra and includes a number of pigment forms with spectral bands of low intensity. This region is also influenced by vibrational bands of the short-and long-wavelength Pchlde forms.

**Key words:** Chlorophyll, fluorescence excitation, fluorescence emission, fluorescence lifetime, NADPH-protochlorophyllide oxidoreductase, protochlorophyllide

### INTRODUCTION

Biosynthesis of chlorophyll (Chl) begins with the synthesis of  $\delta$ -aminolevulinic acid (ALA) from glutamic acid. Two molecules of ALA react to yield porphobilinogen. Four molecules of porphobilinogen form the ring structure of protoporphyrin IX (proto). To make Chl, a magnesium chelatase inserts Mg into proto ring. Chl synthesis involves further modifications of the ring, including attachment of the phytol chain. In angiosperms, the synthesis of Chl requires a light-dependent Pchlde reductase enzyme (Buchanan *et al.*, 2000).

In the dark grown angiosperms, the light-dependent synthesis of Chl is arrested, which in turn, blocks the thylakoid formation and results in the formation of etioplasts (Staehelein, 2003). Etioplasts represent an arrested stage in the normal development of proplastids into chloroplasts. The most notable feature of the etioplasts is the presence of one or more prolamellar bodies, PLBs. The lack of functional thylakoids, the absence of Chl, the accumulation of monovinyl (MV) and

divinyl (DV) derivatives of Pchlde and the presence of POR are some other features of etioplasts (Whatley *et al.*, 1982; Ryberg and Sundqvist, 1991; Ryberg *et al.*, 1993).

A complex signalling network that combines environmental and genetic information coordinates the activities of the nucleus and the chloroplast. Necessary for this is a transduction of information from plastids to the nucleus as well as the flux of information from the nucleus to the plastids (Strand *et al.*, 2003). Such a signal is required continuously for the expression of nuclear photosynthetic genes (Sullivan and Gray, 1999). It has been demonstrated that the accumulation of Mg-proto IX is necessary and sufficient to regulate the expression of many nuclear encoded chloroplastic proteins (Strand *et al.*, 2003). The transmission of a plastid signal does not obligatorily require light but can occur in the dark and both in leaves and roots (Sullivan and Gray, 1999). A large number of proteins localised to the chloroplast are actually encoded in the nucleus.

Chl biosynthesis starts from the formation of ALA a common tetrapyrrole precursor. The ALA biosynthesis is regulated by the levels of POR and Pchlde or by heme

(Ilag *et al.*, 1994). The subsequent energy-independent reactions of ALA lead to proto. Energy-dependent incorporation of magnesium into Proto gives rise to the specific Chl series (Beale, 1999). A light governed reduction of Pchlride results in formation of chlorophyllide (Chlide). Pchlride reduction in the dark and under irradiation revealed two different types of enzymes to which can be referred to as a light-independent Pchlride reductase (DPOR) and a light-dependent Pchlride reductase (LPOR), which have been reviewed by Timko (1998) and Schoefs (2005). Angiosperms are missing the light-independent enzymes. LPOR in *Arabidopsis thaliana* has three isoforms, PORA, PORB and PORC. However, some angiosperms such as cucumber have only one isoform of POR (Kuroda *et al.*, 1995).

Conformational change of POR within sub-nano seconds switches the enzyme into an active state. Sytina *et al.* (2008) reported that prior to excitation of the enzyme-substrate complex with a laser pulse induces a more favourable conformation of the active site than non-active state. This effect is triggered during the Pchlride excited state lifetime and persists on a long timescale (Sytina *et al.*, 2008).

Chl is associated with proteins localized to the thylakoids in the chloroplast. The Chls are present in different spectral forms in the Chl-protein complexes. They are organized in light-trapping antennas in such a way that the pigments absorbing light of relatively short wavelengths are found at the periphery of the antenna whereas those absorbing at longer wavelength are generally found closer to the reaction centre (Grondelle *et al.*, 1994). Light energy is thus transferred from the antenna and trapped in the reaction centre.

Spectral forms of Pchlride, as a pigment, characterized by its absorbance and excitation and emission fluorescence. Initially two forms, one phototransformable form absorbing around 650 nm and one non-transformable form absorbing around 635 nm were identified. Pchlride is bound to the specific site of the POR protein and might also be present in a free form but the reason for the presence of different forms are still obscure. This review summarizes the knowledge about different spectral Pchlride forms and attempt to find an explanation for their existence.

**The prolamellar body:** The etioplast inner membranes are organized into two structurally different systems. One is the PLBs, with tubular membranes connected into a highly regular three-dimensional lattice of which the membranes are physically continuous with the other system the prothylakoids, PTs, flat perforated membranes that extend from the surface of the PLBs. The PLBs are composed of

regular tetrapodal units made up from a continuous bilayer assembled to form a diamond cubic lattice. The POR enzyme is the main protein in the PLBs and constitutes nearly 90% of its protein content. In the PLBs, POR in ternary complex with Pchlride and NADPH forms a large aggregate of photoactive Pchlride form (Böddi *et al.*, 1990; Schulz and Senger, 1993). At the onset of greening the PLBs can be of importance for efficient capture of light energy for optimal photoconversion of Pchlride to Chlide (Masuda *et al.*, 2003). The PLBs are also considered as a storage room for membrane lipids and in fact it was shown that many photosynthetic proteins are present already in the dark in the PLBs. The flat perforated PT membrane stretching out into the stroma may act as a precursor of the thylakoids.

The PLB formation is correlated to the presence of POR. The overexpression of POR allowed the creation of large PLB membranes indicating that the formation of PLBs is correlated to the aggregation of ternary complexes of NADPH-POR-Pchlride. Also Masuda showed that POR functions in PLB assembly (Masuda *et al.*, 2003). However, an intact PLB structure is not a prerequisite for the preservation of the spectral properties of the ternary complex of POR. High salt concentrations caused a disintegration of the structure of isolated PLBs, but had no effect on the spectral properties of the ternary complex of POR, especially in the presence of NADPH (Selstam *et al.*, 2007).

Irradiation induces a series of changes in chemical, ultra structural and spectral properties of etioplasts. Pchlride is photoconverted to Chlide and then esterified to Chl. After irradiation the PLBs lose their regular structures and the perforated PTs increase in length. The PLBs are then dispersed and disappear completely within a few hours and are replaced by newly formed thylakoids (Henningsen, 1970). The photoreduction of Pchlride to Chlide not only triggers the transformation of the PLBs but the formation of the first Chl molecules initiate the synthesis of some and stabilize the accumulation of other chloroplast proteins.

