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Chromium (VI) Reduction by *Streptococcus* Species Isolated from the Industrial Area of Abeokuta, Ogun State, Nigeria

¹Parvaze Ahmad Wani, ¹Ibrahim Odunola Zainab, ²Idris Adegbite Wasiu and ¹Kuranga Oluropo Jamiu

¹Department of Biological Sciences,

²Department of Chemical Sciences, College of Natural and Applied Sciences, Crescent University, Abeokuta, Ogun State, Nigeria

Corresponding Author: Parvaze Ahmad Wani, Department of Biological Sciences, College of Natural and Applied Sciences, Crescent University, Abeokuta, Ogun State, Nigeria

ABSTRACT

Chromium (VI) contamination has accelerated due to rapid industrialization worldwide. Aim of this study is to check the bacterial species for their tolerance towards multiple metals, antibiotics and plant growth promoting activity and further check whether these bacteria are reducing Cr (VI). Bacterial strains were isolated from metal contaminated soils of Abeokuta. All of the isolates showed tolerance to lead, zinc and chromium (VI). Bacterial specie also showed tolerance towards antibiotics, 100% of the isolates were tolerant to Septrin, Chloramphenicol, Sparfloxacin, Amoxicillin, Augmentin, Tarivid and Streptomycin, whereas 83.33% were tolerant to Gentamycin and Pefloxacin and 33.33% were resistant to Ciprofloxacin. All bacterial species were positive to ammonia, whereas strain PZ3 and PZ4 were found to be positive to HCN. Among all the strains, only *Streptococcus* spp. PZ4 showed reduction of Chromium (VI). Maximum reduction (85%) of chromium (VI) was observed at pH 7 by *Streptococcus* spp. PZ4. Similarly, *Streptococcus* spp. PZ4 reduced the chromium considerably at pH 5 (51.25%), pH 6 (72.5%), pH 8 (67.5%) and at pH 9 (45%), at a concentration of 100 µg Cr mL⁻¹ after 120 h of incubation. *Streptococcus* spp. PZ4 reduced chromium (VI) at a concentration of 50 µg Cr mL⁻¹ (47.5%), 100 µg Cr mL⁻¹ (91.25%) and 150 µg Cr mL⁻¹ (134.17%). Due to above properties strains could therefore be used as bioremediators of metals in soils contaminated with heavy metals and can also increase the yield of various crops under heavy metal contamination.

Key words: *Streptococcus* species, metal tolerance, chromium (VI) reduction, plant growth promoting activities

INTRODUCTION

The contamination of chromium (VI) mainly due to the use of Cr (VI) in leather, tanning, metallurgy, electroplating, textile and pigment manufacturing industries (Wang and Xiao, 1995; Pattanapitpaisal *et al.*, 2001; Sultan and Hasnain, 2007). Chromium occurs either in trivalent or hexavalent which affect growth of microorganisms present in the environment (Ortegel *et al.*, 2002). Hexavalent chromium is highly soluble in water, permeable through biological membranes and it interacts with proteins and nucleic acids which makes it more toxic and carcinogenic than trivalent (Kamaludeen *et al.*, 2003; Ackerley *et al.*, 2006). Reduction of Cr (VI) leads to the formation of stables, less soluble and less toxic Cr (III). Reduction of toxic Cr (VI) to Cr (III) is thus a useful process for remediation of Cr (VI) affected environments (Jeyasingh and Phillip, 2005).

The reduction/detoxification of Cr (VI) by microbes is, however, inexpensive and environmentally safe approach and provides a viable option to protect the environment from chromium toxicity. The reduction of Cr (VI) has been reported in *Bacillus* sp. (Elangovan *et al.*, 2006; Chaturvedi, 2011), *Pseudomonas* sp. (Rahman *et al.*, 2007), *Escherichia coli* (Bae *et al.*, 2005), *Microbacterium* (Pattanapitpaisal *et al.*, 2001), *Ochrobactrum intermedium* (Faisal and Hasnain, 2005) and *Micrococcus* (Sultan and Hasnain, 2005).

