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Research Article

Contamination of Water Tank of Schools: A Public Health Emergency

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Abstract

Background and Objective: Daily consumption, hygiene and food preparation needs are around 190 liters of water for each individual. Allied to the lack of water is its poor distribution and contamination of the resource. The water tank is a tank designed to store water for human consumption. Maintenance-free water tanks are of great importance for the spread of disease. The study aimed to evaluate the influence of reservoirs on the microbiological quality of drinking water in schools. **Materials and Methods:** In this study investigated water supply quality problems of schools due to their storage through physical, chemical and bacteriological analysis. The 39 samples of water used for human consumption collected in public and private elementary and elementary schools. The pH, chlorine and fluorine presented results that were out of the standard established by the law in force. **Results:** Some water samples analyzed showed different results before and after the reservoir, noting a sensitive interference of water quality by the reservoir. Although the average sample of chlorine results after the reservoir was lower than the minimum allowed by legislation, there was little presence of total coliforms. **Conclusion:** There was a significant influence of reservoirs on the microbiological quality of drinking water in schools.

Key words: Quality, water, fecal coliform, total chloroform, pH, chlorine, fluorine

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Data Availability: All relevant data are within the paper and its supporting information files.

INTRODUCTION

Of the many uses that water can have, some are closely related to human health, mainly when used as a drink, preparation and food intake¹. The quality intended for human consumption must be adequate to maintain health. It is defined by potability standards that describe the permitted or tolerated quantities for various elements that may be present in public water supply². All countries have their water supply legislation and must comply with approximately 50 parameters, divided into four groups: microbiological, toxicological agents, organoleptic and operational³.

Surveillance of water quality for human consumption is a set of actions taken by Public Health agencies to assess the risks of water supply systems and alternative solutions for human health². Infectious microbial agents can cause water-related diseases. The bacterial indicators in water measured are total coliforms (TC) and fecal coliforms (FC). The FC belong to the coliform group and have as habitat the intestinal tract of mammals. *Escherichia coli* are FCs that are preferential indicators of contamination. The permitted TC and HR indices in drinking water are based on epidemiological studies. The FC are related to hygienic conditions and FC to fecal contamination, which indicate the proportion of *E. coli*.

The potability of water guaranteed the treatment process carried out before distribution to the population, which depends on the characteristics of the source that supplies it, as conventional treatments hardly remove some toxic chemical compounds³. Large amounts of chlorine are used for use as household disinfectants in drinking water to control bacteria and odors. Its use has disadvantages such as cancer risk for people who consume chlorine water more than 90% more likely. However, without disinfecting drinking water can be caused infectious diseases. Contaminants may be in public water supply systems due to certain faulty hydraulic works or improper treatment practices⁴. Microorganisms and toxic substances may be present in the pipes and may contaminate the water supplied to the population⁴.

For a treatment, distribution and storage program to successfully perform its functions, the home storage system must also be efficient. In order to prioritize the health problems of school students, the aim of this study was to investigate the water supply problems of schools in the state of SP-Brazil, of deep and superficial origin through physical, chemical and bacteriological analysis.

MATERIAL AND METHODS

Study area: The samples of drinking were collected from August to October, 2013.

Samples: The 39 samples of drinking water from 20 schools selected from municipal, state and private schools in the city of São Carlos-SP (Brazil). To choose the schools to be sampled, the city map was overlaid, relating the location of the schools with the location of the reservoirs and wells used to supply water to the population. The school's choice depended on location about the source of water from the surface or underground wells used for supply. At each school water samples were collected at two points, one before the reservoir and one after the reservoir.

Samples collected in approximately 500 mL plastic vials with a leakproof protective cap. The vials previously cleaned, containing no substances that could alter any result.

Physicochemical analysis: For this pH analysis, the portable pH meter Cristol Microph 2001 was used.

Turbidity values were determined using the Del Lab-DLM-2000B microprocessor turbidity meter. The water sample was placed in the appropriate glass cuvette and inserted and positioned in the apparatus according to the existing brand. Direct reading gives results in the Nephelometric Turbidity Unit (NTU).

Chlorine values were determined using the Del Lab colorimeter. The water sample was placed to well mark No 01. The powder reagent was added with a small clean paddle and dried. Bowl # 01 introduced into the right hole of the comparator and Bowl # 02 with distilled water in the left hole, rotating the colorimetric disc compared to the color of the reacted sample with the contained disc colors. The reading expressed in milligrams per liter of chlorine.

The method used for fluorine analysis was performed by the reaction of spends, based on the combination of the fluoride ion and an intense red zirconium pigment complexed by the HACH DR 2500 spectrophotometer. The results expressed in ppm (mg L^{-1}).

Bacteriological analysis: The search for fecal and TC performed using the Colilert technique⁵.

Statistical analysis: The results were analyzed statistically by analysis of variance (ANOVA) and the Tukey test was applied at a value $p < 0.05$.

RESULTS

Among the samples taken from the school easel before the reservoir, 13 (65%) are non-standard pH (6.5-9.5) established by current legislation, in only 12 (60%) samples taken after the reservoir. All samples with pH below 6.0 were non-standard. Since the rest of the samples are within the established value (Fig. 1a).