**POR:** POR catalyzes trans addition of hydrogen across the C-17-C-18 double bond of Pchlride. Pchlride reduction occurs by dynamically coupled nuclear quantum tunneling of a hydride anion followed by a proton on the microsecond time scale in the Pchlride excited and ground states, respectively (Heyes *et al.*, 2009). POR is a nucleus-encoded enzyme, which has been cloned and sequenced from different plants has been identified in *Arabidopsis* and its gene is cloned and sequenced (Oosawa *et al.*, 2000; Su *et al.*, 2001). Different POR genes have been reported in various angiosperms. *Arabidopsis*

has a small gene family consisting of PorA, PorB and PorC. PORA and PORB, two structurally similar but differentially regulated isoforms of POR were first identified in barley (Holtorf *et al.*, 1995) and *Arabidopsis* (Armstrong *et al.*, 1995) but recently also in wheat. The third isozyme, PORC, characterized in *Arabidopsis*, is not present in darkness but is induced by high light irradiation and was suggested to also have a photoprotective role during greening (Oosawa *et al.*, 2000; Masuda *et al.*, 2003). PORA is known to be down-regulated by phytochrome (Apel, 1981). Far-red light working through phytochrome A can thus cause a down-regulation of PORA and an inhibition in greening. However, far-red stimulated Pchl<sub>id</sub> and Chl formation has also been known for a long time (Klockare, 1980). Sineshchekov found that the phytochrome regulation of the Pchl<sub>id</sub> level was dependent on plant species and development in a complex pattern (Sineshchekov *et al.*, 2006).

The chloroplast DNA of *Arabidopsis thaliana* is a circular molecule of 153 kb. However, estimations show that the chloroplast hosts more than 2000 (*Arabidopsis* 2100, rice 4500) different proteins (Lopez-Juez and Pyke, 2005). More than 90% of them are encoded by nuclear genes. Import of nucleus-encoded proteins into the plastid has been shown to occur mainly by a general import pathway (Chen and Schnell, 1999). The precursor form of the nuclear encoded POR protein, pPOR, consists of approximately 400 amino acids and has a molecular weight of about 41 to 44 kDa. The mature form has a molecular weight of about 33 to 38 kDa. The pPOR follows the general import pathway (Aronsson *et al.*, 2001). Pea pPOR could be cross-linked to pea Toc75, a major protein in the translocation complex indicating that POR uses the general import pathway (Aronsson *et al.*, 2000). Teakle and Griffiths (1993) showed that the pPOR of 41 kDa was imported into isolated wheat chloroplasts and that stromal-processing proteases cleaved off the transit peptide, giving POR its mature size of approximately 36 kDa (Teakle and Griffiths, 1993).

The amino acid sequences of POR from barley (PORA), oat, *Arabidopsis* (PORA) and pea have large similarities (Schulz and Senger, 1993). The monocotyledons, oat and barley, displayed 96% similarity and compared to the dicotyledons, *Arabidopsis* and pea, the barley sequence revealed similarities of 81 and 83%, respectively. The sequences are characterised by high contents of basic and hydrophobic amino acids and contain highly conserved regions, for instance in the parts where the cysteines are located (Schulz and Senger, 1993).

In *Arabidopsis*, the sequence similarities for the precursor POR protein were 87% between PORA and PORB, 74% between PORC and PORA and 76% between PORC and PORB. In the mature form the sequences have a higher similarity 83% between PORC and PORA or PORB leaving the main sequence diversity to be in the transit peptide (Oosawa *et al.*, 2000). In barley, the precursors of PORA and PORB share 75% and the mature forms 81.5% sequence similarities, whereas the transit peptides have only 46% sequence similarity (Holtorf *et al.*, 1995).

A study of *Synechocystis* POR showed that the ternary complex initially is formed by the binding of NADPH to POR and thereafter Pchl<sub>id</sub> (Heyes *et al.*, 2000). The presence or absence of Pchl<sub>id</sub> did not influence the binding parameters of NADPH to POR. This indicated either independent binding sites for NADPH and Pchl<sub>id</sub> or an regulated binding of the co-factor and the substrate. Pchl<sub>id</sub> could bind to POR in the absence of NADPH. When NADPH binds to POR-Pchl<sub>id</sub> it is suggested to induce a conformational change of the complex leading to protection of sensitive cysteine residues (Heyes *et al.*, 2000).

The ternary complex is present as a dimer but can also form larger complexes. The dimer interface was predicted to correspond to residues 198-208 and 271-292, based on results from X-ray structural analysis of other members of the reductases/epimerases/dehydrogenases, RED family, (Jörnvall *et al.*, 1995). Analysis of POR from the etioplast inner membranes of wheat by two-dimensional electrophoresis showed five different isoforms of PORA. Four with a pI around 8-9 and one with an pI more close to five (Blomqvist *et al.*, 2008) indicating a certain degree of posttranslational modification.

**Pchl<sub>id</sub> and Chl formation:** Pchl<sub>id</sub> as an intermediate plays an important role in the regulation of Chl biosynthesis. Pchl<sub>id</sub> accumulates in dark-grown angiosperms and is mainly localized to the PLBs (Sundqvist and Dahlin, 1997; Beale, 1999). The Pchl<sub>id</sub> content of isolated PLBs are more than ten times of that of PTs. The protochlorophyll (Pchl), which esterified form of Pchl<sub>id</sub> and Pchl<sub>id</sub> content of leaves increases rapidly up to 7 to 9 days and then the rate of accumulation slows down (Klein and Schiff 1972). Cotyledons contain Pchl and as leaf development proceeds in the dark, Pchl<sub>id</sub> accumulation increases while the formation of Pchl is attenuated and eventually ceases before the maximum in total pigment accumulation is achieved (Lancer *et al.*, 1976).

*In vivo*, Pchl<sub>ide</sub> exists in four closely related species: MV- and DV-Pchl<sub>ide</sub> or MV- and DV-Pchl, Pchl. However, Pchl<sub>ide</sub> can also exist in two other types with different chemical structure, i.e., Pchl<sub>ide</sub> a and b. (Reinbothe *et al.*, 1999). The occurrence of Pchl<sub>ide</sub> b, however, has been proven not to be present (Kolossof and Rebeiz, 2003).

The proportion of DV-Pchl<sub>ide</sub> in the non-photoactive and photoactive Pchl<sub>ide</sub> pools increases in correlation to the POR content. The MV and DV Pchl<sub>ide</sub> function equally well as POR substrates, although minor spectroscopic differences can be traced in the intermediates (Heyes *et al.*, 2006).