Detoxification of chromium by microbes may occur directly or indirectly and is affected by pH, chromate concentration, incubation periods and the types of microbes involved. In the direct mechanism microbes enzymatically (chromium reductases) reduce chromium (Losi *et al.*, 1994) while in the indirect method, reductants or oxidant, such as H₂S, reduce chromium (DeFilippi and Lupton, 1992). Furthermore, in growing cultures with added carbon sources as electron donors and in cell suspensions, Cr (VI) reduction can be predominantly aerobic or anaerobic but generally not both. Interestingly, chromium reductases can catalyse reduction of Cr (VI) to Cr (III) anaerobically (Lovley and Coates, 1997), aerobically (Cervantes *et al.*, 2001) and also both anaerobically and aerobically (Marsh and McInerney, 2001). The Cr (VI) reductase may be present in the membrane fraction of the cells of PGPR, as found in *Pseudomonas fluorescens* and *Enterobacter cloacae* (Wang *et al.*, 1990). Further evidence suggested that cytochrome c548 was involved in the reduction of Cr (VI) by membrane vesicles. In the presence of H₂ and excess of hydrogenase, cytochrome C3, a periplasmic protein, in the soluble cell free fraction of *D. vulgaris* (Lovley and Coates, 1997), reduced Cr (VI), 50 times faster than did the Cr (VI) reductase of *P. ambigua* with NADH and NADPH, as electron donor (Horitsu *et al.*, 1987). Under anaerobic condition Cr (VI) reduction is due to the action of enzymes associated with membranes of the electron transfer system (Cervantes and Campos, 2007). The reduction to Cr (III) results in the formation of insoluble precipitate [Cr(OH)₃] which is easily removed from wastewater (Jeyasingh and Phillip, 2005). The enzyme chromium reductase found in *P. ambigua* (Campos-Garcia *et al.*, 1997) and *Bacillus* sp. (Wang *et al.*, 1991) were purified and characterized. More recently, to clone a chromate reductase gene, novel soluble chromate reductase of *P. putida* was purified to homogeneity and characterized (Puzon *et al.*, 2002). The reductase activity was NADH- or NADPH-dependent. Reduction of Cr (VI) by H₂S produced by the bacterial cells is found in soil environments which are rich in sulfate under anaerobic conditions (Losi *et al.*, 1994). Hydrogen sulfide, produced in acid sulfate soil under reducing conditions, is easily precipitated as FeS in reduced soils (Eary and Rai, 1991) and sediments. Fe (II) and H₂S, both microbially produced, are effective reductants of Cr (VI) under reduced conditions as is the FeS (Karnachuk, 1995).

In addition, chromium reducing bacteria also synthesize plant growth promoting substances (Wasi *et al.*, 2008). Therefore, the use of bacteria for reduction/detoxification of chromium is one of the preferred choices and is considered as cost effective approach in bioremediation technologies. The present study was therefore under taken (1) To determine the resistance pattern of soil bacteria to heavy metals and antibiotics and (2) To check chromium reduction under varying pH and chromium concentration.

MATERIALS AND METHODS

Collection of soil sample: The soil samples for the isolation of heavy metal resistance microorganisms were collected from the contaminated soils of Abeokuta, Ogun state, Nigeria.

Isolation of bacteria: Bacteria were isolated from the contaminated soils of Abeokuta on nutrient agar medium by spread plate technique. One gram of soil sample was added to a flask containing 100 mL of normal saline solution and was serially diluted. A 10 µL of each suspension was spread

plated on solid nutrient agar. Plates were incubated at $28\pm 2^\circ\text{C}$ for 24 h and the bacterial colonies were then purified and preserved on nutrient agar slants for further experiments.