It was found 2 (10%) samples below the established value for chlorine by the legislation in force among those collected before the reservoir. However, for samples taken after the reservoir 100% of cases were within the recommended by legislation. The mean value found for samples taken before the reservoir was 0.65 mg L⁻¹, within the value established by the Brazilian Ministry of Health (BMH) and the mean value for samples collected after the reservoir was 0.16 mg L⁻¹, below the recommended by the BMH (Fig. 1b). The average turbidity

value found for samples taken before the reservoir was 0.3 NTU, the same as for samples taken after the reservoir (Fig. 1c).

Samples 14 (70%) obtained from the non-standard fluorine total for samples taken before the reservoir, while 18 (95%) samples were non-standard samples for those collected after the reservoir (Fig. 1d). For the determination of the fluorine value, an average of 0.46 mg L⁻¹ before the reservoir and 0.3 after the reservoir obtained, all outside the standard established by the legislation, minimum of 0.6 mg L⁻¹ and maximum of 0.8 mg L⁻¹ (Fig. 1d).

From samples taken after reservoir 2 (10%) were positive for TC. The same occurred with 6 (32%) samples among those collected after the reservoir. All samples were negative for FC (*Escherichia coli*) (Fig. 2). Current legislation generally dictates the absence of TC and FC as the standard for microbiological quality for drinking water.

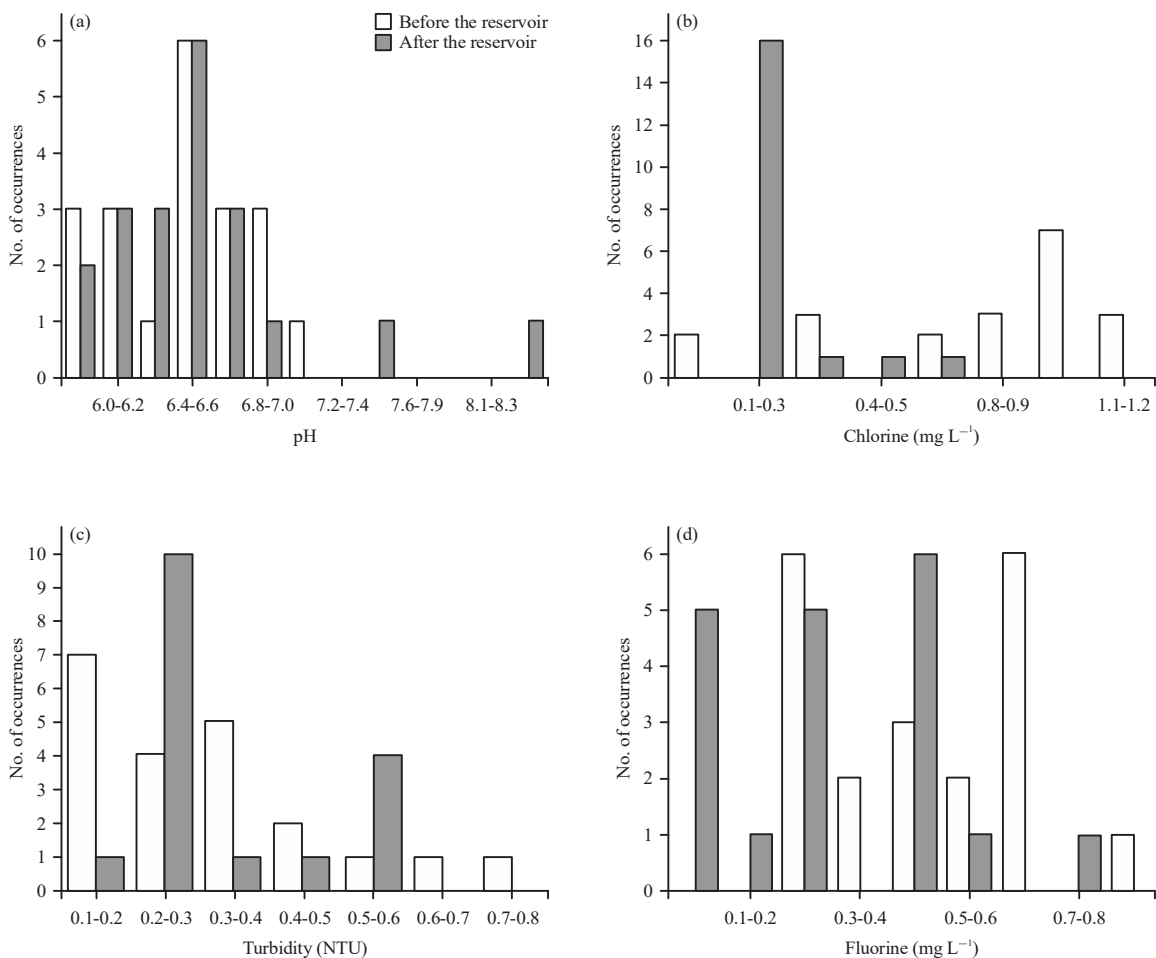


Fig. 1(a-d): Physicochemical analysis of water, (a) pH, (b) Chlorine, (c) Turbidity and (d) Fluorine

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