Photoreduction of Pchl<sub>ide</sub> is a major regulatory step in the Chl biosynthesis. Pchl<sub>ide</sub> reduction catalyzed by POR is an ultra fast enzymatic reaction occurring within femto seconds. In the reaction the photoactive Pchl<sub>ide</sub> is transformed to Chl<sub>ide</sub> via a light driven step and two dark reactions (Wilks and Timko, 1995). However, Heyes and Hunter using a thermophilic form of POR identified two additional dark steps (Heyes and Hunter, 2004). Photochemistry and the first dark step result in formation of the POR-Chl<sub>ide</sub>-NADP<sup>+</sup> complex. The second dark reaction and two additional dark steps identified by Heyes and Hunter have been shown to represent a series of ordered product-release and cofactor-binding events: NADP<sup>+</sup> is released from the enzyme and then replaced by NADPH, before release of the Chl<sub>ide</sub> product and subsequent binding of the Pchl<sub>ide</sub> substrate to enable the next catalytic cycle to proceed (Heyes and Hunter, 2004, 2005). During this process two hydrogen atoms which are derived from the NADPH and Tyr 275 of the POR protein, are added at carbons 17 and 18 of Pchl<sub>ide</sub>, respectively (Wilks and Timko, 1995).

In the photoreduction reaction the proper substrate for POR is the excited Pchl<sub>ide</sub> not the pigment in the ground state (Fujita, 1996). The catalytic mechanism of POR involves two additional steps, which do not require light. The first involves the conversion of the product of the initial light-driven reaction, a non-fluorescent radical species, into a new intermediate that has an absorbance maximum at 681 nm and a fluorescence peak at 684 nm. During the second dark step the absorption and the emission bands gradually blue shift to yield the product, Chl<sub>ide</sub>. Another new pathway of Chl biosynthesis from long-wavelength Pchl<sub>ide</sub> fluorescing at 686 nm is also suggested in which a photo transformation of Pchl<sub>ide</sub> fluorescing at 686 nm into Pchl<sub>ide</sub> fluorescing at 653 nm, which is then photo transformed to Chl<sub>ide</sub> (Ignatov and Litvin, 2002).

The first recognizable product of photoreduction at room temperature is a transient Chl<sub>ide</sub> with absorption and emission maxima at 678 and 690 nm, respectively,

which then converts to a Chl<sub>ide</sub> form absorbing at 684 nm. This form in turn, has a blue shift, called Shibata shift, to a 672 nm absorbing form. These spectral shifts occur also during Chl accumulation in plants grown in photoperiodic light (Schoefs and Franck, 2008). The shifts seem to disappear during maturation of leaves and might be coupled to the regulation of development (Lee *et al.*, 2007). The shift has been interpreted as a liberation of the pigment from the enzyme and the Chl<sub>ide</sub> absorbing at 672 nm has been regarded as a free pigment. Several other proposals have been suggested as explanations for the different Chl<sub>ide</sub> spectral forms and their interconversions such as pigment aggregation, esterification, association or disassociation with a specific type of membrane or enzyme (Franck *et al.*, 1997). The spectral shifts coincide with the transformation of the PLB structure and a re-localization of POR from the PLBs to the developing thylakoids. Most of the newly formed Chl<sub>ide</sub> is esterified to Chl a in parallel to or slightly slower than the Shibata shift. A Pchl<sub>ide</sub> esterified with phytol already in dark-grown plants was found to be important as a structural component substituting the Chl in the cytochrome b<sub>6</sub>f complex. In this case the phytol moiety of the pigment was the important part (Reisinger *et al.*, 2008).

**Spectral properties of Pchl<sub>ide</sub> in dark-grown plants:**

Pchl<sub>ide</sub> is spectrally heterogeneous. Several spectral forms of Pchl<sub>ide</sub> are present in dark-grown plants (Böddi *et al.*, 1992; Stadnichuk *et al.*, 2005). The proportions of spectral Pchl<sub>ide</sub> forms varies depending on plant species (Fig. 1A, B), plant tissue (Fig. 2) and

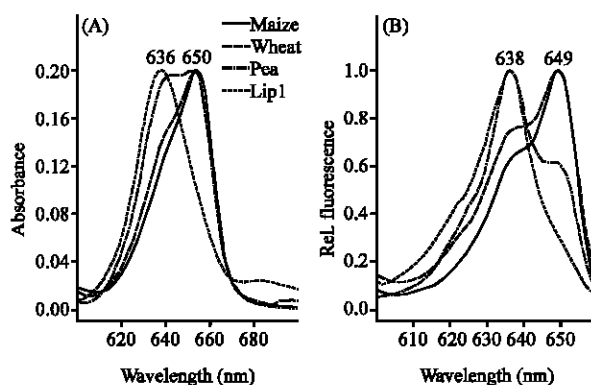


Fig. 1: Room temperature absorption (A) and low temperature (77 K) excitation (B) spectra of 7-day-old dark-grown maize, wheat, pea and lip1 showing the species dependency of spectral forms. Excitation spectra were measured by the emission set at 705 nm. The spectra were normalized at the highest peak

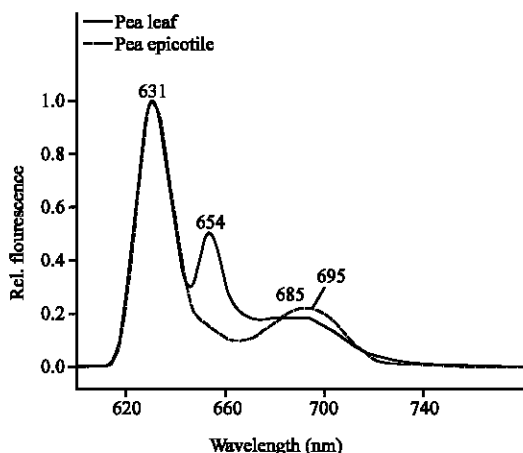


Fig. 2: Low temperature (77 K) fluorescence emission spectra of dark-grown pea leaves and epicotyls. The spectra are normalized at the highest peak. Excitation wavelength was 440 nm