Evaluation of bacterial strains for metal tolerance: The isolated bacterial strains from the contaminated soil were tested for their sensitivity/resistance to three heavy metals viz., chromium, zinc and copper by agar plate dilution method (Holt *et al.*, 1994) using nutrient agar. The freshly prepared agar plates amended with increasing concentration of chromium ($0-700\ \mu\text{g mL}^{-1}$), zinc ($0-700\ \mu\text{g mL}^{-1}$) and copper ($0-700\ \mu\text{g mL}^{-1}$) were spot inoculated ($10\ \mu\text{L}$) with 10^8 cells mL^{-1} . Plates were incubated at $28\pm 2^\circ\text{C}$ for 72 h and the highest concentration of heavy metals supporting growth was defined as the Maximum Resistance Level (MRL). Each experiment was replicated three times.

Determination of antibiotic sensitivity: To determine susceptibility to antibiotics, the bacterial strains were tested for their sensitivity to ten antibiotics. The reactions to antibiotics were determined by the disc diffusion method (Bauer *et al.*, 1966). The bacterial strains were grown in nutrient broth at $28\pm 2^\circ\text{C}$ for 24 h. A 0.1 mL of the over-night grown culture was spread on the surface of nutrient agar. The antibiotic discs of known potency were then placed on the agar surface and the plates were incubated at $28\pm 2^\circ\text{C}$ for 24 h. The zones of inhibition around the antibiotic discs were measured and the strains were classified as Resistant (R), Intermediate (I) and Susceptible (S), following the standard antibiotic disc sensitivity testing method (Difco Laboratories, 1984) to the following antibiotics: Septrin ($30\ \mu\text{g}$), Chloramphenicol ($30\ \mu\text{g}$), Ciprofloxacin ($10\ \mu\text{g}$), Sparfloxacin ($10\ \mu\text{g}$), Amoxicillin ($30\ \mu\text{g}$), Augmentin ($30\ \mu\text{g}$), Gentamycin ($10\ \mu\text{g}$), Pefloxacin ($30\ \mu\text{g}$), Streptomycin ($30\ \mu\text{g}$) and Tarivid ($10\ \mu\text{g}$).

In vitro assay of hydrogen cyanide and ammonia: Hydrogen cyanide production by bacterial isolates was detected by the method of Bakker and Schippers (1987). For HCN production, the bacterial strains were grown on an HCN induction medium (30 g tryptic soy broth, 4.4 g glycine and 15 g agar L^{-1}) at $28\pm 2^\circ\text{C}$ for four days. For each bacterial isolate, $100\ \mu\text{L}$ of 10^8 cells mL^{-1} was placed in the centre of the petri plates. A disk of Whatman filter paper No. 1 dipped in 0.5% picric acid and 2% Na_2CO_3 was placed at the lid of the petri plates. Plates were sealed with parafilm. After four days incubation at $28\pm 2^\circ\text{C}$, an orange brown colour of the paper indicating HCN production was observed.

For ammonia production, the bacterial strains were grown in peptone water (g L^{-1} : peptone 10 g, NaCl 5 g, pH 7) and incubated at $30\pm 2^\circ\text{C}$ for four days. One milliliter of Nessler reagent was added to each tube and the development of yellow color indicating ammonia production was recorded (Dye, 1962).

Chromium (VI) reduction: To assess the effect of pH on hexavalent chromium [Cr (VI)] reduction *in vitro*, the Nutrient Broth (NB) was amended with $100\ \mu\text{g mL}^{-1}$ of Cr (VI) and the autoclaved medium was adjusted to pH 5, 6, 7, 8 and 9 with 1M HCl or 1 M NaOH and incubated at $28\pm 2^\circ\text{C}$ for 120 h. Further, to assess the effect of different concentrations ($0, 50, 100$ and $150\ \mu\text{g mL}^{-1}$) of Cr (VI), the $\text{K}_2\text{Cr}_2\text{O}_7$ were amended in Nutrient broth and incubated at $28\pm 2^\circ\text{C}$ for 120 h. For Cr (VI) reduction, 1 mL culture from each flask was centrifuged (6000 rpm) for 10 min at 10°C and Cr (VI) in the supernatant was determined by 1, 5-diphenyl carbazide method (Eaton *et al.*, 1992) upto 120 h. Briefly, the test samples were acidified (pH 1-2) and 1, 5 diphenyl carbazide ($50\ \mu\text{g mL}^{-1}$) was added and Cr (VI) concentration was detected by spectrophotometer (spectronic 20D) at 540 nm.