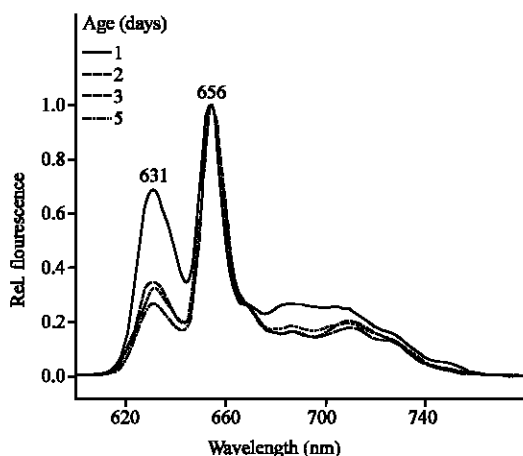


Fig. 3: Low temperature (77 K) fluorescence emission spectra of dark-grown wheat at various ages. The spectra are normalized at the highest peak. The plant age, calculated from the start of imbibition, is indicated. Excitation wavelength was 440 nm

developmental age of the plant (Fig. 3) (Mysliwa-Kurdziel *et al.*, 2003; Amirjani *et al.*, 2006). Marchand found that plastids in the bundle sheath cells of maize accumulated preferentially the short-wavelength form of Pchl<sub>id</sub> (Marchand *et al.*, 2004). In green and greening leaves of maize Pchl<sub>id</sub> accumulated after a period of dark treatment also participates in the formation of different spectral forms (Amirjani and Sundqvist, 2004). The heterogeneous spectroscopic properties of the various Pchl<sub>id</sub> forms reflect differences in their environments. Some contributing factors are the

association of Pchl<sub>id</sub> to the POR protein and NADPH to make a ternary complex, a possible POR phosphorylation and an aggregation of ternary complexes. Short-wavelength form of Pchl<sub>id</sub> is presumably localized in lamellar membranes of PTs and/or small and loose PLBs unlike the longer wavelength Pchl<sub>id</sub> complexes (Böddi *et al.*, 1994). Such changes in localization can be of basic importance. Short-wavelength forms of Pchl<sub>id</sub> can come together, thus the  $\delta$ -electron systems of the Pchl<sub>id</sub> molecules in their active sites interact and the exciton interaction causes the red shift in their emission maxima to 644 or 655 nm (Kosa *et al.*, 2006). The appearance of different Pchl<sub>id</sub> forms can also be due to a change in the oxidative state of the cofactor, NADP (Franck *et al.*, 1999). The lipid composition surrounding the enzyme is also of importance for the spectral properties of POR bound Pchl<sub>id</sub> (Klement *et al.*, 2000). Recent measurements on the chromophores Pchl and Pchl<sub>id</sub> in different solvents indicate that spectral properties such as the Stokes shifts and the fluorescence lifetimes are affected by the phytol chain (Mysliwa-Kurdziel *et al.*, 2008). To characterize the spectral forms of Pchl<sub>id</sub> the absorption, fluorescence excitation and emission spectra have been used.

**Absorption and fluorescence excitation:** Absorption spectrum of Pchl<sub>id</sub> in organic solutions shows two strong bands, a Soret band located around 440 nm and a long-wavelength band located around 625 nm. The peak positions, however, vary in different solutions as the latter band has a peak position at 623 or 626 nm in 80% acetone or in methanol, respectively (Hendrich and Bereza, 1993). The *in vivo* absorption spectrum of Pchl<sub>id</sub> at both room and low temperature (77 K) has a peak at 650 nm and a shoulder at 636 nm, the proportion of which depends on the plant species. For example the 636-nm shoulder of the 7-day-old maize is slightly smaller than that of 7-day-old wheat and for pea the short- and long-wavelength peaks have approximately the same height. In contrast, in *lip1* mutant of pea, which shows the morphology of light grown plants when grown in dark, the 636-nm band is dominating (Fig. 1). The ratio of the mentioned absorption peaks changes during growth of the plants due to an initial increase of the 650 nm peak (Amirjani *et al.*, 2006) followed by a decrease during senescence.

The 77 K fluorescence excitation spectra show similar differences as absorption spectra which varies with plant species and different developmental stages. For instance it has been reported that excitation spectra of wheat have a dominant peak at 649 nm when the emission wavelength was set at 710 nm (Fig. 4). The variations are not restricted to plant species but are found also between different parts of a given plant, e.g., epicotyls and leaves.

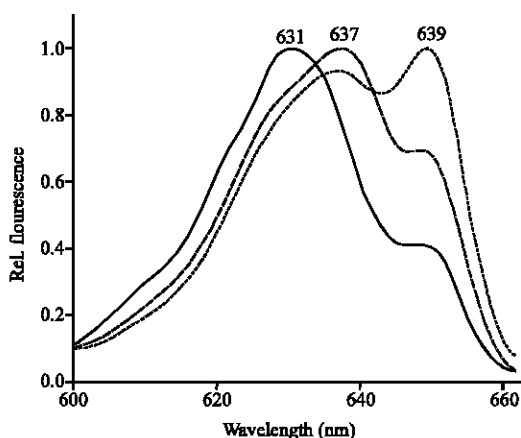


Fig. 4: Low temperature (77 K) excitation spectra of ALA treated maize leaves. The spectra are recorded at 695 (—), 710 (---) and 725 (- - -) nm. The spectrum of control (—) is recorded at 710 nm. The spectra were normalized at the highest peak

In the excitation spectra of pea epicotyls the 672 nm peak is most pronounced (Stadnichuk *et al.*, 2005). Furthermore, at an early developmental age the excitation spectra also changed in the Pchl<sub>ide</sub> region. In 1-day-old pea, only a small band is seen at 649 nm but one day later a 636-nm shoulder appeared which became the dominant peak when the plants were 5 days old (Amirjani *et al.*, 2006).

The heterogeneous nature of Pchl<sub>ide</sub> may best be shown by a series of excitation spectra recorded at an increasing emission wavelength from ALA treated dark-grown maize leaves (Fig. 4). The excitation spectra recorded with the emission set at 695 nm contained contributions from four Pchl<sub>ide</sub> forms emitting at 628, 635, 644 and 656 nm (Stadnichuk *et al.*, 2005). With increasing emission wavelength the dominant band gradually shifted to the red side of the spectra. When measured at 710 nm the vibrational band of the 656 nm Pchl<sub>ide</sub> form was dominating and the excitation peak at 649 nm was evident. With a stepwise increase in the emission wavelength the peak position remained at 649 nm. But the short-wavelength bands were also found in the excitation spectra. This fact can be ascribed to an energy transfer from short-to long-wavelength Pchl<sub>ide</sub> forms.

**Fluorescence emission:** During fluorescence emission the electronic transition occurs from the lowest energy level, Q<sub>y</sub> (0,0) and from its vibrational sublevel, Q<sub>y</sub> (0,1) (Gouterman, 1978).

Böddi *et al.* (1992) reported four spectral forms of Pchl<sub>ide</sub> with emission maxima at 633, 645, 657 and 670 nm. Later experiments with bean leaves indicate that the

633 nm band of non-photoactive Pchl<sub>ide</sub> is made of four bands at 625, 631, 637 and 643 nm. Thus at least seven Pchl<sub>ide</sub> species are present in bean leaves (Schoefs *et al.*, 2000).

The identified Pchl<sub>ide</sub> spectral forms can be sorted into three groups. The first group is designated as short-wavelength forms with fluorescence in the 625-646 nm spectral region with corresponding excitation bands in the 620-640 nm region. This group has a heterogeneous nature and is made by four components fluorescing at around 628, 635, 642 and 644 nm (Table 1) (Stadnichuk *et al.*, 2005). The 628 and 635 nm forms are both monomeric and the 628 nm form is suggested to be unbound Pchl<sub>ide</sub> while the 635 nm form represents a Pchl<sub>ide</sub> bound to the POR protein. The second group includes the long-wavelength Pchl<sub>ide</sub> forms with emission maxima between 652 and 657 nm. A Pchl<sub>ide</sub> form belonging to this group with peak position at 653 nm was found shortly after flash irradiation of dark-grown leaves. The presence of NADP<sup>+</sup> can contribute to the formation of this form. The third group is found in the extreme red region (670-730 nm) (Table 1) of the fluorescence emission spectra and includes a number of pigment forms with spectral bands of low intensity. This region is also influenced by vibrational bands of the short- and long-wavelength Pchl<sub>ide</sub> forms. Sironval has claimed that the region of 660-725 in addition to the vibronic bands also contains bands coming from electronic transitions (Sironval *et al.*, 1967). The main short-wavelength bands between 628-635 nm have vibronic bands in the region 685-693 nm. The long-wavelength bands have vibronic satellites above this value. The long-wavelength 710 nm emission band of Pchl<sub>ide</sub> fulfils the conditions of a vibronic satellite of the intensive main 656-nm emission peak. With regard to the distance of the Q<sub>y</sub> (0,0) and Q<sub>y</sub> (0, 1) bands, the band-widths and the presence of the 728 nm band even after excitation with long-wavelength (677 nm) light, the 710 nm band is the most probable vibrational band for 656 nm Pchl<sub>ide</sub> form. Analyses of low-temperature fluorescence emission and excitation spectra suggest that there are Pchl<sub>ide</sub> spectral forms fluorescing at 666, 680, 690, 698 and 728 nm together with vibrational (0,1) bands positioned at 675, 687, 697 and 710, which then corresponds to the 628, 635, 644 and 656 nm band, respectively (Table 1). Suppression of vibrational Q<sub>y</sub> (0,1) bands of the short-wavelength forms of Pchl<sub>ide</sub> can verify the presence of far-red Q<sub>y</sub>(0,0) components. This could be achieved by using actinic light of wavelengths longer than the absorption peak positions of the short-wavelength. Subtraction of the emission spectrum of heat-denatured leaves excited with the same excitation light enhances the fluorescence emission

Table 1: Spectral parameters of Pchlde bands. Spectral parameters of Pchlde bands obtained after curve resolution of the 77 K excitation spectrum (emission 740 nm) and two types of emission spectra (excitation 440 and 460 nm) of etiolated leaves of wheat. The peak positions can vary slightly when different plant varieties are used for measurements

Parameter	(Spectral values)										
Excitation (Em: 740)	620	628	635	641	649	658	668	677	686	69	
Emission (Ex: 440)	628	635	(642)	644	656	666					
Emission (Ex: 460)				646	656	668	680	690	698	728	

spectrum as the heat-denatured leaves contained no long-wavelength or far-red Pchlde forms. Subtracting the spectrum obtained (Stadnichuk *et al.*, 2005). The peak positions and relative ratio between the fluorescence emission peak heights depend on plant (Fig. 1) species and plant tissues (Fig. 2) (Stadnichuk *et al.*, 2005) as well as external factors such as salt stress (Abdelkader *et al.*, 2007).

The peak position and relative ratio between intensity of the different peaks vary also with the age of tissues. To examine the effect of developmental age on peak positions and intensities the tissues from 1-to 12-day-old dark-grown plants have been used. The 631 nm band is the first Pchlde form, which appeared in dark-grown tissues (Amirjani *et al.*, 2006). In the very young stage the short-wavelength Pchlde forms are dominating with about 80% of the total Pchlde fluorescence for pea and maize. In a later stage of development the Pchlde emission spectra was dominated of the two well-known peaks, 631 and 654-656 nm. The ratio of fluorescence from the short-wavelength Pchlde to that of total Pchlde fluorescence decreased to a minimum of 0.22 in the 12-day-old pea and to a minimum of 0.1 for 6-day-old maize leaves. In older leaves the ratio slightly increased (Fig. 5A, B) (Amirjani *et al.*, 2006).

**Phototransformability of Pchlde forms:** Pchlde forms are divided into flash-photoactive and non-flash-photoactive on the basis of their ability to transform to Chlide. Those Pchlde molecules sitting in the active site of the POR-NADPH complex are flash-photoactive, those which are not or have NADP<sup>+</sup>, are not flash-photoactive (Franck *et al.*, 1999). However, this characterization of the different forms is not crystal clear, as depending on irradiation time, different Pchlde forms can be photo transformed or not. The non-flash-photoactive Pchlde forms have fluorescence excitation peaks at 620-640 nm and emission peaks at 627-646 nm (Stadnichuk *et al.*, 2005). During a prolonged irradiation, short-wavelength Pchlde can also be photo transformed to Chlide, but this process is probably dependent on a conversion of non-flash-photoactive short-wavelength Pchlde into long-wavelength flash-photoactive Pchlde. The ratio of non-flash-photoactive Pchlde forms to the sum of Pchlde forms in the emission fluorescence spectra decreased while the total amount of Pchl(ide) increased during

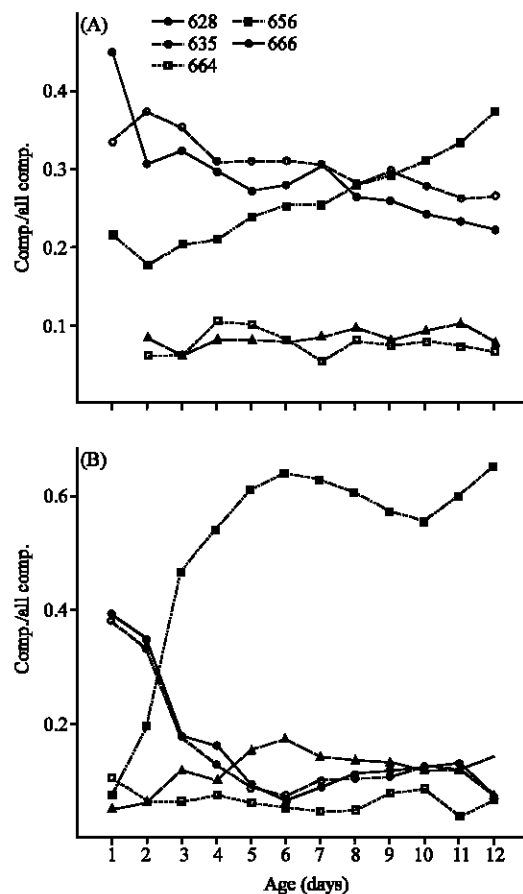


Fig. 5: The ratio of single Gaussian components to the sum of the components during development in darkness. The components were obtained from Gaussian deconvolution of low temperature (77 K) fluorescence emission spectra of dark-grown pea (A) and maize (B) at various ages. The excitation was 440 nm.