Data of three replicates were subjected to statistical analysis using pair samples T test with significant level of $p < 0.05$. The values indicate the Mean \pm SD of three replicates.

RESULTS AND DISCUSSION

Evaluation of bacterial for metal tolerance: The bacterial strains were evaluated for their tolerance to various concentrations of chromium (VI) and other metals like zinc and copper using agar plate dilution method (Fig. 1). Generally, bacterial strains showed a varied level of tolerance to heavy metals. Among the bacterial strains, *Bacillus* spp. PZ3 and *Streptococcus* spp. PZ4 showed highest tolerance to Chromium (vi) at concentration 700 $\mu\text{g mL}^{-1}$, *Pseudomonas* spp. PZ1 and *Streptococcus* spp. PZ2 showed highest tolerance to Copper and Zinc at concentration 700 and 700 $\mu\text{g mL}^{-1}$, respectively. There are reports which have shown the resistance of bacteria to heavy metals (Wani and Khan, 2013). There are many reports which have shown varied level of tolerance by bacteria. This varied level of resistance could be due to the variation in growth conditions employed (Rajkumar *et al.*, 2005). For example, *Rhizobium leguminosarum* have shown a tolerance level of 92.9 μM to zinc (Delorme *et al.*, 2003) while *Rhizobium* species isolated from nodules of *Trifolium repense* tolerated 300 mg kg^{-1} nickel and when grown in nickel amended soil could nodulate the legume crop effectively (Smith and Giller, 1992). In the present study, *Bacillus* spp. PZ3 and *Streptococcus* spp. PZ4 showed highest tolerance to Chromium (vi) at a concentration of 700 $\mu\text{g mL}^{-1}$ to each strain, *Pseudomonas* spp. PZ1 and *Streptococcus* spp. PZ2 showed highest tolerance to Copper and Zinc at concentration 700 and 700 $\mu\text{g mL}^{-1}$, respectively. Bacterial strains showed a high tolerance to chromium which was followed by copper and then zinc. The metal tolerant strains were characterized by physiological, morphological and biochemical characteristics. The strain PZ1 and PZ6 was characterized as *Pseudomonas* spp. PZ2 and PZ4 was characterized as *Streptococcus* spp. PZ3 was characterized as *Bacillus* spp. while PZ5 was characterized as *Micrococcus* spp.

Rhizobium leguminosarum biovar *trifolii* isolated from sewage sludge treated soil showed a high tolerance level of 0.24-0.26 mM to Ni^{2+} and 6.0-8.0 mM to Zn^{2+} (Purchase and Miles, 2001). Similarly, metal tolerance by *Rhizobium*, *Bradyrhizobium* and *Azotobacter* (Pajuelo *et al.*, 2008) and varying level of resistance among other PGPR (*Bacillus* and *Pseudomonas*) have also been reported (Yilmaz, 2003; Thacker *et al.*, 2007; Wani *et al.*, 2008).

Antibiotic resistance of bacterial strains: Resistance to antibiotics varied from one bacterial strain to another strain (Table 1). Among bacterial species, 100% of strains were resistant to

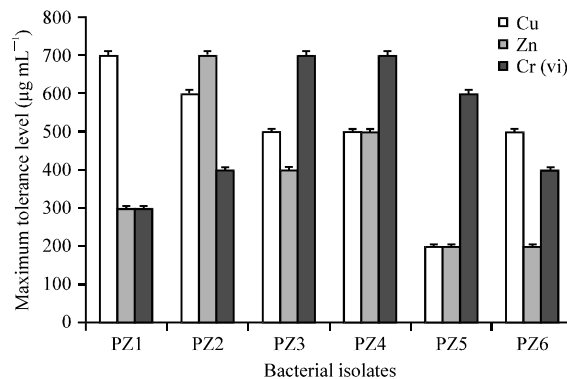


Fig. 1: Maximum tolerance level shown by different bacterial isolates

Table 1: Resistance pattern of bacterial species to various antibiotics

Antibiotics	Concentrations ($\mu\text{g mL}^{-1}$)	No. of resistant isolates	
		No.	%
Septtrin	30	6	100.00
Chloramphenicol	30	6	100.00
Sparfloxacin	10	6	100.00
Ciprofloxacin	10	2	33.33
Amoxicillin	30	6	100.00
Augmentin	30	6	100.00
Gentamycin	10	5	83.33
Pefloxacin	30	5	83.33
Tarivid	10	6	100.00
Streptomycin	30	6	100.00

Table 2: Plant growth promoting activities of metal resistant bacterial isolates

Bacterial isolate No.	Plant growth promoting activity	
	Ammonia	HCN
<i>Pseudomonas</i> spp. (PZ1)	+	-
<i>Streptococcus</i> spp. (PZ2)	+	-
<i>Bacillus</i> spp. (PZ3)	+	+
<i>Streptococcus</i> spp. (PZ4)	+	+
<i>Micrococcus</i> spp. (PZ5)	+	-
<i>Pseudomonas</i> spp. (PZ6)	+	-

Septtrin, Chloramphenicol, Sparfloxacin, Amoxicillin, Augmentin, Tarivid and Streptomycin, 83.33% were resistant to Gentamycin and Pefloxacin whereas 33.33% were resistant to Ciprofloxacin. Bacterial resistance to antibiotics is an emerging problem these days. Resistance to antibiotics by microbes could be due to change in the genetic makeup, can be due to a genetic mutation or by transfer of antibiotic resistant genes between organisms in the environment (Spain and Alm, 2003). With these considerations, the antibiotic resistance among PGPR was studied which differed from antibiotic to antibiotic for all the PGPR strains. Multi antibiotic resistances shown by the bacterial strains (e.g., *Bacillus* spp. PZ3 and *Pseudomonas* spp. PZ6) may be because of their tolerance to metals which have been reported in many studies (Yilmaz, 2003; Verma *et al.*, 2001). It has been observed that metal and antibiotic resistant organisms can adapt faster to metal stress in the environment due to R-factors than by mutation and natural selection (Silver and Misra, 1988). Similar observations on antibiotics resistance by PGPR strains have been reported (Thacker *et al.*, 2007). The variation in the resistance to many tested antibacterial drugs (antibiotics) may possibly be due to the differences in growth conditions and exposure of PGPR to stress conditions or toxic substance as well as presence or absence of resistance mechanisms that could be encoded either by chromosome and/or R-plasmid (Spain and Alm, 2003).

In vitro assay of HCN and ammonia: The plant growth promoting rhizobacterial strains were tested further for the synthesis of ammonia and hydrogen cyanide using peptone water and HCN induction medium, respectively (Table 2). Generally, all PGPR strains were found positive for ammonia while PZ3, PZ4 were found to be positive for HCN. In the present study bacterial strains were positive to plant growth promoting activities and produced substantial amount of HCN and Ammonia. The ammonia released by these bacterial strains plays a signaling role in the interaction

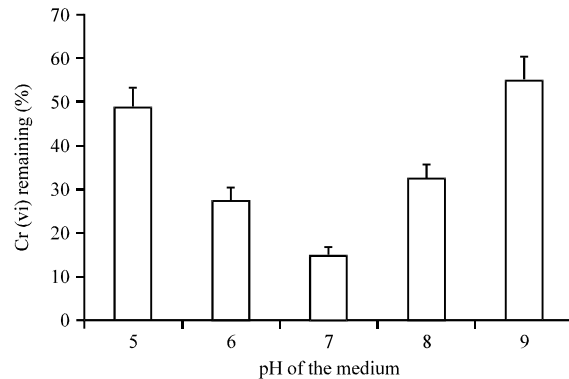


Fig. 2: Effect of pH on Cr (VI) reduction ability of *Streptococcus* spp. PZ4 after 120 h of growth in nutrient broth

between plant growth promoting bacteria and plants (Becker *et al.*, 2002). Moreover, the ammonia released by the bacterial strain is known to increase the glutamine synthetase activity (Sood *et al.*, 2002). In addition, ammonia transporters found in several PGPR are thought to be involved in the reabsorption of NH_4^+ released as a consequence of NH_3 diffusion through the bacterial membrane (Van Dommelen *et al.*, 1997).