development (Fig. 5) (Amirjani *et al.*, 2006). It has been shown that a treatment with ALA results in an increase of the non-flash-photoactive Pchlde forms (Amirjani and Sundqvist 2006). The non-flash-photoactive Pchlde has been considered to represent a pool of free pigment (Kahn *et al.*, 1970) since, its band position is similar to Pchlde in solution. Bystrova reported, however, that a 635-nm Pchlde form is bound to a protein and transmit the excitation energy to the long-wavelength pigment forms



Table 2: Fluorescence lifetime (ns) for pea (wild type), pea (lip1), wheat and maize using excitation light of 440 or 460 nm. For each lifetime corresponding fractions are given in brackets

Em (Ex) (nm)	Species	Species			
		Pea	Lip1	Wheat	Maize
633 (440)	$\tau_1(f_1)$	0.4 (0.18)		0.6 (0.11)	
	$\tau_2(f_2)$	6.7 (0.82)	7.1 (0.99)	6.2 (0.89)	
637 (440)	$\tau_1(f_1)$	0.6 (0.12)			
	$\tau_2(f_2)$	7.0 (0.88)	6.8 (0.99)		
642 (440)	$\tau_1(f_1)$	0.4 (0.12)	7.1 (0.99)		
	$\tau_2(f_2)$	6.2 (0.88)			
642 (460)	$\tau_1(f_1)$		0.6 (0.16)		
	$\tau_2(f_2)$	4.0 (1.00)	7.0 (0.84)		
656 (440)	$\tau_1(f_1)$	0.5 (0.20)	0.7 (0.16)	0.6 (0.09)	0.7 (0.04)
	$\tau_2(f_2)$	6.4 (0.77)	6.5 (0.84)	5.1 (0.91)	6.7 (0.96)
656 (460)	$\tau_1(f_1)$	0.6 (0.11)	0.3 (0.36)	0.8 (0.27)	0.4 (0.28)
	$\tau_2(f_2)$	5.4 (0.89)	7.0 (0.63)	5.6 (0.79)	6.3 (0.72)
670 (440)	$\tau_1(f_1)$	0.3 (0.24)			
	$\tau_2(f_2)$	5.7 (0.67)	5.6 (0.99)		
670 (460)	$\tau_1(f_1)$				
	$\tau_2(f_2)$	2.0 (1.00)	2.7 (0.99)		

(Bystrova *et al.*, 1988). Furthermore, lifetime measurements indicated that the short-wavelength Pchl<sub>a</sub> forms are not free pigments but probably combined with proteins or lipids since their lifetime are shorter than that of Pchl<sub>a</sub> in diethyl ether,  $\tau = 9.3$  ns, or acetone,  $\tau = 8.8$  ns measured at 77 K (Table 2).

The main pigment form of mature etiolated leaves in many plant species is the flash-photoactive Pchl<sub>a</sub> form, depending on the age species, etc as discussed above. This form can be photo transformed by a short (1 m sec) light flash. The main flash-photoactive Pchl<sub>a</sub> form has a long-wavelength fluorescence emission band located around 656 nm, with slight variations depending on species. In some plants, such as pea and its lip1 mutant, this form is, however, a minor form (Stadnichuk *et al.*, 2005; Amirjani and Sundqvist, 2006). The main flash-photoactive Pchl<sub>a</sub> is regarded to be an aggregate of ternary complexes, composed of the photoenzyme, POR, its Pchl<sub>a</sub> substrate and its reduced NADPH cofactor. The photoactive Pchl<sub>a</sub> pool is spectrally and chemically heterogeneous (Stadnichuk *et al.*, 2005). A minor form of photoactive Pchl<sub>a</sub> absorbing around 638 nm and fluorescing at 643-644 nm has been characterized both *in vivo* and *in vitro*, which seems to be the first transformed Pchl<sub>a</sub> form into Chl<sub>a</sub> when etiolated leaves are illuminated with light of low pfd weak irradiation (Böddi *et al.*, 1991). This is interesting as there seems to be a formation of an excited state with fluorescence around 644 nm (Dietzek *et al.*, 2006). Schoefs reported the photoactive Pchl<sub>a</sub> form to be composed of three components fluorescing at 644, 652 and 657 nm (Schoefs *et al.*, 2000).

Photoactive Pchl<sub>a</sub> exists not only in the leaves of etiolated plants but also in greening and green leaves after a period of darkness (Amirjani and Sundqvist, 2004).

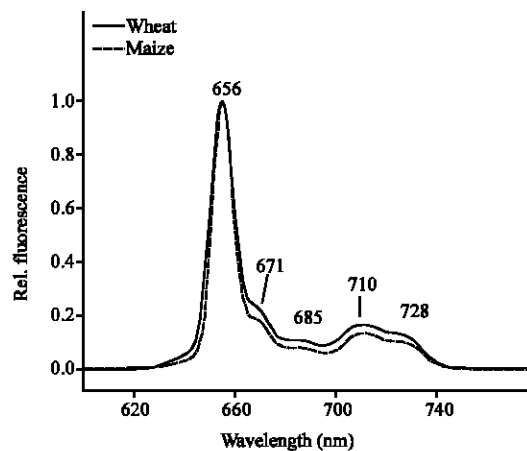


Fig. 6: Low temperature (77 K) fluorescence emission spectra of dark-grown plants. Maize and wheat leaves excited by 460-nm light, respectively. The spectra were normalized to the highest peak

Ignatov demonstrated the occurrence of photoactive Pchl<sub>a</sub> in green leaves under light conditions (Ignatov and Litvin, 2002). We showed that in angiosperms the pathway of Chl biosynthesis in greening and green leaves proceeds through the 655-nm Pchl<sub>a</sub> form, similar to the pathway in dark-grown leaves (Amirjani and Sundqvist, 2004).