Chromium (VI) reduction: Chromium is an environmental pollutant released from various industries including tanneries, metal cleaning and processing, chromium plating, wood processing and alloy formation. Chromium [Cr (VI)] is the most toxic and carcinogenic (Kamaludeen *et al.*, 2003) due to its high solubility, rapid permeability and their intracellular proteins and nucleic acids (Reeves *et al.*, 1983). The reduction of Cr (VI) leads to the formation of stable, less soluble and less toxic Cr (III) and is thus, a useful process for remediation of Cr (VI) affected environments (Thacker *et al.*, 2007). Thus detoxifications of chromium by bacterial strains is thus a good technique to clean the environment from chromium. Therefore, the present study was designed to determine the Cr (VI) reducing ability of the metal tolerant strains.

Among all the strains only one strain *Streptococcus* species PZ4 showed chromium reducing ability under *in vitro* conditions as this strain was highly resistant to chromium (VI). This study was carried out to access the (1) Effect of different pH values on the reduction of Cr (VI) and (2) The effect of chromate concentration on chromium (VI) reduction.

The effect of different pH values on the reduction of chromium (VI) is shown in Fig. 2. Maximum reduction (85%) of chromium (VI) was observed at pH 7 by *Streptococcus* spp. PZ4. Similarly, PGPR isolates *Streptococcus* spp. PZ4 reduced the chromium considerably at pH 5 (51.25), pH 6 (72.5%), pH 8 (67.5%) and at pH 9 (45%), respectively, at a concentration of $100 \mu\text{g Cr mL}^{-1}$ after 120 h of incubation.

In this study, the chromium reducing ability of PGPR strain was assessed using nutrient broth supplemented with 50, 100 and $150 \mu\text{g mL}^{-1}$ of $\text{K}_2\text{Cr}_2\text{O}_7$ in order to determine the effect of chromium (VI) reducing ability of the selected culture under *in vitro* conditions (Fig. 3). The time for total reduction of chromium (VI) increased with increase in the concentration of chromium (VI). During this study, the complete reduction of chromium (VI) occurred after 120 h by *Streptococcus* spp. PZ4 (Fig. 2) at $50 \mu\text{g mL}^{-1}$ of chromium. The PGPR strain *Streptococcus* spp. PZ4

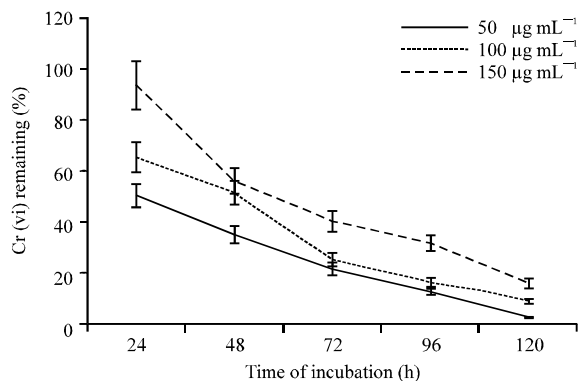


Fig. 3: Effect of Cr (VI) on Cr (VI) reduction ability of *Streptococcus* spp. PZ4 in nutrient broth after 120 h of incubation

reduced chromium (VI) at concentration of 50 µg Cr mL⁻¹ (47.5%), 100 µg Cr mL⁻¹ (91.25%) and 150 µg Cr mL⁻¹ (134.1%), respectively. Our study is in correlation with the study of Yang *et al.* (2009) who also observed considerable reduction of chromium.

CONCLUSION

This study concludes that the bacterial strains not only tolerated heavy metals, antibiotics, produced plant growth promoting substances but also reduced Chromium (VI) under different pH and chromium concentration. Due to multifarious properties expressed by the bacterial strains, these strains could therefore be used as bioremediators of metals in soils contaminated with heavy metals and can also increase the yield of various crops under heavy metal contamination.

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