**Pchl<sub>a</sub> spectral forms in different plant species:** *In vivo* the Pchl<sub>a</sub> peak positions and relative intensity ratio vary with different plant species. The variation is evident in absorption, fluorescence emission and excitation spectra. The absorption spectrum of maize, barley and wheat has in the red region a peak at 650 and a 636 nm band appears as a shoulder. For pea the 636 nm band is an evident peak and in lip1 the 636-nm absorption band is dominant (Fig. 1) (Amirjani *et al.*, 2006).

In the fluorescence emission spectra the differences between the species are conspicuous. The fluorescence spectra of wheat and maize have a dominant peak located at 656 nm and a small peak at 631 nm. The latter peak is for maize slightly smaller than for wheat (Fig. 6) (Mysliwa-Kurdziel *et al.*, 2003).

As an extreme the emission spectrum of the *aurea* mutant of tomato (*Solanum lycopersicum* L.), which is unable to synthesize the linear tetrapyrrole chromophore of phytochrome and is one of the most exhaustively characterized photomorphogenic mutants (Terry, 1997; Terry and Kendrick, 1999; Terry *et al.*, 2001), does not have a short-wavelength Pchl<sub>a</sub> form in darkness and during an early stage of greening. In contrast in pea and its lip1 mutant the 631 nm emission band is the dominant

band with 440 nm excitation. Pea has a small peak at 654 nm whereas lip1 mutant does not show any evident peak in this region. The epicotyl of dark-grown pea plants also has a dominant short-wavelength Pchlde peak (Fig. 2). The peak is, however, slightly asymmetric around 654 nm indicating the presence of the long-wavelength form. The same is true for maize mesocotyles and coleoptiles of triticale (Savchenko *et al.*, 2005).

Dark-forced wine (*Vitis vinifera*) leaves and stems were dominated by the short-wavelength Pchlde form. Differences are also seen in the wavelength region between 660 and 730 nm. The 666, 680 and 728 nm wavelengths Pchlde forms are more pronounced in wheat and maize (cf. (Fig. 6). Using a statistical method calculating the average of the absolute deviation (AVEDEV) of datapoints in a spectrum from their mean function and plotting the AVEDEV values against the wavelength clearly showed the bands at 626, 636 and 656nm for pea epicotyl (Szenzenstein *et al.*, 2008).

The cause for the species dependent spectral differences can be sought on different levels. As mentioned the fluorescence spectra from epicotyls of pea are quite different from those of the leaf tissues. In maize and wheat the spectra from different part of leaves are different. Therefore the presence of immature palisade parenchyma or spongy parenchyma, or like in wheat and maize, immature mesophyll cells might be important for the spectral forms. The occurrence of different types of PLBs can also be of importance. The correlation between Pchlde<sub>650-656</sub> and the presence of PLBs has been shown earlier (Ryberg and Sundqvist, 1982). Differences in the interaction between Pchlde, POR and the PLB/PT membranes depending on a variation in the lipid composition of the different membranes can also be possible. Finally, when it comes to POR itself there might be differences in the interaction between the Pchlde pigment and the POR protein.

**Reaccumulation of Pchlde during greening:** When illuminated plants return to darkness they re-accumulate photoactive Pchlde. The re-accumulation proceeds in different ways. Some Pchlde, which are bound to an NADP<sup>+</sup> in a ternary complex, can transform into photoactive Pchlde within seconds when the NADP<sup>+</sup> is reduced. The 633 nm fluorescing form can be transformed into photoactive Pchlde within minutes, when newly formed Chlide is removed from the active site of POR and replaced by new Pchlde from the precursor pool. During a prolonged dark-period (h), newly synthesized Pchlde also contribute to the re-accumulation. Duration and intensity of the illumination light determine the ratio between re-accumulated Pchlde fluorescing at 633 and

653 nm. The spectral properties of Pchlde re-accumulated in plants illuminated for a short time (1 h) were specific for the plant species and resembled those of the Pchlde accumulated in the dark-grown plants (Amirjani and Sundqvist, 2004). The re-accumulation of long-wavelength Pchlde probably reflects the re-formation of the PLBs since the plastids still have a capacity to develop etioplast features when the plants are returned to darkness after illumination (Schoefs and Franck, 2008). With prolonged illumination (24 h) the re-accumulation of the long-wavelength Pchlde form apparently decreased. The decrease might be due to a decrease in POR as the PORA gene is known to be down-regulated after irradiation (Apel, 1981). However, the excitation spectra showed that there could be more Pchlde present absorbing around 650 nm than revealed by the fluorescence emission spectra. There seems to be an energy transfer from the long-wavelength Pchlde to the Chl (Brouers and Sironval, 1974) and this energy transfer diminish the contribution of the long-wavelength Pchlde form to the fluorescence emission at 656 nm (Amirjani and Sundqvist, 2004). Energy may transfer from a donor chromophore, initially in its electronic excited state, to an acceptor chromophore (in proximity, typically less than 10 nm) through nonradiative dipole-dipole coupling. This mechanism is termed Förster resonance energy transfer and is named after the German scientist Theodor Förster. When both chromophores are fluorescent, the term fluorescence resonance energy transfer is often used instead, although the energy is not actually transferred by fluorescence (Lakowicz, 1999). Thus judging from both fluorescence emission spectra and fluorescence excitation spectra the spectral forms of the re-accumulated Pchlde also in green leaves seemed to be dependent on plant species and present in proportions similar to those in the corresponding dark-grown material.

**Fluorescence lifetime of Pchlde forms:** The fluorescence lifetime is one of the most important characteristics of a fluorophore. When a fluorophore absorbs the energy of a photon the molecule become excited and can be in the excited state during a time period corresponding to its fluorescence lifetime. The lifetime determines the time available for the excited fluorophore to diffuse and interact with its environment before it returns to its ground state. During the fluorescence lifetime, energy migration can occur.

Earlier lifetime measurements on Pchlde indicate that photoactive and non-photoactive Pchlde have different lifetimes. Two lifetime components were found in time-resolved fluorescence measurements on etioplast membranes. The low temperature *in vivo* fluorescence

lifetime was found to be shorter (5.1-7.1 ns) than the fluorescence lifetime of Pchl<sub>ide</sub> in organic solvents measured under the same experimental conditions (around 9 ns) (Mysliwa-Kurdziel *et al.*, 2003). However, similar to the absorption, excitation and emission fluorescence spectra, the lifetimes of the Pchl<sub>ide</sub> forms differed between the plants (Table 2).

The lip1 mutant excited at 440 nm, has one Pchl<sub>ide</sub> fluorescence lifetime of approximately 7 ns with emission set at 633, 637 or 642 nm. This fluorescence lifetime can be related to both the 635 and 642 nm forms of Pchl<sub>ide</sub> since both forms are present, although in different proportions, in the steady-state fluorescence spectra. When the emission wavelength was set at 670 nm, only one component, now with a lifetime of 5.6 ns can be found. For the 656 nm emission two components of the fluorescence lifetime were observed. When the sample was excited at 460 nm two lifetime components are observed for the emission at 642 and 656 nm. For this excitation wavelength, the fast component has a lifetime value about two times longer (0.6 ns) for the emission at 642 nm than at 656 nm. However, its fractional intensity was lower (Table 2). The fast component decreases concomitant with the increase of the fractional intensity, when comparing the results obtained for emission at 656 nm with the two excitations of 440 and 460 nm. Wild-type pea leaves excited at 440 nm have two fluorescence lifetime components for different emission wavelengths. For the excitation at 440 nm and the emission at 633 nm, the main contribution to total fluorescence is from the Pchl<sub>ide</sub><sub>628-635</sub> form (0.78) and the rest (0.22) from Pchl<sub>ide</sub><sub>635-642</sub>. Thus, in wild-type pea it seems the fast component came from the Pchl<sub>ide</sub><sub>635-642</sub> form. For the emission wavelength set at 637 and 642 nm, the contribution of Pchl<sub>ide</sub><sub>628-633</sub> in the steady-state spectra decrease and that of Pchl<sub>ide</sub><sub>635-642</sub> increases and finally reach 0.18 and 0.77, respectively. However, the fractions of the fluorescence lifetimes stay more or less at the same levels, which do not reflect the changes in the steady-state fluorescence of the different Pchl<sub>ide</sub> forms. The slow component has a lifetime between 5.7 and 7.0 ns. For the emission wavelength of 642 nm, the slow component for wild-type pea has a lower value (6.2 ns) than the slow component for lip1 (7.1 ns). For the excitation at 460 nm two lifetime components are found for the emission at 656 nm, whereas only one component was determined for the 642 and 670 nm emission (Table 2). Decay Association Spectra (DAS) resembled the fluorescence spectrum of wild-type pea leaves for both the slow and fast lifetime components (Mysliwa-Kurdziel *et al.*, 2003). Thus, in the case of wild-type pea, not only 654-nm Pchl<sub>ide</sub> but also 635 and 642 nm Pchl<sub>ide</sub> forms show a complex character of the

fluorescence decay that can be described only using two fluorescence lifetime components.

**Organization of pigment-protein complexes:** The Pchl<sub>ide</sub> molecule suits a pocket on the POR protein with the active site at the bottom (Townley *et al.*, 2001). During phototransformation the hydride transfer occurs from NADPH, which is located at the bottom of the substrate-binding pocket. Conserved Tyr and Lys residues in POR supply the protons for the reaction (Wilks and Timko, 1995). Three sites on the Pchl<sub>ide</sub> molecule are of special importance for POR activity. The central metal atom should preferably be a magnesium atom. The propionic side chain on the porphyrin ring D should be whole and the complete structure of ring E should be present. The penta-coordination of the magnesium has contributions from the four tetrapyrrole nitrogens and one ligand from the POR. The POR enzyme allows a certain degree of modification of the side chains on ring A and B even if these modifications give spectral shifts. Thus Pchl<sub>ide</sub> a and b with different electronic spectra have almost identical reaction rates *in vitro*.

The short-wavelength Pchl<sub>ide</sub> forms are monomeric and probably bound to a protein. It might be that the monomeric Pchl<sub>ide</sub> forms are not phototransformable but a dimeric form is necessary. Then the 635 nm Pchl<sub>ide</sub> form may be bound to the POR and even in the active site. The upper part of the Pchl<sub>ide</sub> molecule containing ring A and ring B protrudes from the POR protein (Townley *et al.*, 2001). This creates a possibility for an interaction between different Pchl<sub>ide</sub> molecules. A formation of dimers can be caused by an interaction between two POR-bound Pchl<sub>ide</sub> molecules with overlapping A and B rings (Fig. 7). The 642-644 nm fluorescing Pchl<sub>ide</sub> form can be regarded as a dimer of the ternary complex of NADPH, POR and Pchl<sub>ide</sub>. This form is flash-photoactive and the phototransformation is faster than for the main photoactive Pchl<sub>ide</sub> form. This form is mostly present in small amounts suggesting that it easily form larger aggregates when it becomes more abundant.

If the interaction between the Pchl<sub>ide</sub> molecules is limited to only a part of the free end of the Pchl<sub>ide</sub> molecule there can be room for an interaction also for a third POR-bound Pchl<sub>ide</sub> molecule. In this way large aggregates with interacting Pchl<sub>ide</sub> molecules can be formed. The aggregates are then associated with the PLB membrane and this interaction also seems to influence the spectral properties of the Pchl<sub>ide</sub>. Far-red forms of Pchl<sub>ide</sub> could be regarded as representing a higher degree of aggregation (Fig. 7). Another possibility is that the dimers formed via the pigments interact through the POR molecules and creates the larger aggregates. The Pchl<sub>ide</sub>

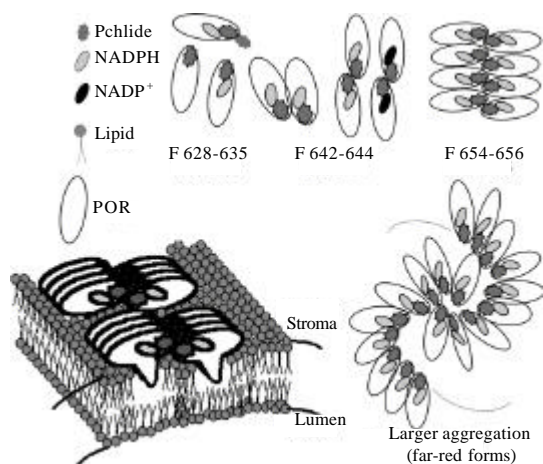


Fig. 7: Schematic diagram illustrating the aggregation state of ternary complexes of Pchl a, POR and NADPH

molecules are in good contact for the interaction of the  $\pi$  electron system. A size of the aggregates of 8 interacting POR molecules (Fig. 7) has been suggested as energy transfer units. A possibility is also that a free (not POR-bound) Pchl a molecule can be attached to a POR-bound Pchl a. At a flash irradiation such a Pchl a molecule would not be phototransformed as it is not at the active site. After phototransformation it can rapidly replace the newly formed Chl a and contribute to a rapid regeneration of phototransformable Pchl a. If such non-POR bound Pchl a molecules could substitute for a POR-bound Pchl a during formation of the large aggregates a considerable variation in the spectral properties and the degree of phototransformability could be expected.

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I would like to dedicate this paper to Professor Christer Sundqvist.

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